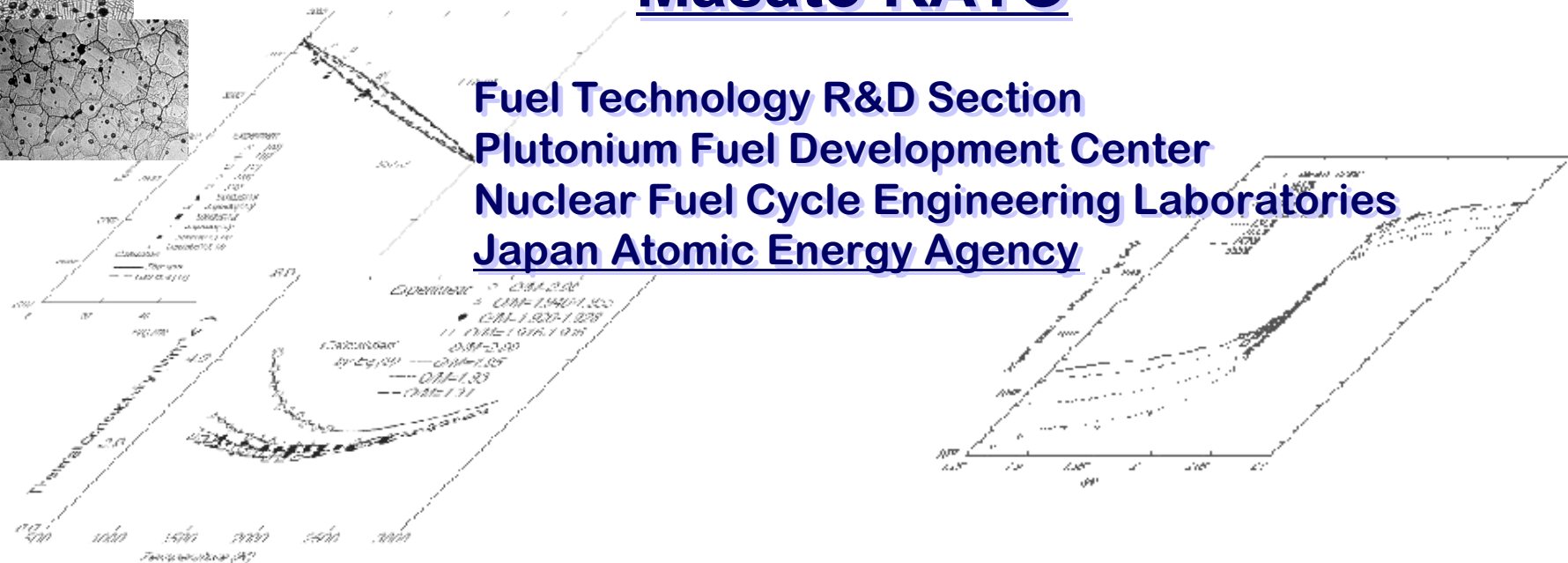
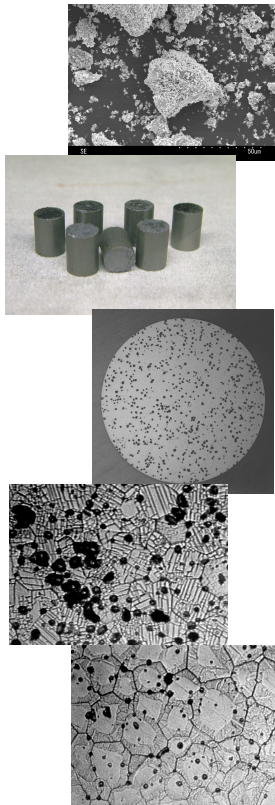


# Development of Np and Am bearing MOX fuels for Japan sodium cooled fast reactors

Masato KATO

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Plutonium Fuel Development Center  
Nuclear Fuel Cycle Engineering Laboratories  
Japan Atomic Energy Agency



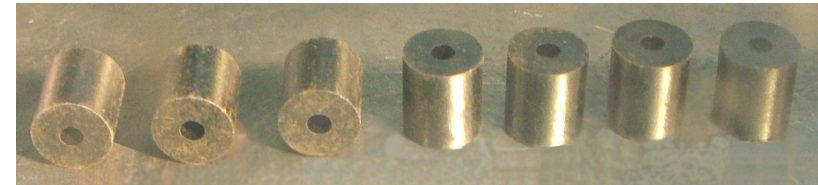
# Background

## — Development of MOX containing MA —

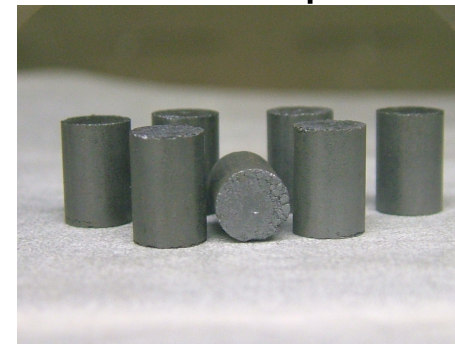
- Japan Atomic Energy Agency has developed homogeneous mixed oxide containing minor actinides (MA-MOX) for Japan sodium-cooled fast reactors.

### *Subjects to be carried out*

1. Database and models for properties
2. Advanced fabrication process
3. Analysis code of irradiation behaviour of low O/M MA-MOX pellet



Hollow MOX pellets



Pellets of  $(\text{Np}_{0.02}\text{Am}_{0.02}\text{Pu}_{0.3}\text{U}_{0.64})\text{O}_2$

### Specification of the fuel pellet

Type	: Hollow type
Pu content	: 20 - 30%
MA content	: ~ 5%(Np+Am+Cm)
Density	: 95%TD
O/M	: <1.97

# Background

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1. Database and models for properties
2. Advanced fabrication process
3. Analysis code of irradiation behaviour of low O/M MA-MOX pellet

### *Experimental evaluation*

1. Property Measurements
  - Thermodynamic data
  - Thermal properties
  - Chemical properties
2. Pellet Fabrication Test
  - Homogenization
  - O/M adjustment Technology
  - Sintering behavior
3. Irradiation test
  - Actinide redistribution
  - Microstructure change
  - FCCI

# Background

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### *Experimental evaluation*

#### 1. Property Measurements

- Thermodynamic data
- Thermal properties
- Chemical properties

#### 2. Pellet Fabrication Test

- Homogenization
- O/M adjustment Technology
- Sintering behaviour

#### 3. Irradiation test

- Actinide redistribution
- Microstructure change
- FCCI

### *Purpose of this work*

To analysis PIE results by using the measured data to develop an analysis code.



# Contents

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## ***1. Irradiation Test and PIE Results***

- Irradiation condition
- Microstructure changes
- Actinides redistribution

## ***2. Properties of the fuel pellets***

- Evaluation of properties

Lattice parameter, Melting temperature

Thermal conductivity, Oxygen potential

## ***3. Analysis of Irradiation behaviour***

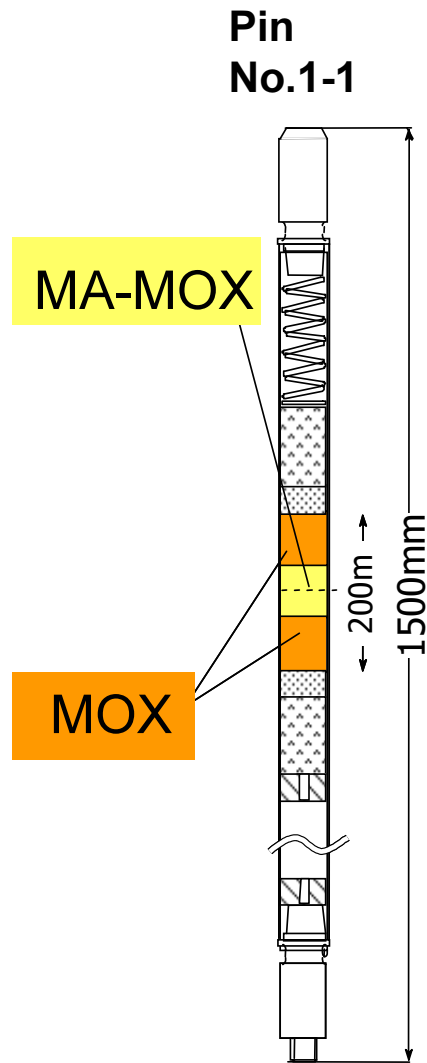
- O/M redistribution
- Relation between Pu redistribution and vapour pressure
- Evaluation of microstructure change

# Irradiation test

## Condition 1

### First Test

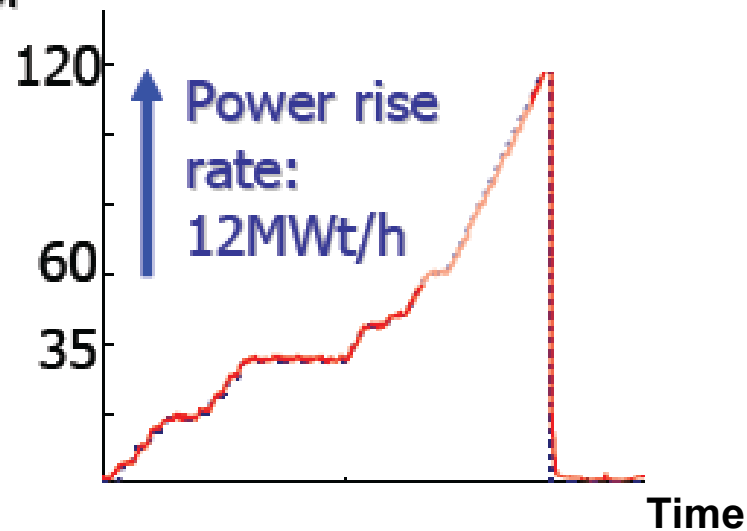
Pin No.	Name	Composition	O/M
No.1-1	MA-MOX	$(\text{Np}_{0.016}\text{Am}_{0.016}\text{Pu}_{0.3}\text{U}_{0.668})\text{O}_y$	1.98
	MOX	$(\text{Pu}_{0.3}\text{U}_{0.7})\text{O}_y$	1.98



O/M=1.98  
Fuel Pin

Held for 10min at 430W/cm

Reactor  
power (MWt)



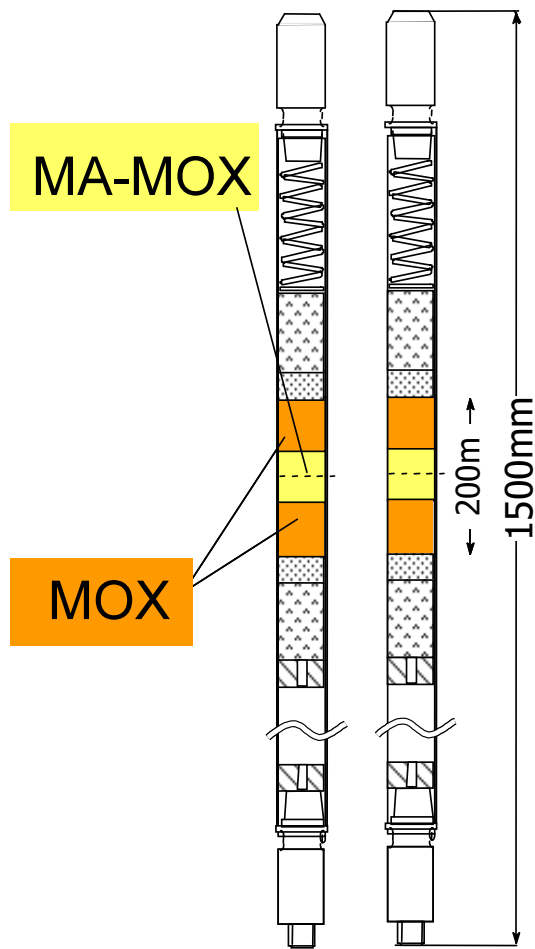
Irradiation pattern

# Irradiation test

## Condition 2

### Second Test

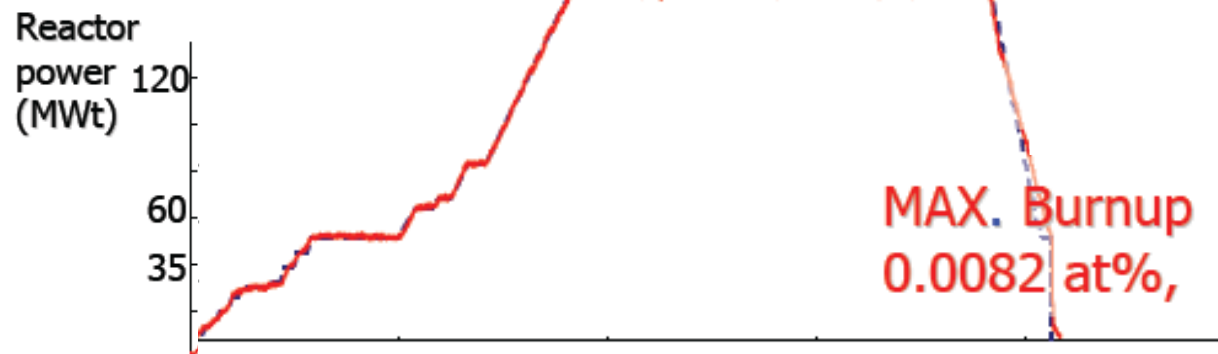
Pin No.2-1 Pin No.2-1



O/M=1.98 , 1.96  
Fuel Pins

Pin No.	Name	Composition	O/M
No.2-1	MA-MOX	$(\text{Np}_{0.016}\text{Am}_{0.016}\text{Pu}_{0.3}\text{U}_{0.668})\text{O}_y$	1.98
	MOX	$(\text{Pu}_{0.3}\text{U}_{0.7})\text{O}_y$	1.98
No.2-2	MA-MOX	$(\text{Np}_{0.016}\text{Am}_{0.016}\text{Pu}_{0.3}\text{U}_{0.668})\text{O}_y$	1.96
	MOX	$(\text{Pu}_{0.3}\text{U}_{0.7})\text{O}_y$	1.96

Held for 24h at 430W/cm

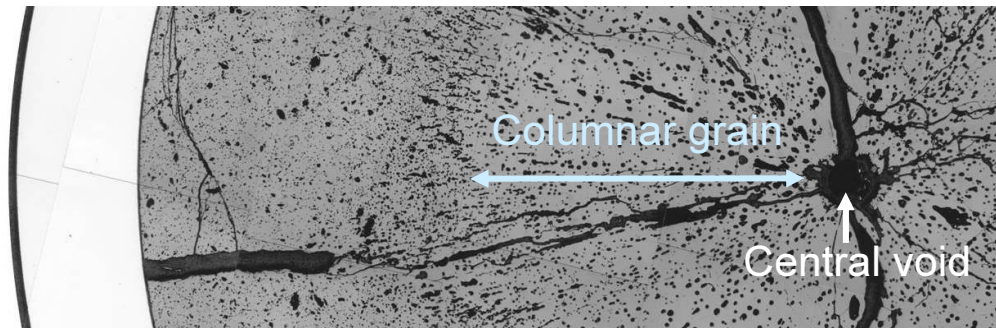


Irradiation pattern

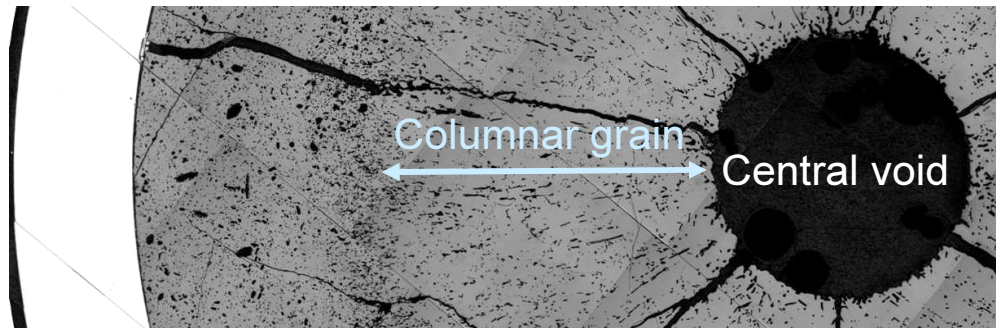
Time 7

# PIE Results

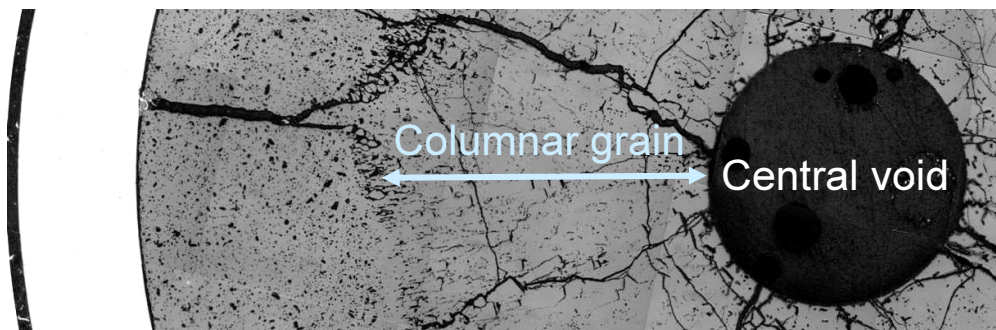
## Microstructure Change



No.1-1 MA-MOX, O/M=1.98, at 427W/cm for **10 min**



No.2-1 MA-MOX, O/M=1.98, at 432W/cm for **24h**



No.2-2 MA-MOX, O/M=1.96, at 429W/cm for **24h**

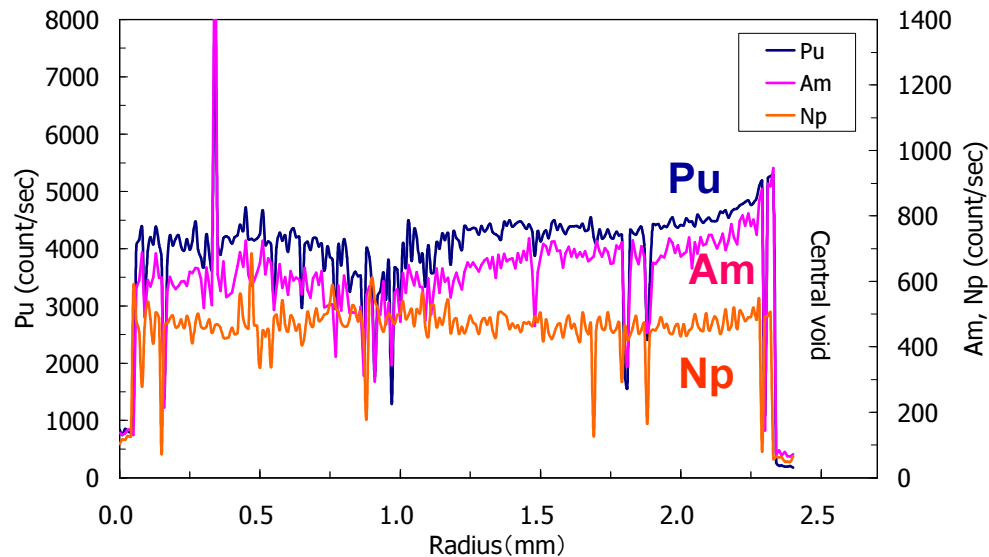
### Observation results

1. The heat rate of the fuels attained about 430W/cm.
2. The pellets were unmelted.
3. Pores migrated to the center.
4. Columnar grain and central void were observed

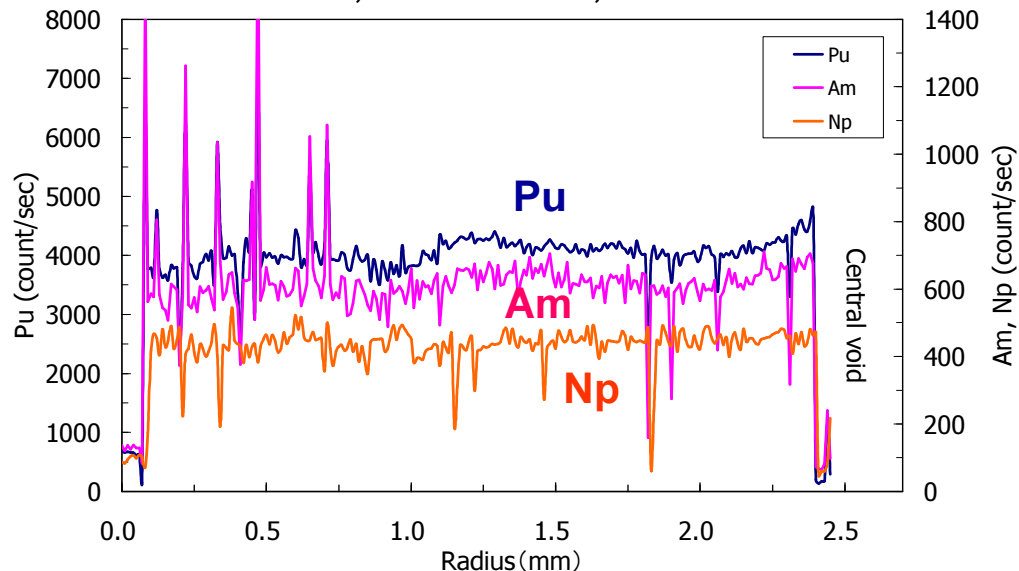


# PIE Results

## Actinide Redistribution



No.2-1 MA-MOX, O/M=1.98, at 432W/cm for 24h



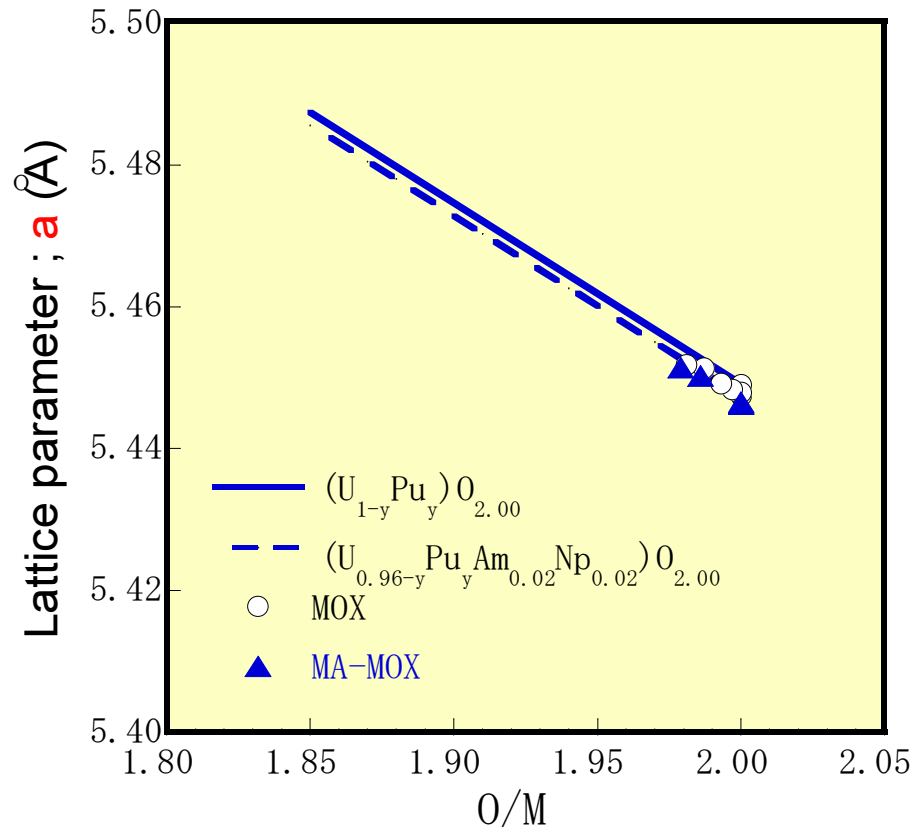
No.2-2 MA-MOX, O/M=1.96, at 429W/cm for 24h

### Observation results

1. The content of Pu and Am increased at pellet center.
2. The increment of Pu and Am decreased with decreasing O/M.
3. The contents of Np and U were flat.

# Properties of fuel pellets (1/4)

## Lattice Parameter



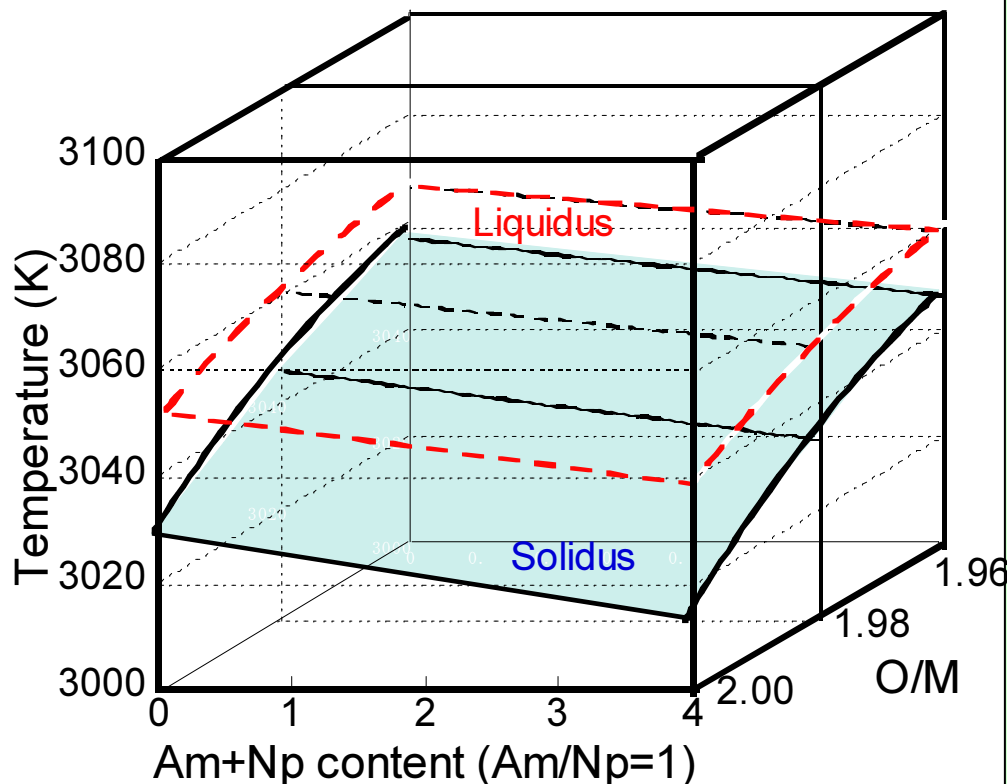
Lattice Parameter vs. O/M ratio

$$a = 4 / \sqrt{3} (r_c (1 + 0.112x) + r_a)$$

$$r_c = (r_U (1 - z - y' - y'') + r_{Pu}z + r_{Am}y' + r_{Np}y'')$$

## Results

1. The model to represent the lattice parameter was derived.
2. MA content slightly caused the lattice parameters to decrease.
3. The difference of theoretical density is 0.01-0.02g/cm<sup>3</sup>.

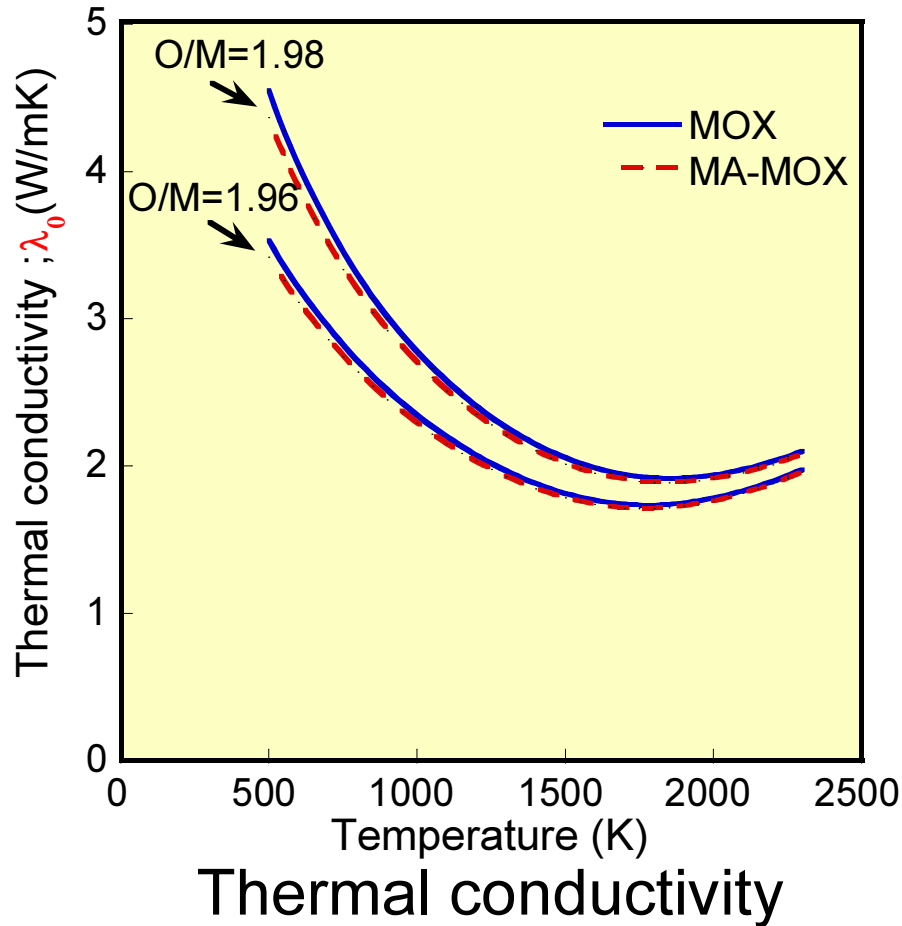


Variation of Solidus and Liquidus Temp.

M.Kato, et al., ICAPP '09

### Results

1. The data were measured and analyzed by the ideal solid solution model.
2. MA content caused to decrease by 2-3K /%MA.
3. The solidus temperature of MA-MOX is over 3000K.



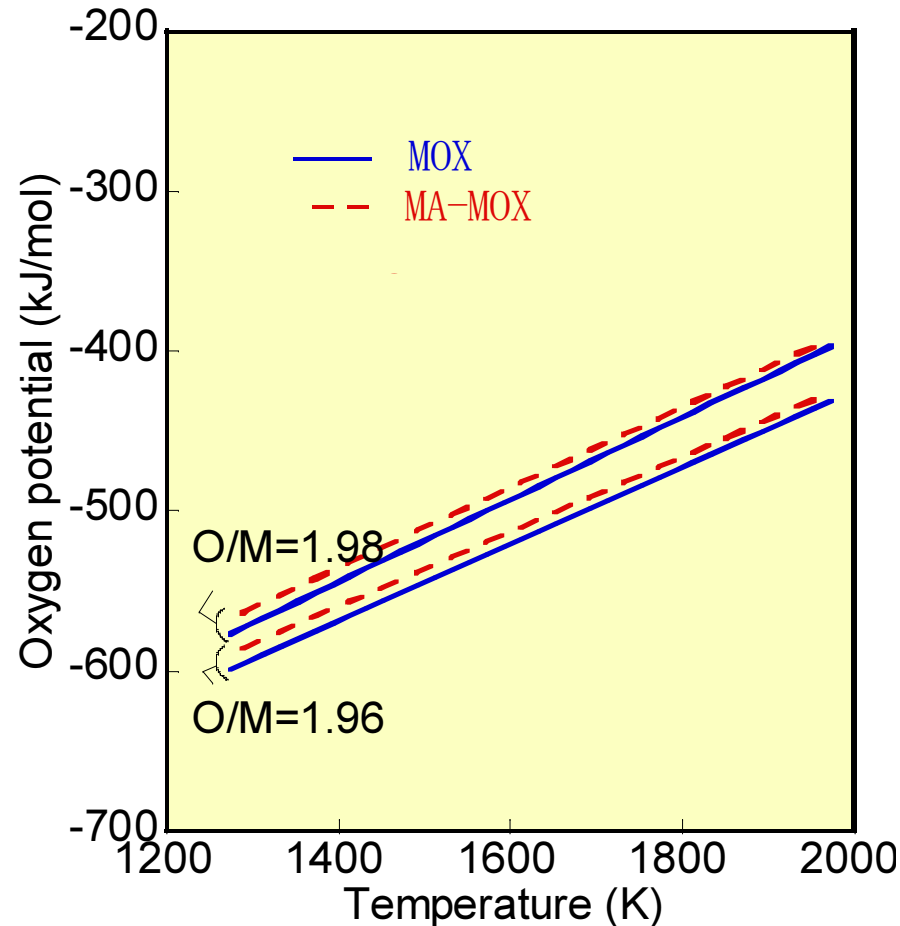
## Results

1. The data were measured as functions of MA content, density, O/M and temperature.
2. MA content caused to decrease slightly in temperatures of less than 1000K.

$$\lambda_0 = \frac{1}{(2.713x + 3.583 \times 10^{-1} \times z_1 + 6.317 \times 10^{-2} \times z_2 + 1.595 \times 10^{-2}) + (-2.625x + 2.493) \times 10^{-4} T} + \frac{1.541 \times 10^{11}}{T^{5/2}} \exp\left(\frac{1.522 \times 10^4}{T}\right)$$

M.Kato, et al., OECD/NEA, Oct. 6-10, 2008, Japan

z<sub>1</sub>: Am content  
z<sub>2</sub>: Np content  
x: Deviation x in MO<sub>2-x</sub>

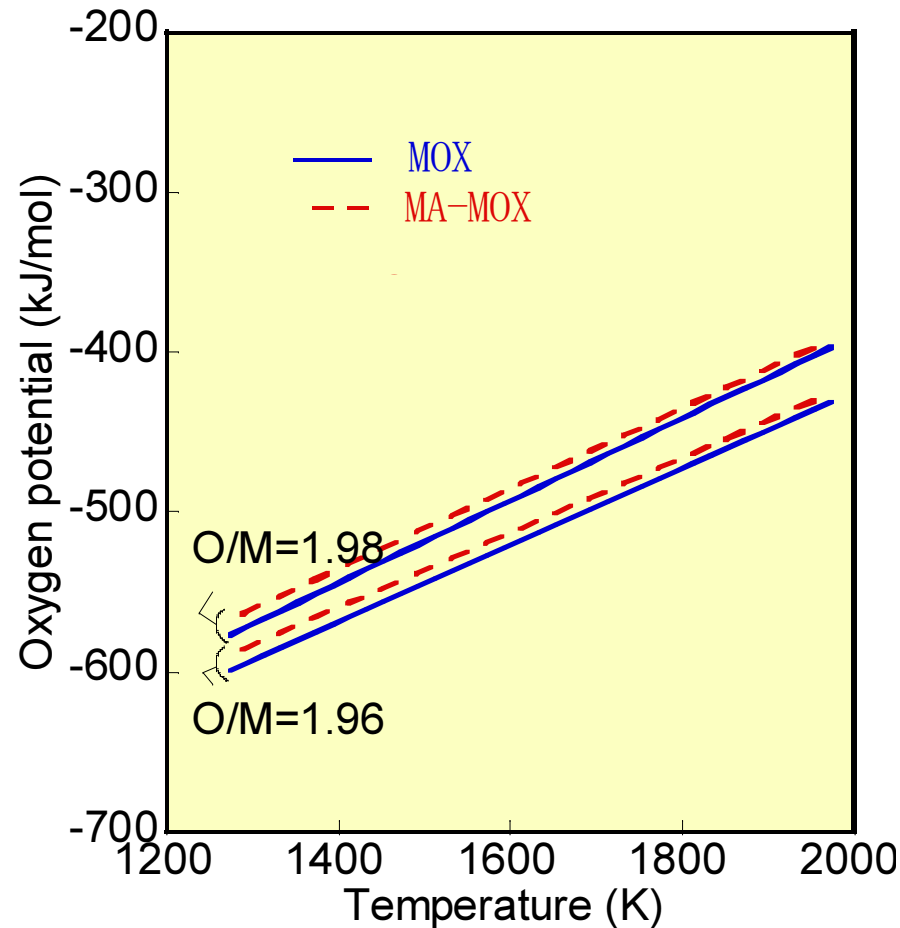


## Results

1. The data were measured by gas equilibrium method.
2. MA content caused the oxygen potential to increase slightly.

## Oxygen potential

M.Kato, et al., J.N.M. 385 (2009) 419



Oxygen potential

## Results

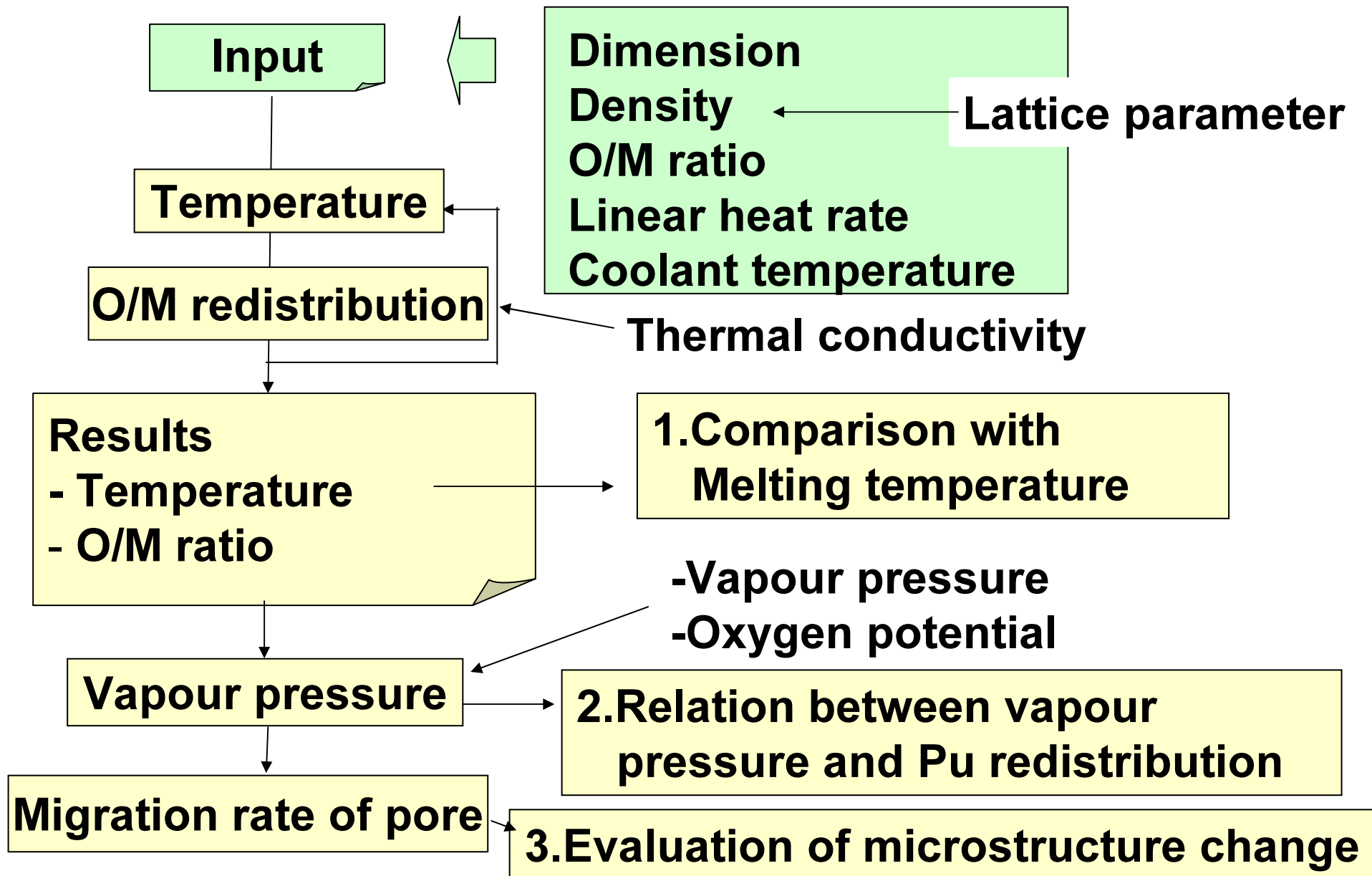
1. The data were measured by gas equilibrium method.
2. MA content caused the oxygen potential to increase slightly.

## Summary

The effect of MA addition on the properties is negligibly small in the operation temperature range of FR fuels

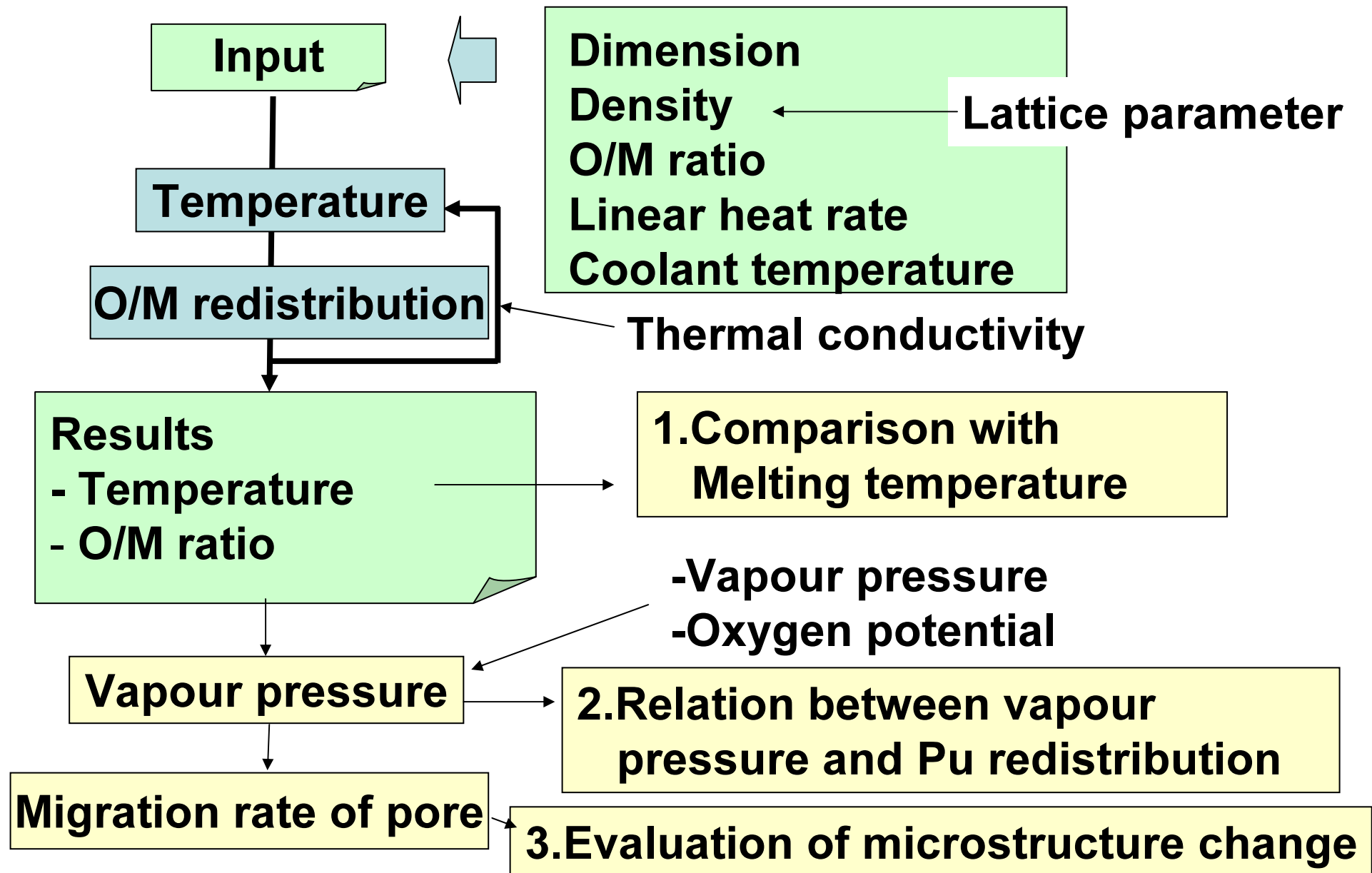


# Analysis of irradiation behaviour Analysis Procedure





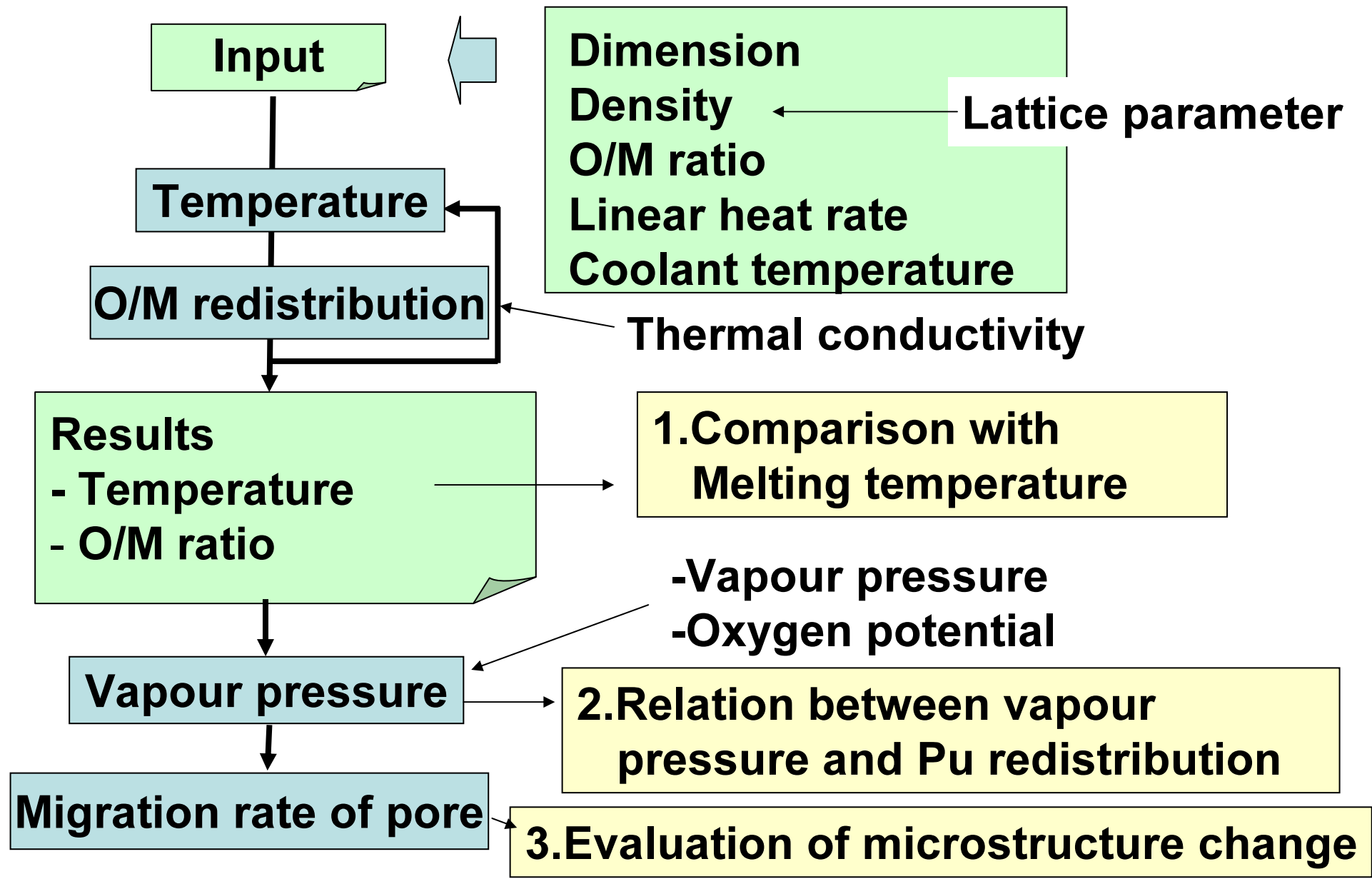
# Analysis Procedure





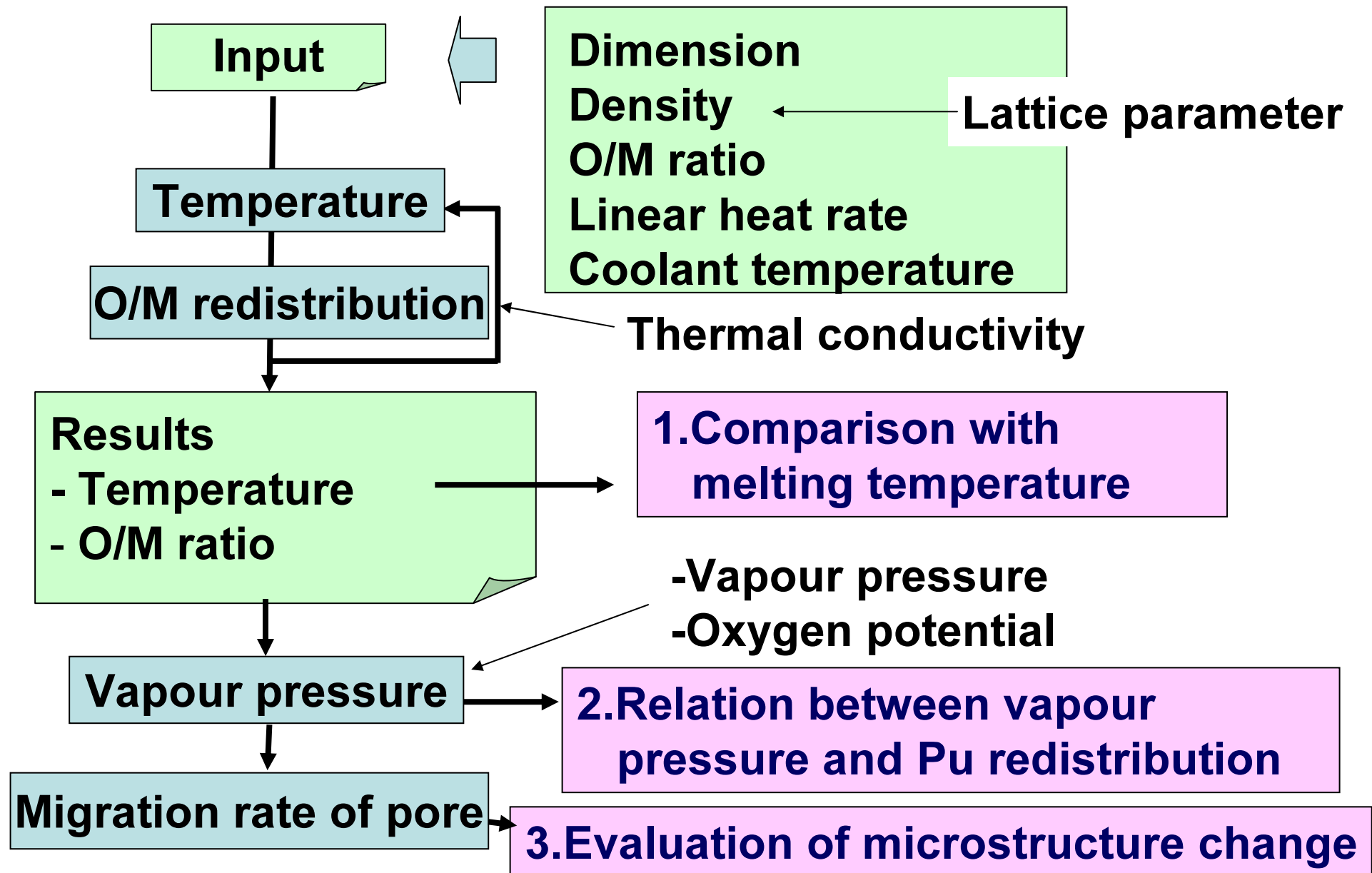


# Analysis Procedure



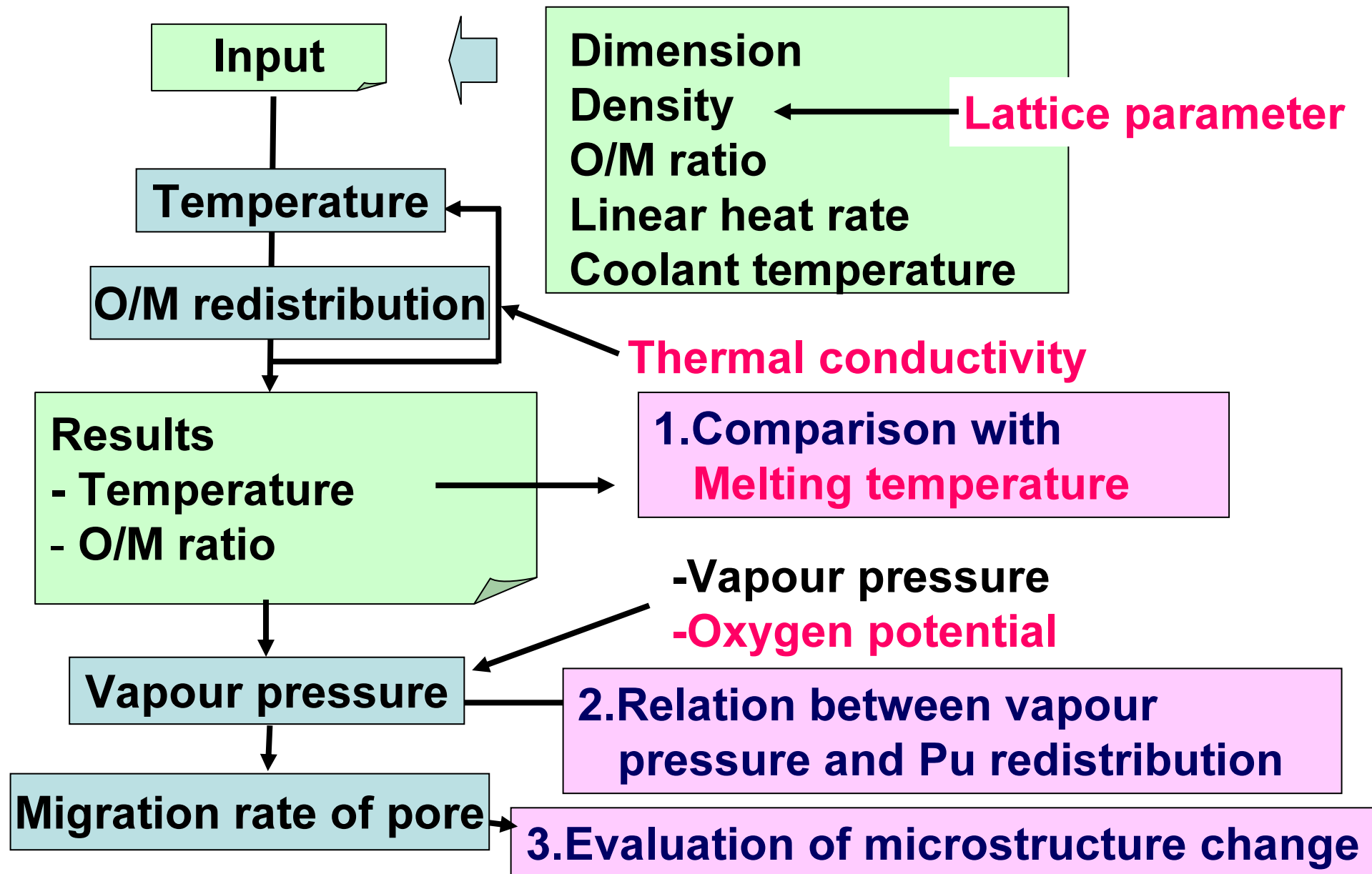


# Analysis Procedure



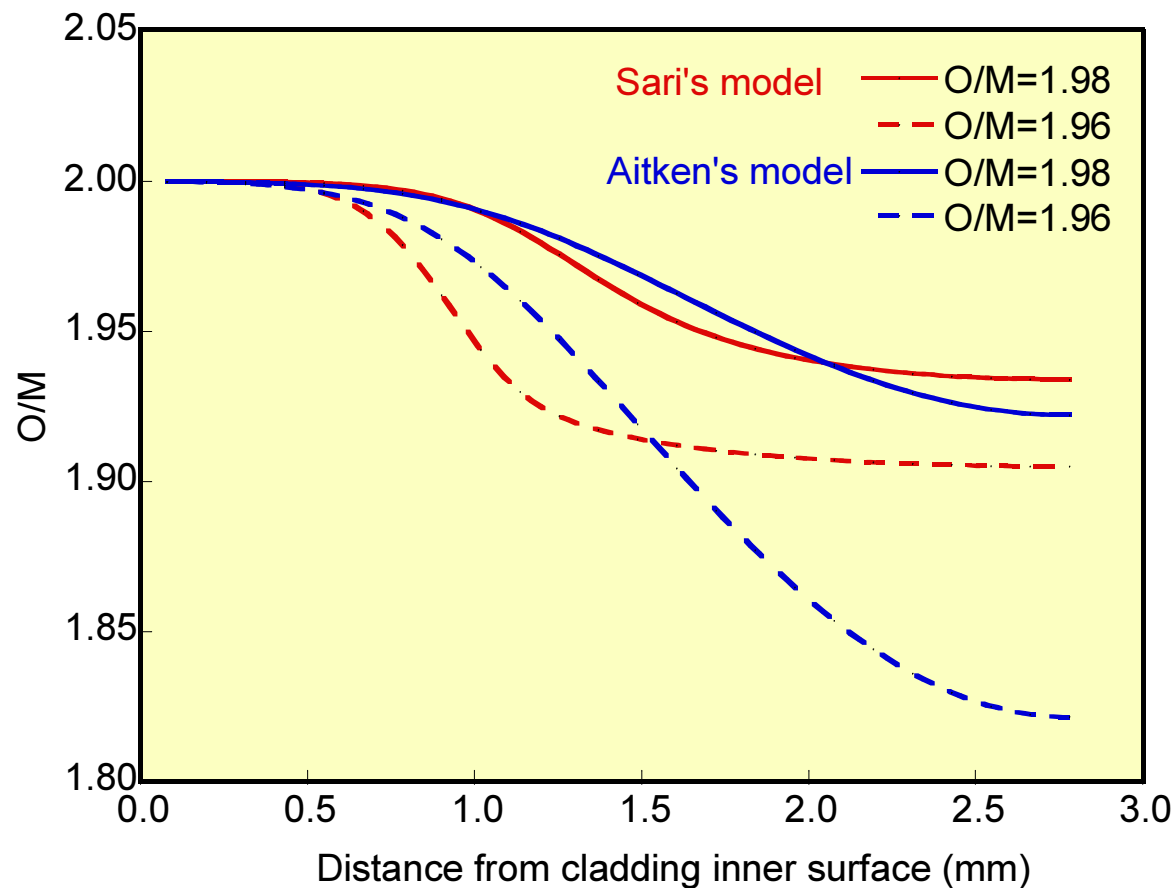


# Analysis Procedure





# O/M Redistribution

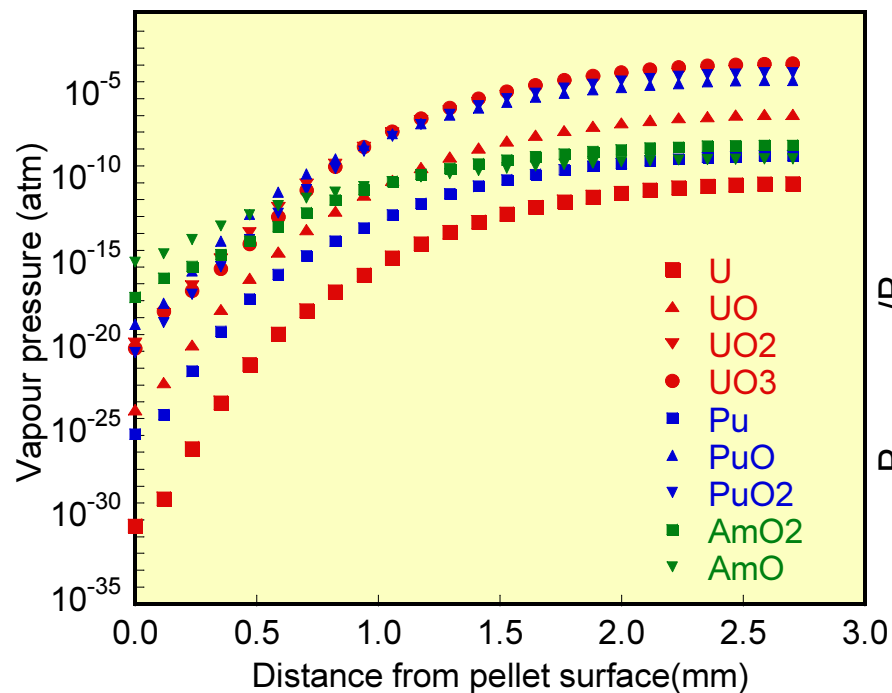


O/M ratio distribution in the radial direction

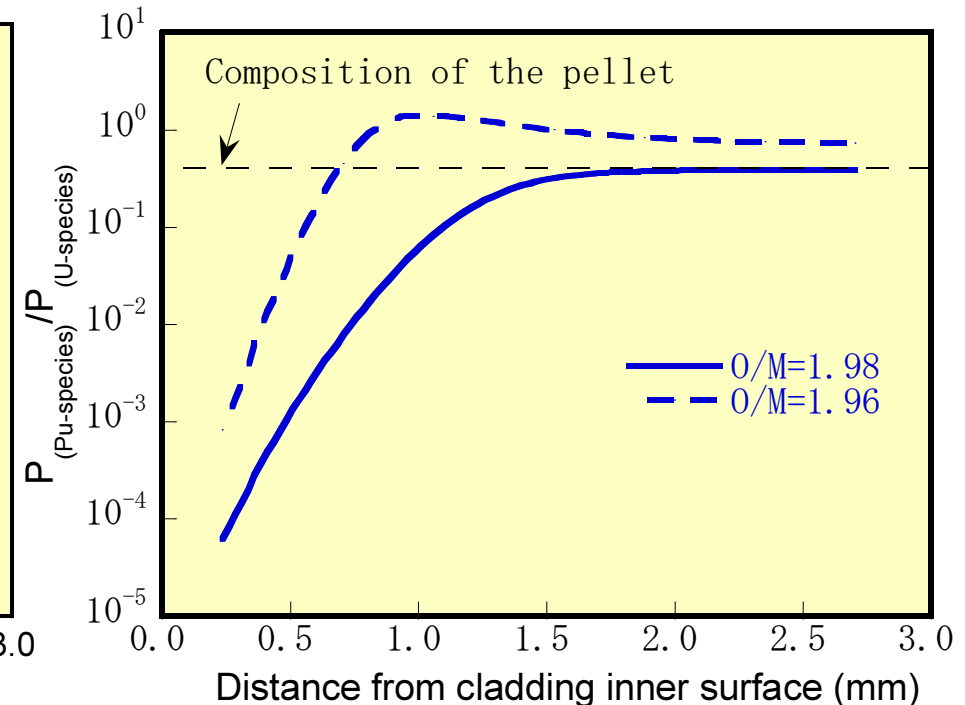
## Results

1. Both models gave almost the same result in high O/M pellets.
2. The results of low O/M pellets differed according to models.
3. Sari's model was employed.

# Evaluation of vapour pressure



Variation of vapor pressure

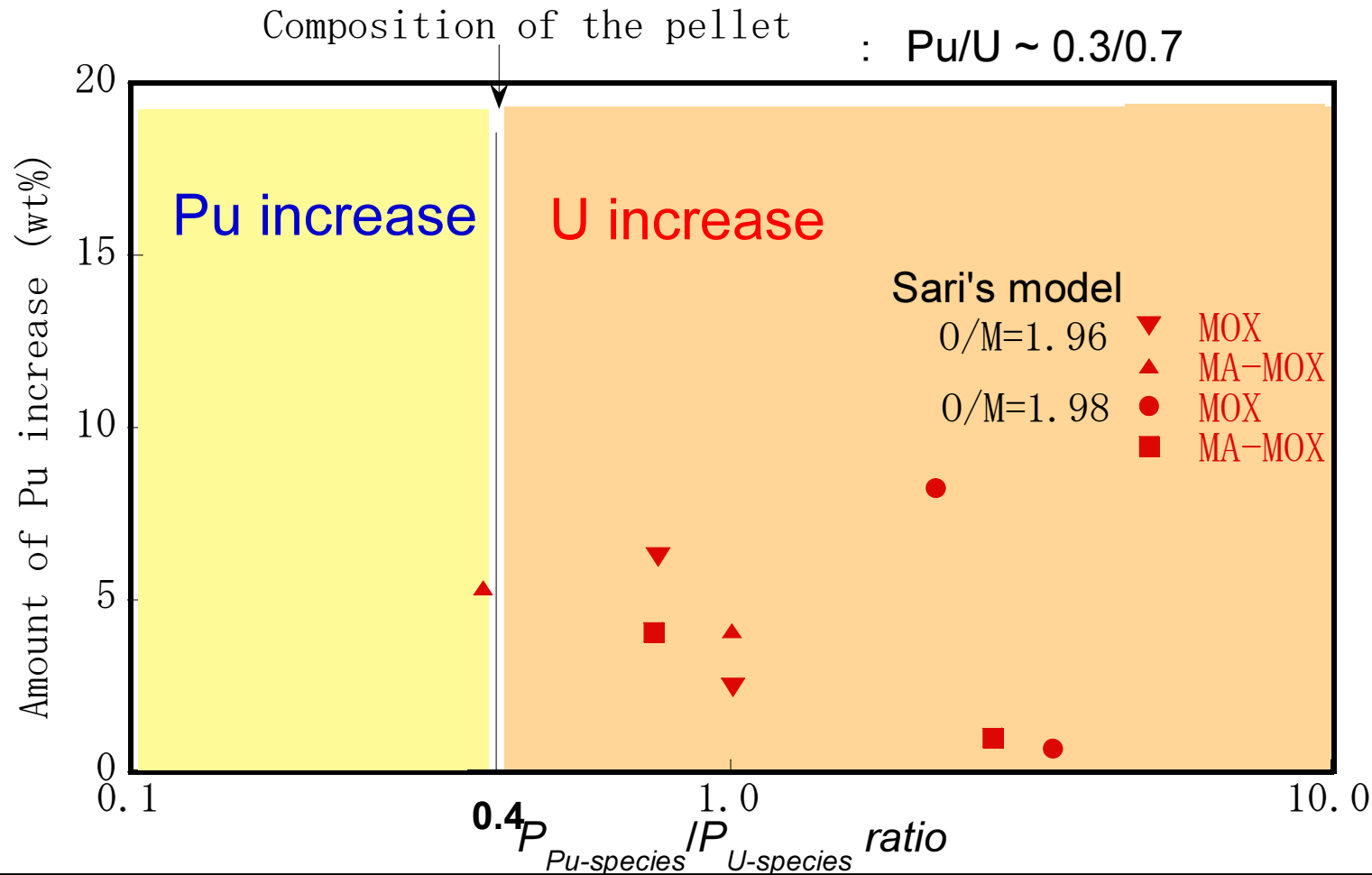


Variation of Pu-/U- species

## Result

1. Variation of vapor pressure ratio of Pu/U-species was evaluated using Rand –Markin model.

# Evaluation of Pu redistribution



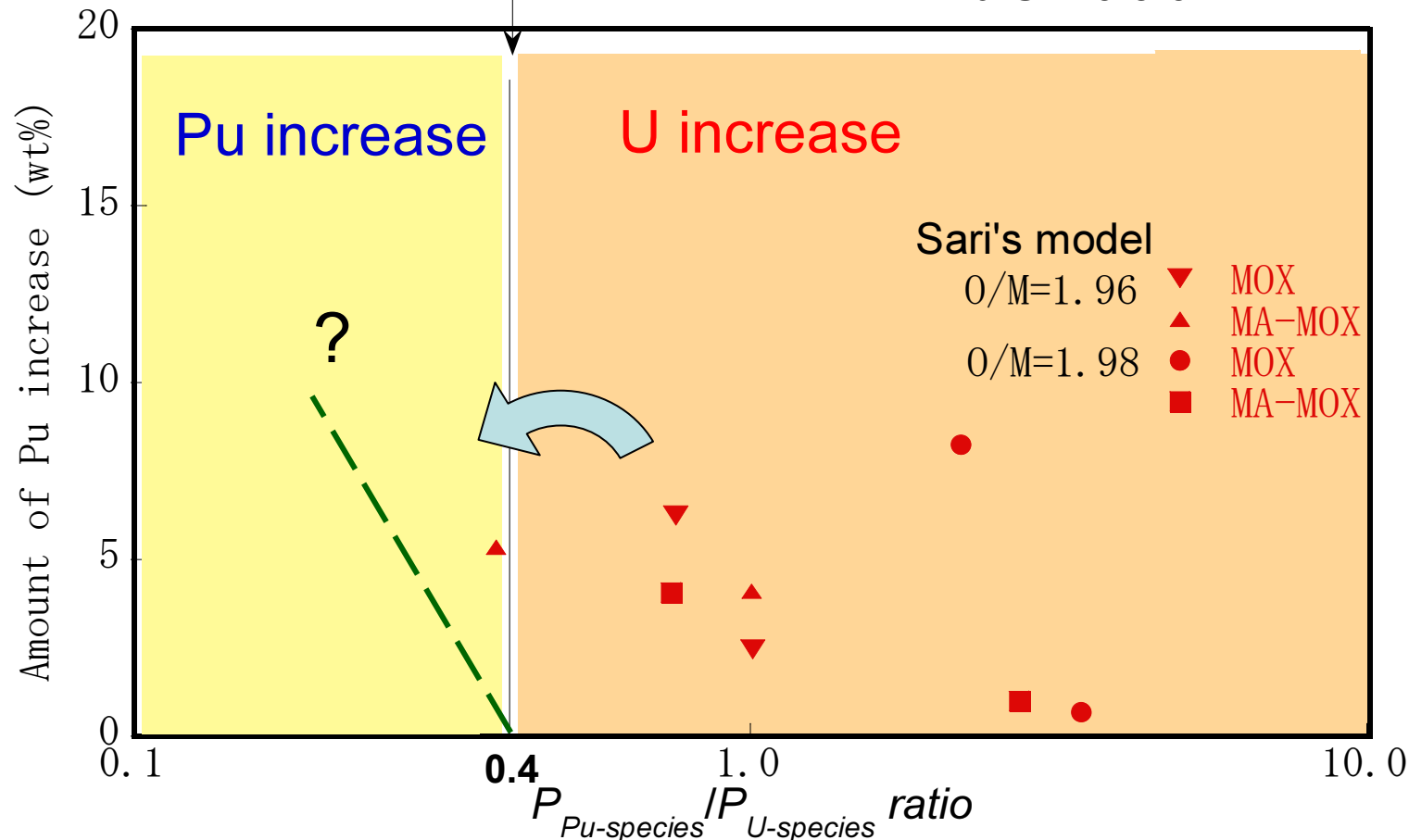
## Result

1. Calculation result shows that U content increases at pellet center.



# Evaluation of Pu redistribution

Composition of the pellet : Pu/U ~ 0.3/0.7



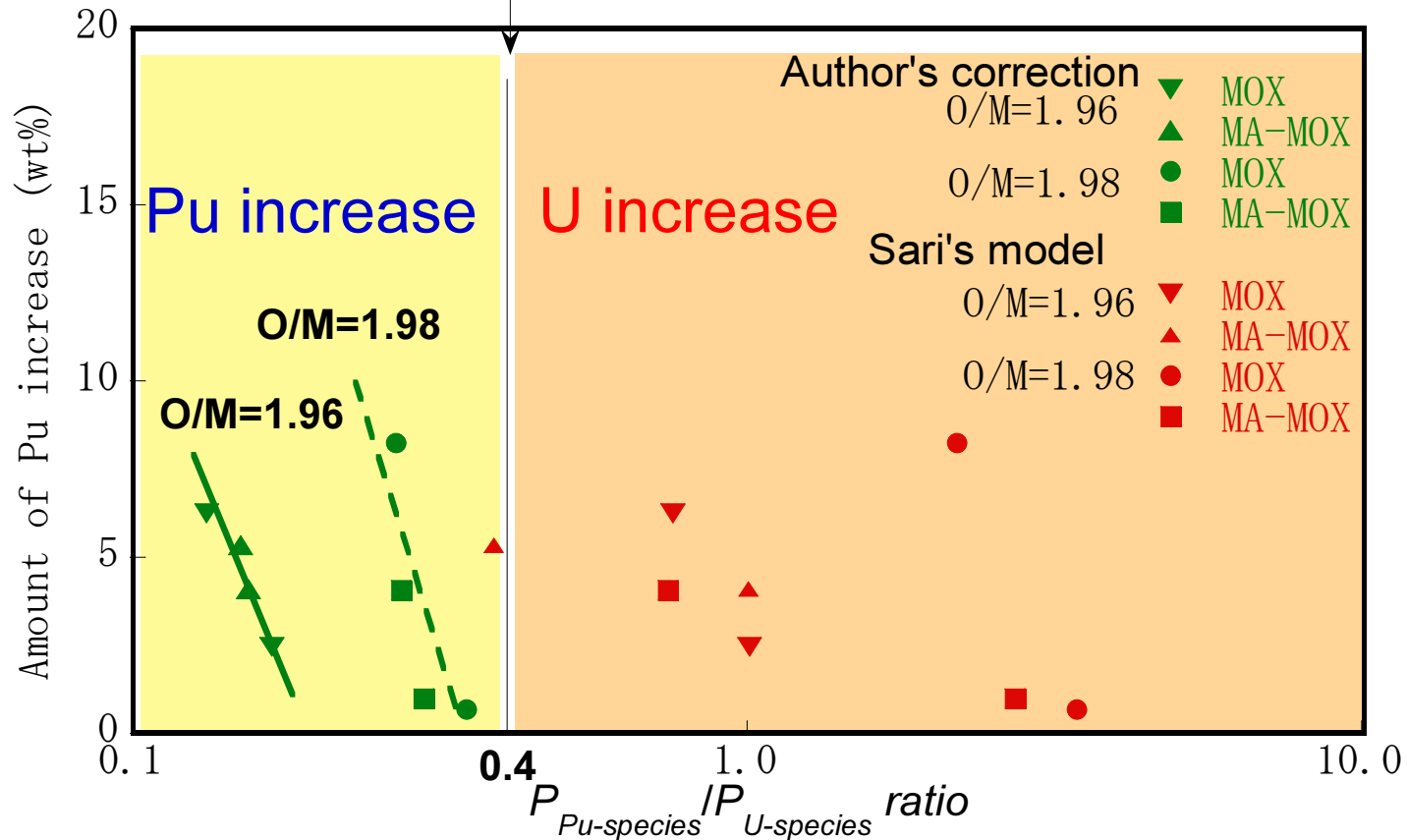
## Expectation

Increment of Pu should be increased with decreasing the  $P_{Pu-species} / P_{U-species}$ .



# Evaluation of Pu redistribution

Composition of the pellet : Pu/U ~ 0.3/0.7



## Results

1. Calculation results show that Pu content increase at pellet centre.
2. Two lines were obtained depending on O/M ratio.



# Reviewing of the model for O/M redistribution

Sari's model for O/M redistribution

$$\ln\left(\frac{x_1}{x_2}\right) = \frac{Q^*}{R} \left( \frac{1}{T_2} - \frac{1}{T_1} \right)$$

$$Q^* = -9.45 \times 10^5 + 5.66 \times 10^5 V_{Pu} - 8.5 \times 10^4 V_{Pu}^2$$

X: deviation in (U,Pu)O<sub>2-x</sub>

T: temperature

R : gas constant,

**Q\*** : the molar effective heat of transport

V<sub>Pu</sub> : Pu valence

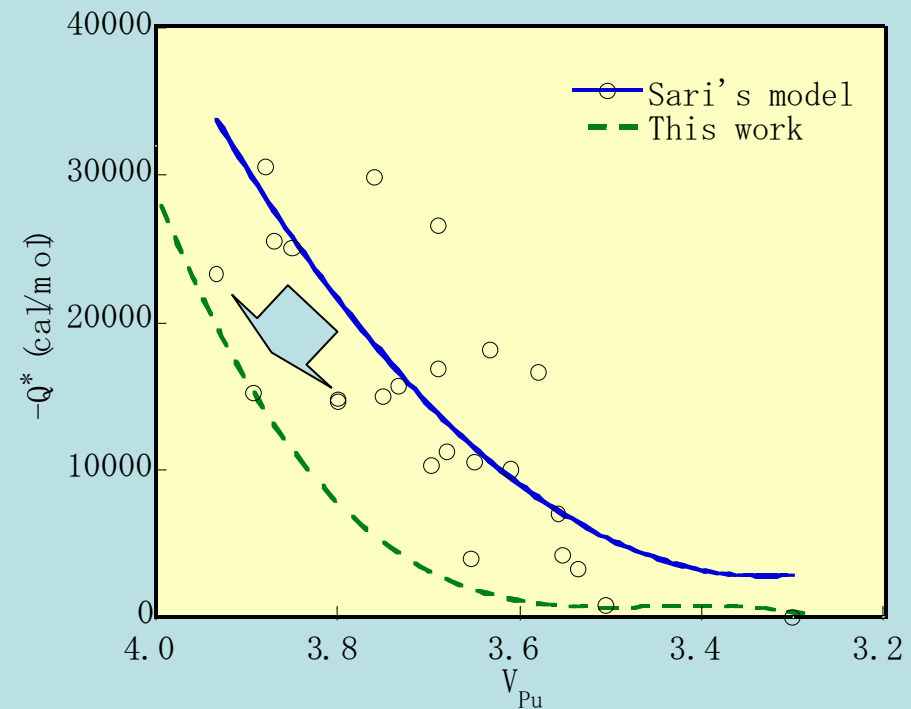
Evaluation of equation for Q\*

Calculation of O/M Redistribution, temperature and vapour pressure

Evaluation of the relationship between the  $P_{Pu\text{-species}}/P_{U\text{-species}}$  and the Pu increase

Corrected Eq.

$$Q^* = -7.01 \times 10^6 + 6.1012 \times 10^6 V_{Pu} - 1.7705 \times 10^6 V_{Pu}^2 - 1.7122 \times 10^5 V_{Pu}^3$$



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Evaluation of equation for  $Q^*$

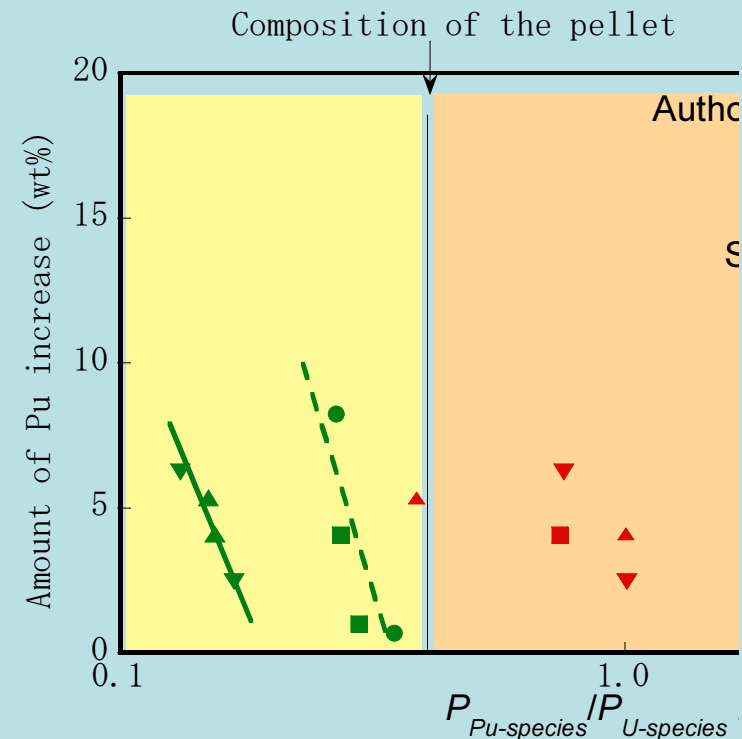
Calculation of O/M Redistribution, temperature and vapour pressure

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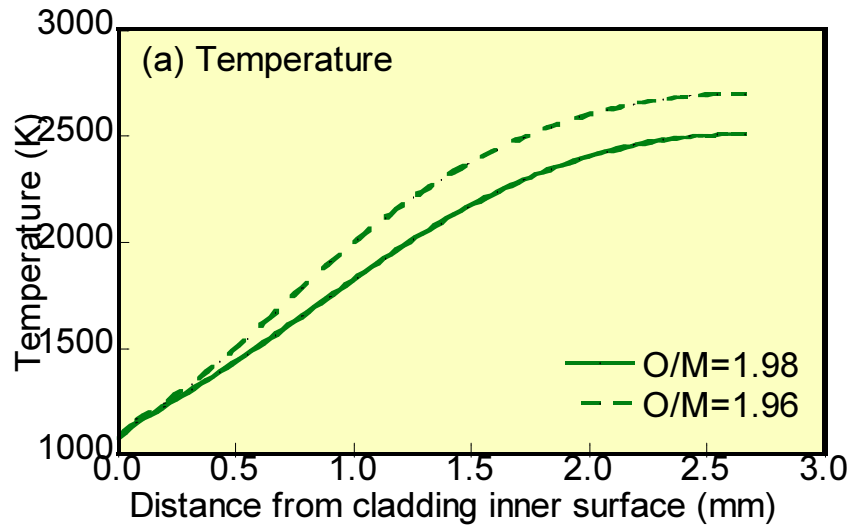
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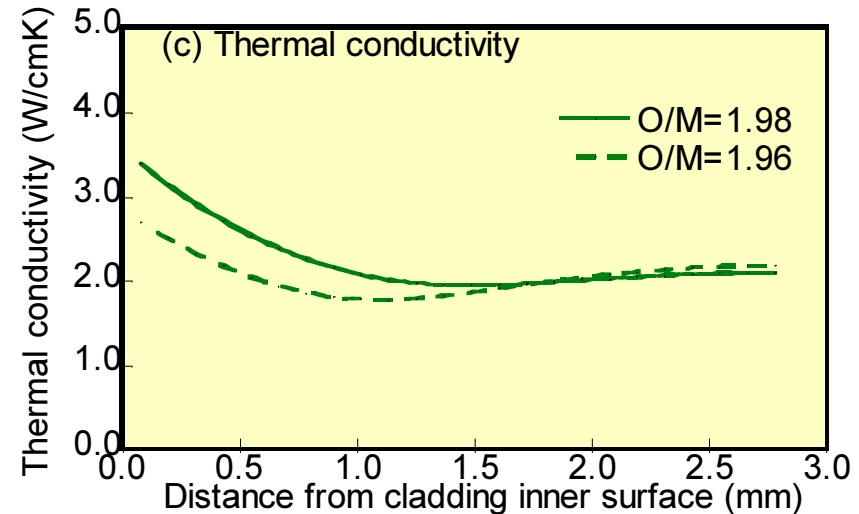
X: deviation in  $(U,Pu)O_{2-x}$   
 T: temperature  
 R : gas constant,  
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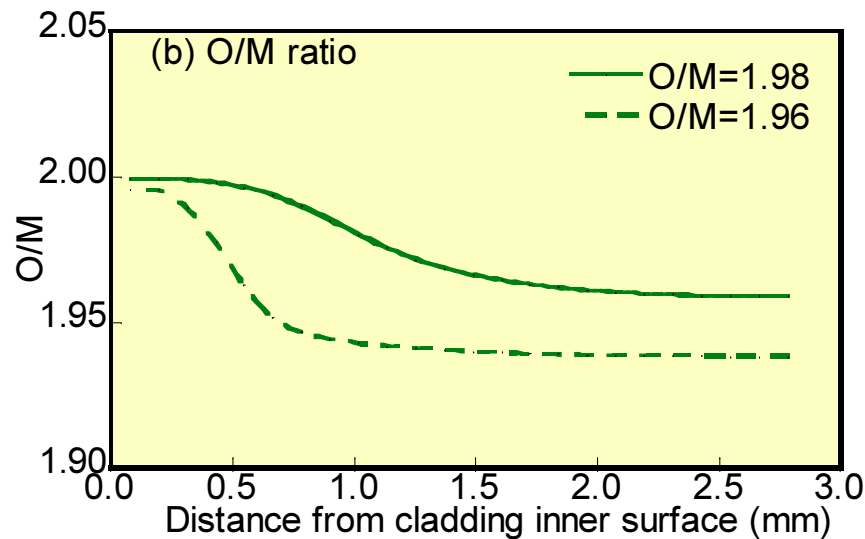
# Calculation results in MA-MOX pellets



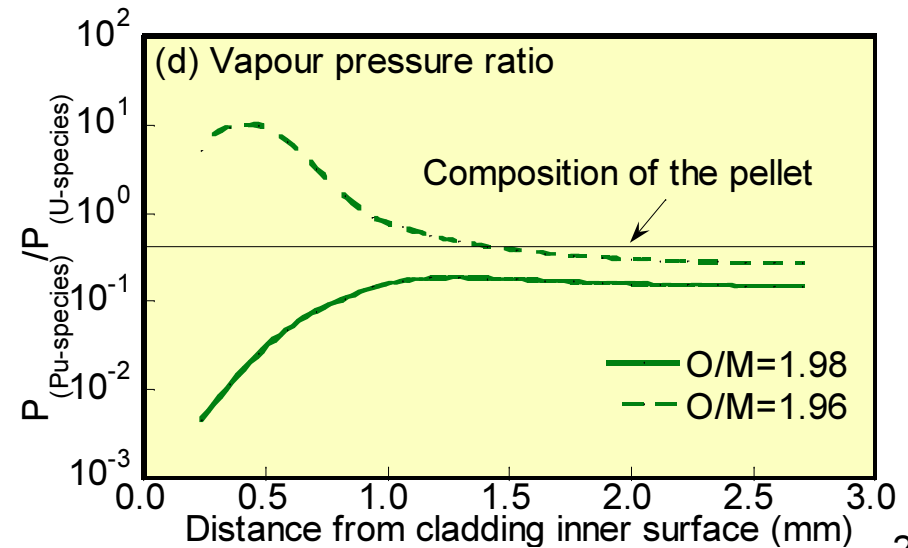
**Temperature**



**Thermal conductivity**

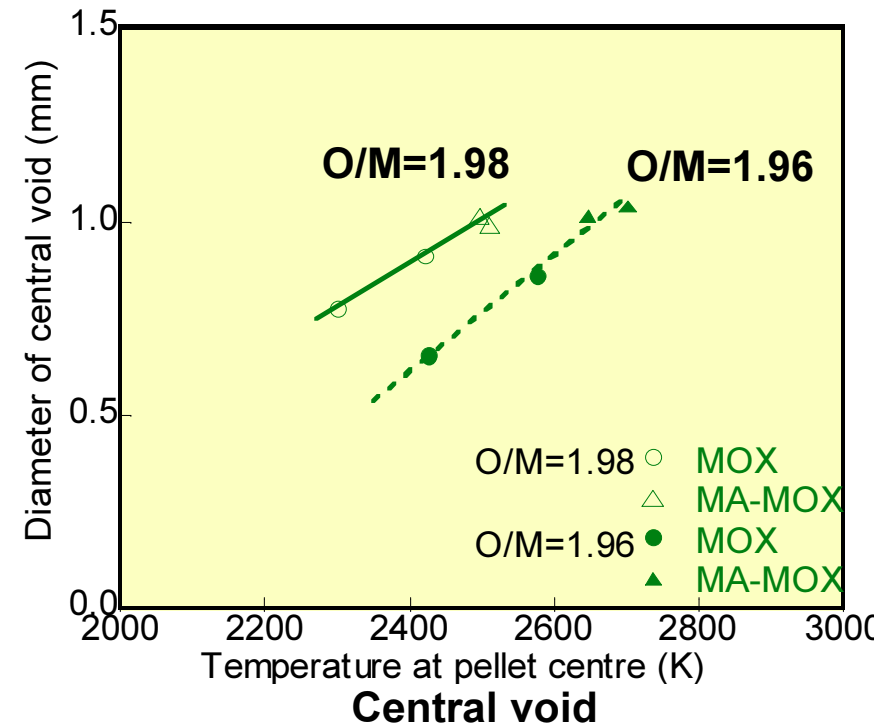
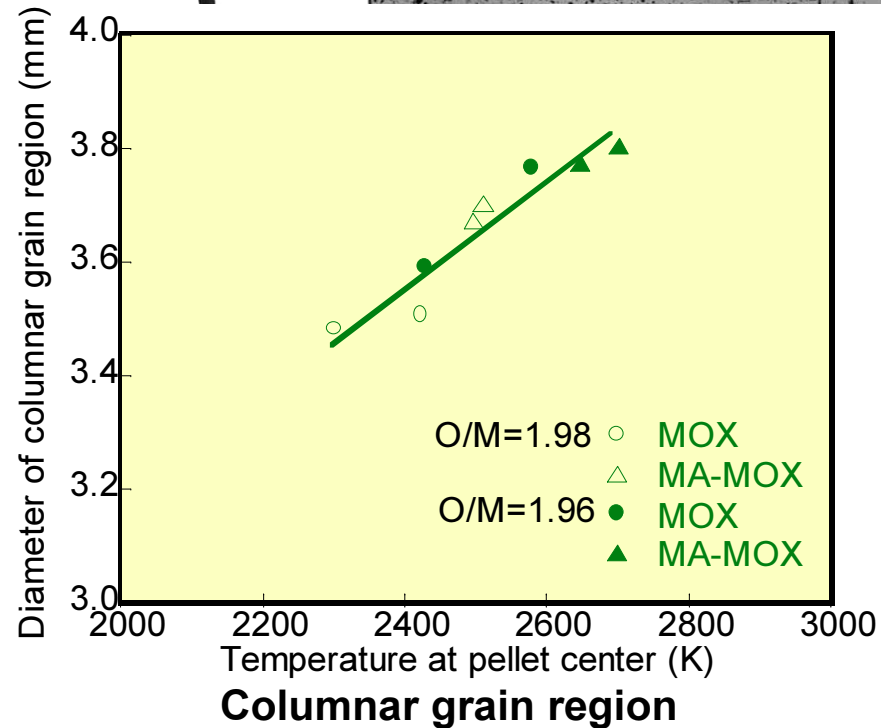
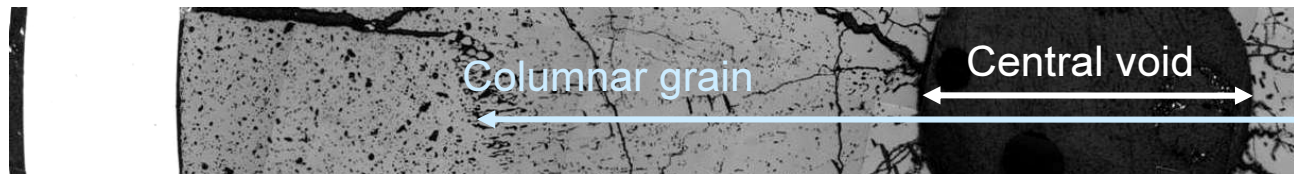


**O/M ratio**



**Vapour pressure ratio of Pu-/U-species**

# Microstructure change



## Results

1. The both region grew with temperature.
2. The central void of high O/M pellet is larger than that of low O/M pellets.



# Summary

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- 1. The physical properties of the Np/Am-MOX were evaluated, and the effect of Np/Am addition was negligibly small.***
- 2. The relation between the Pu redistribution and the vapor pressure was described by correcting Sari's model for O/M redistribution.***
- 3. The low O/M pellet attained higher temperature, however, the diameter of the central void was small as compared with the high O/M pellet.***
- 4. It is needed to measure experimental data and to derive advanced models for describing irradiation behavior of low O/M fuel.***

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