



Acknowledgments:



# From sources to deposits: Recent advances about the unconformity-related U deposits

Julien Mercadier, Antonin Richard, Michel Cathelineau, Marie-Christine Boiron, Irvine R. Annesley<sup>a</sup>, Michel Cuney

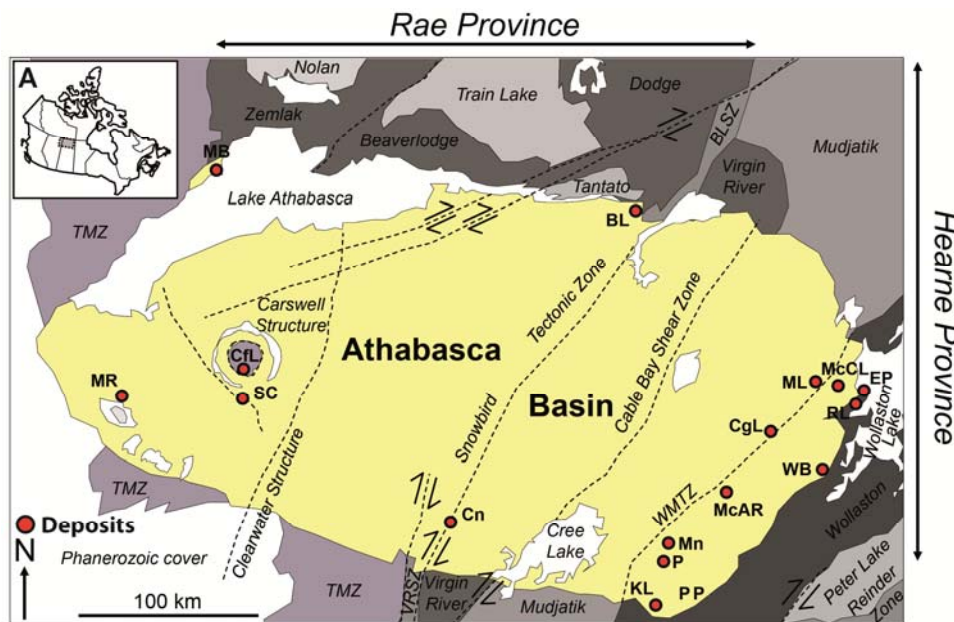
GeoRessources, Université de Lorraine, CNRS, CREGU, France

<sup>a</sup>: Department of Geological Sciences, University of Saskatchewan, Canada

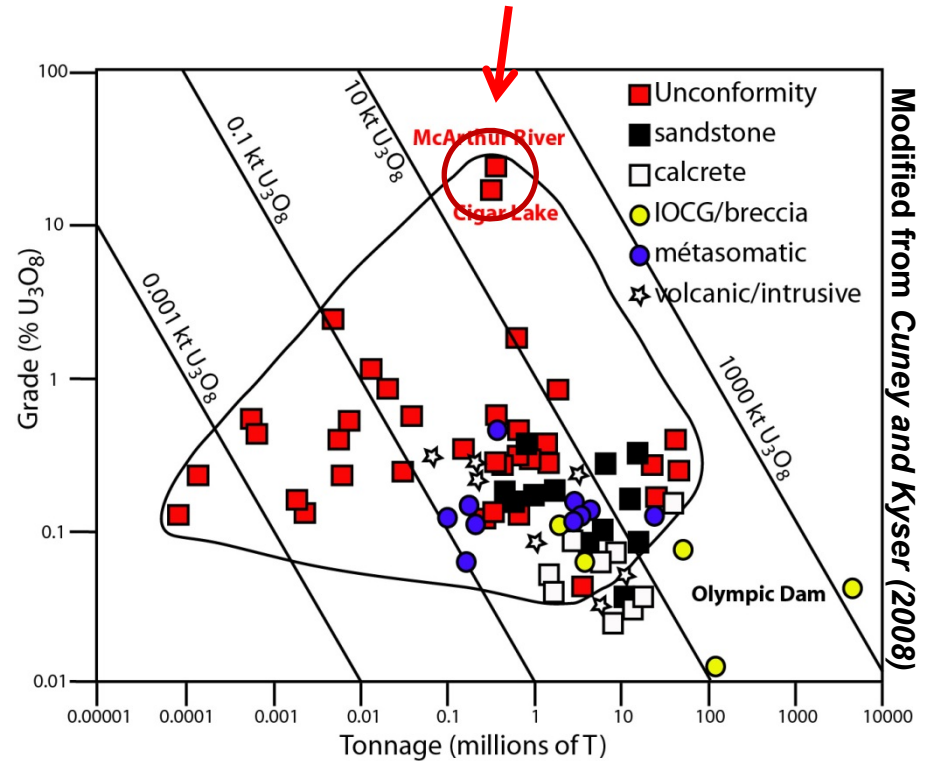
URAM 2014, Vienna, Austria

# The unconformity-related U deposits

## Athabasca Basin – Canada



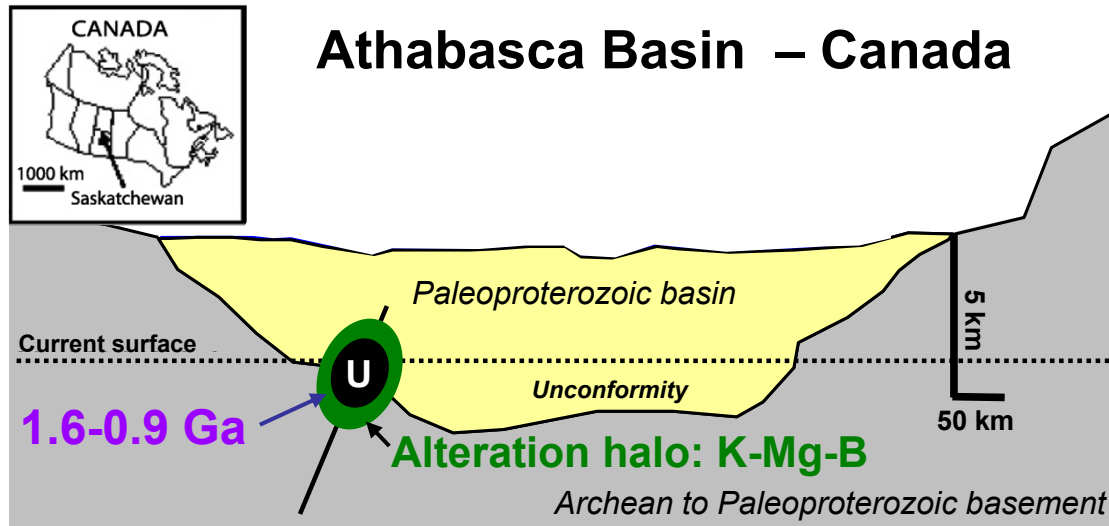
**200 kT U @ 20%**



**> 30 deposits**  
**600 000 T U @ 2% U**  
**33% of world resources**

**Exceptional high-grades**  
**Good tonnages**

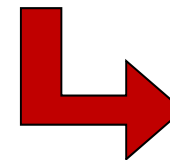
# The unconformity-related U deposits



$[U]_{\text{crust}} \sim 1.7 \text{ ppm}$

$[U]_{\text{unconformity-related}} \sim 20\%$

$\times 10^5$



Uranium oxide

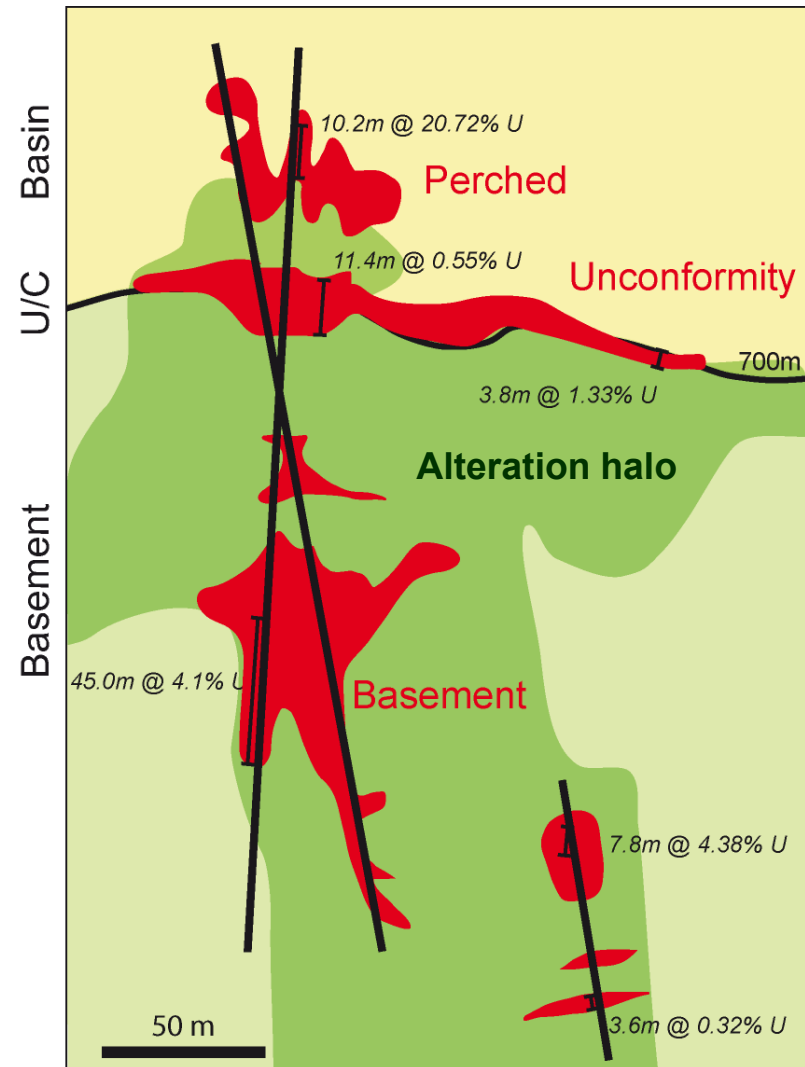
*How to reach these extreme grades?*

## Geochemical cycle of Uranium: Sources – Transport - Deposition

## Current knowledges

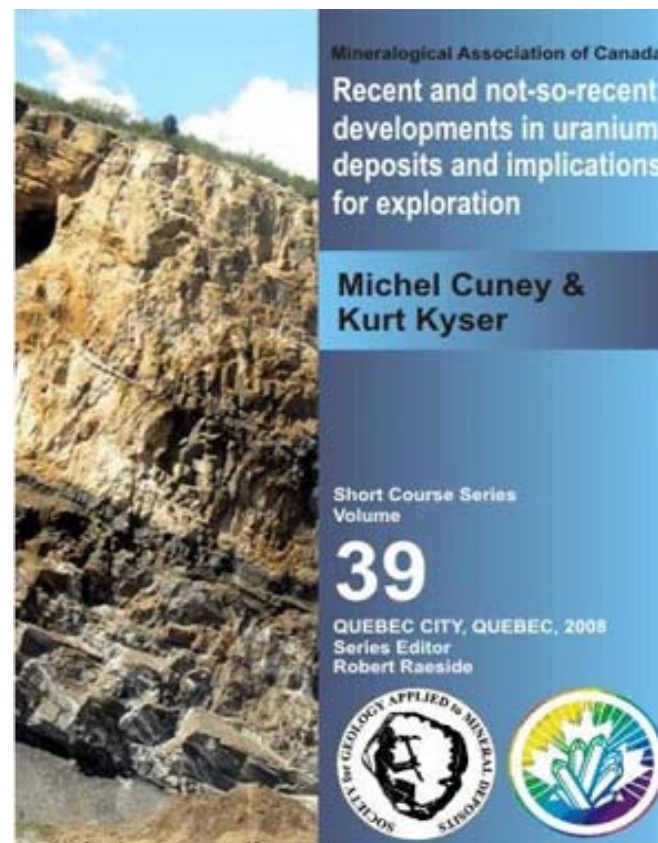
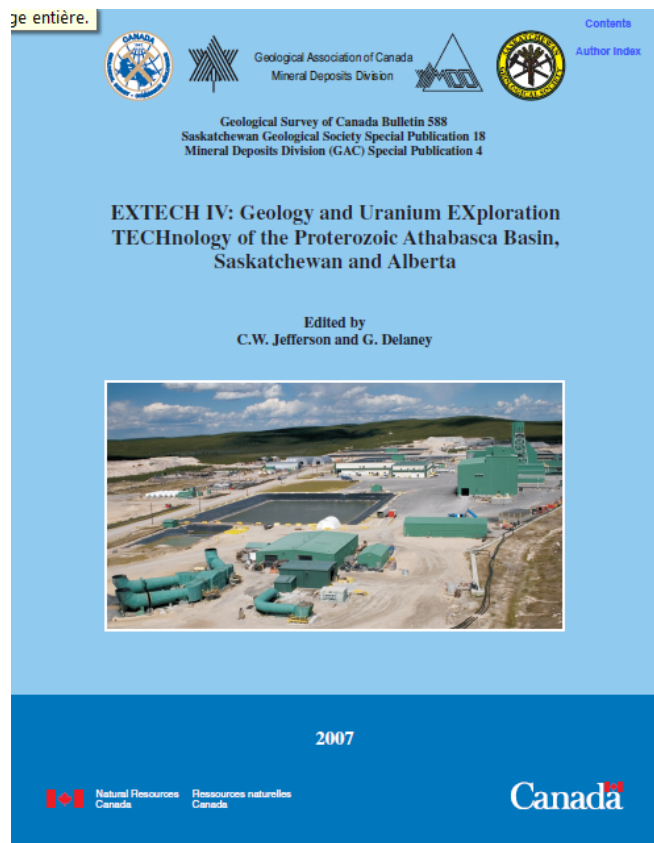
- Hydrothermal deposits (< 250°C)
- U-Pb ages: 1.6-0.9 Ga + late remobilizations
- 3 locations
- Structure-related, graphite
- Strong K-Mg-B alteration : illite-chlorites-dravite/Mg-foitite (Mg Tour)-hydrothermal quartz
- Mineralizing fluids: Na-Ca-rich brines (25-35 wt% eq. NaCl)

Precious witnesses of extremely efficient  
and large-scale fluid flows at the basin /  
basement interface



Kiana deposit (Shea Creek)  
AREVA - UEX

## These deposits : well-studied and a lot known about them



**However still relevant scientific/geological questions**

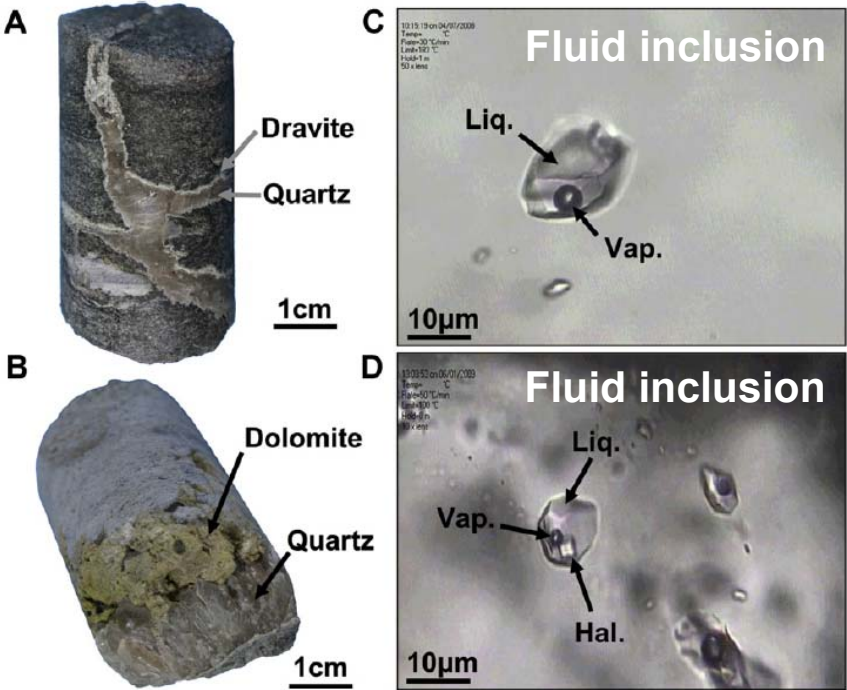
**Can we provide new insights about:**

- the origin of the brines?
- the source of U and other metals?
- the percolation conditions within the basement rocks?
- the chemical modifications at the origin of the formation of mineralizing brines?
- the conditions for the transport and deposition of uranium?

**For the brines, where to find these new information?**

**1. Fluid inclusions = aliquot of the mineralizing brines**

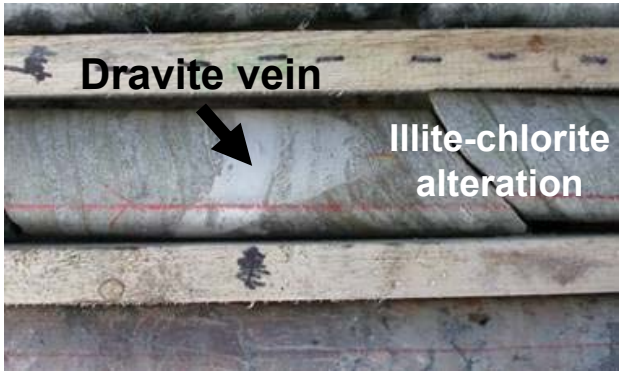
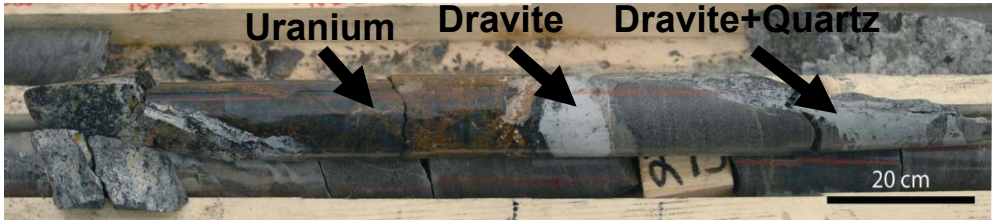
Ex: quartz and dolomite veins



P-Patch and Millennium deposits (East Athabasca)

**2. Hydrothermal minerals contemporaneous of UO<sub>2</sub>**

Ex: dravite/ Mg foitite (Mg-tourmaline)

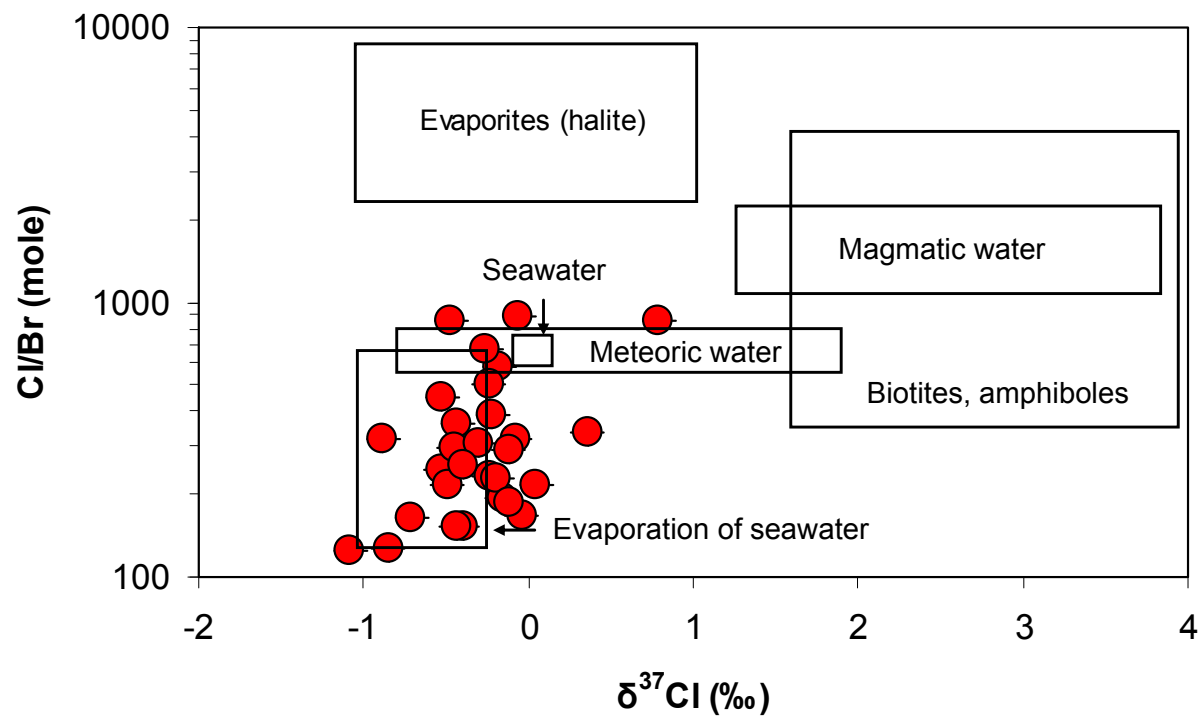


P-Patch deposit (East Athabasca)

## I: Origin of the brines

Technique: Crush-leach of quartz / dolomite veins

Na-Ca-brines: 6 mol/l chlorine



**Cl was concentrated in the brines by the subaerial evaporation of seawater**

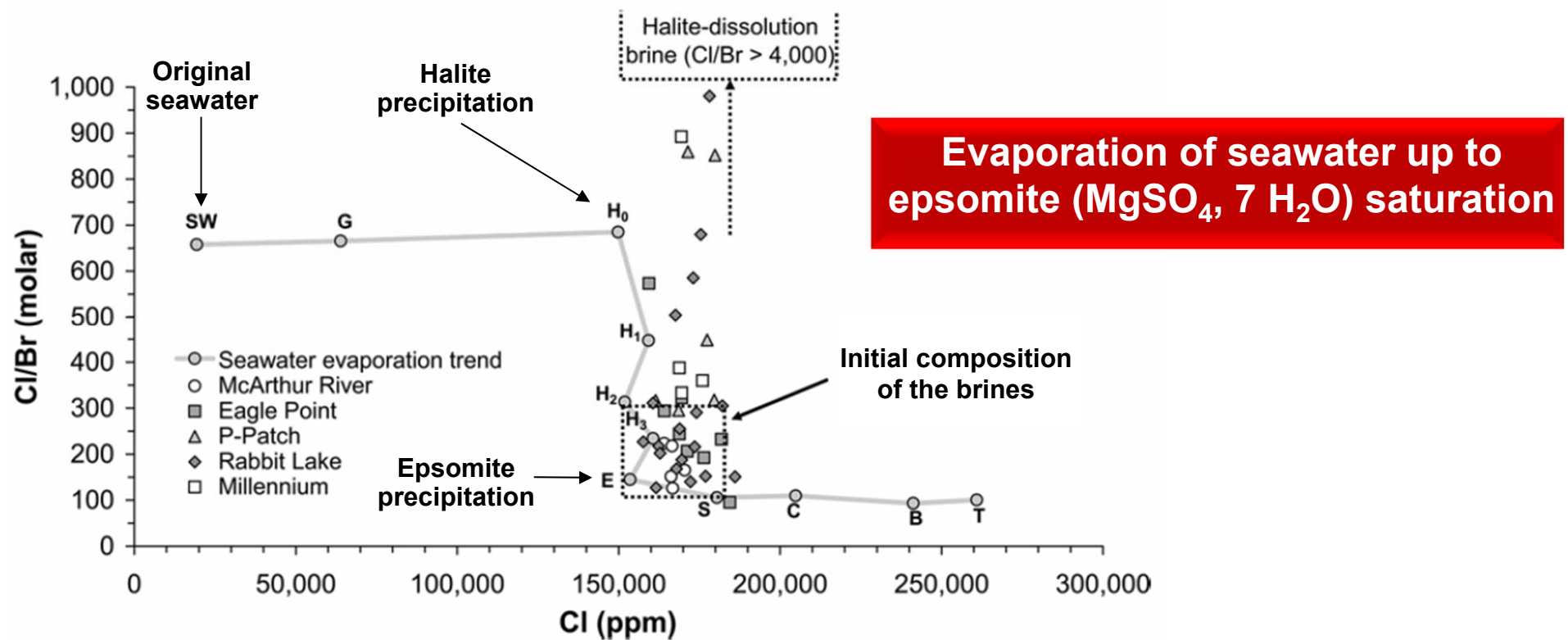
Richard et al. (2011), *Geochimica and Cosmochimica Acta*, 75, 2792-2810



# I: Origin of the brines

Technique: Crush-leach of quartz / dolomite veins

## Na-Ca-brines: 6 mol/l chlorine

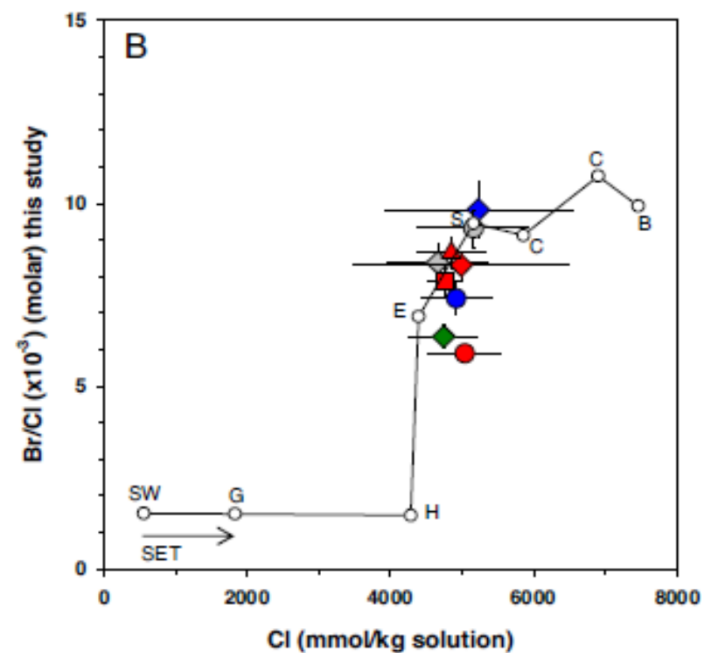
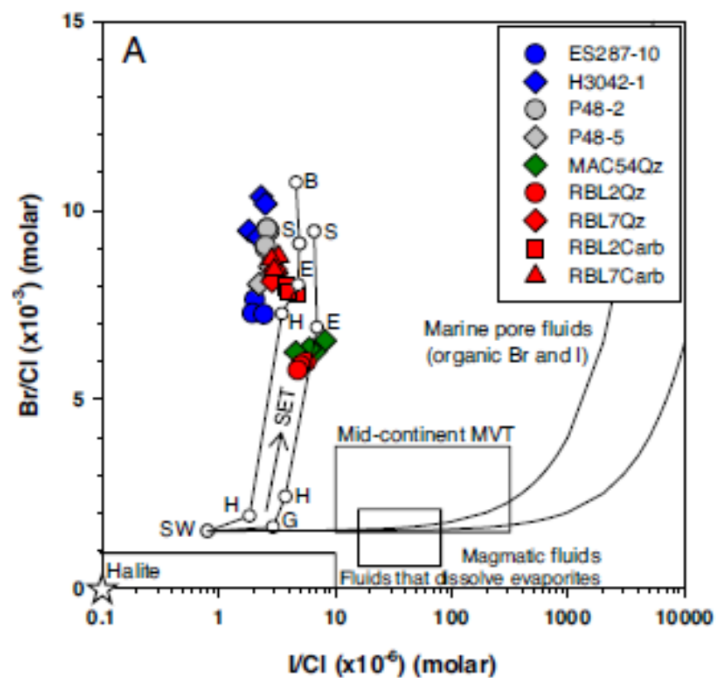


Richard et al. (2011), *Geochimica and Cosmochimica Acta*, 75, 2792-2810

## I: Origin of the brines

Technique: Halogens + noble gases of quartz / dolomite veins

Na-Ca-brines: 6 mol/l chlorine



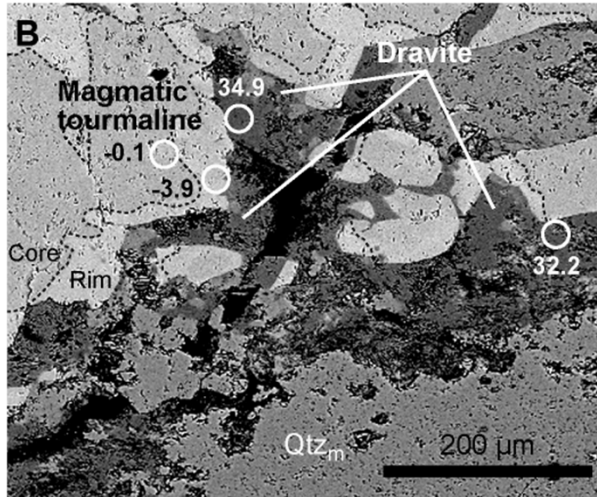
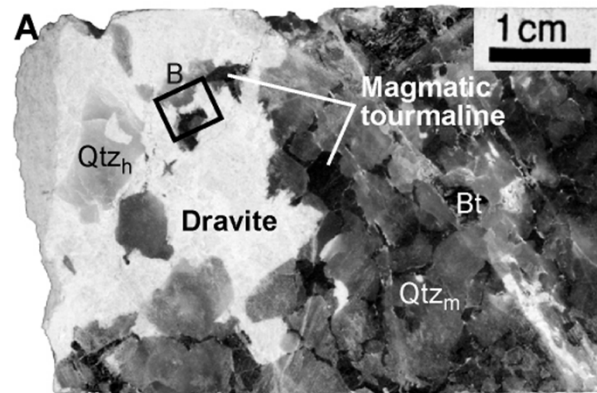
Richard et al. (2014), *Precambrian Research*, 247, 110-125

Evaporation of seawater up to epsomite (MgSO<sub>4</sub>, 7 H<sub>2</sub>O) saturation

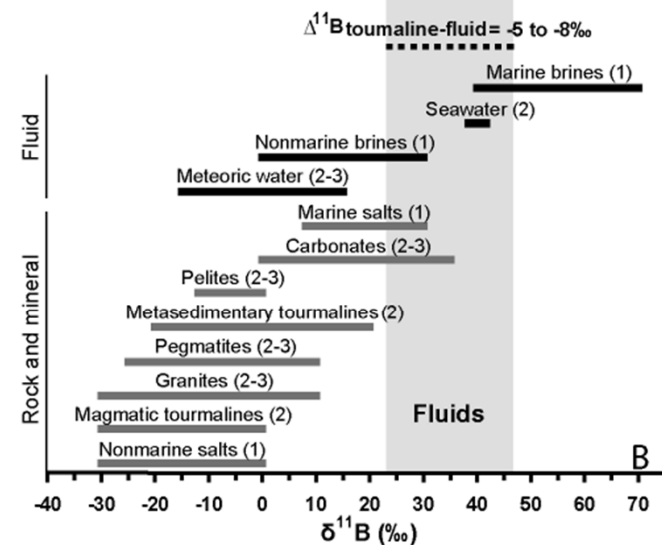
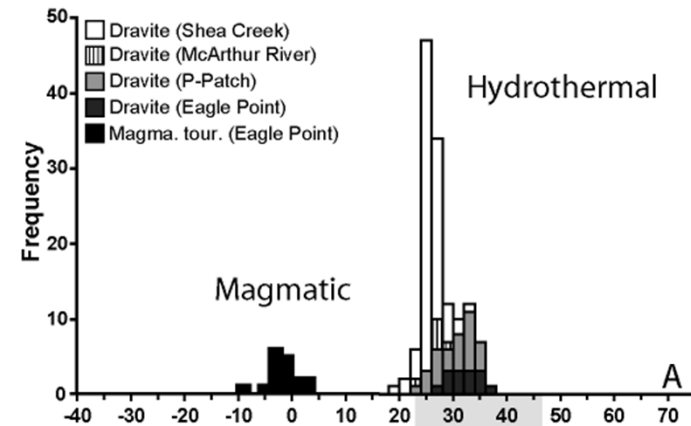
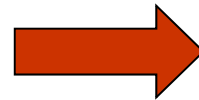
# I: Origin of the brines

## Technique: $\delta^{11}\text{B}$ of dravite (Mg tourmaline) by SIMS

Mercadier et al. (2012), *Geology*, 40, 231-234



○:  $\delta^{11}\text{B}$  (‰) of tourmaline

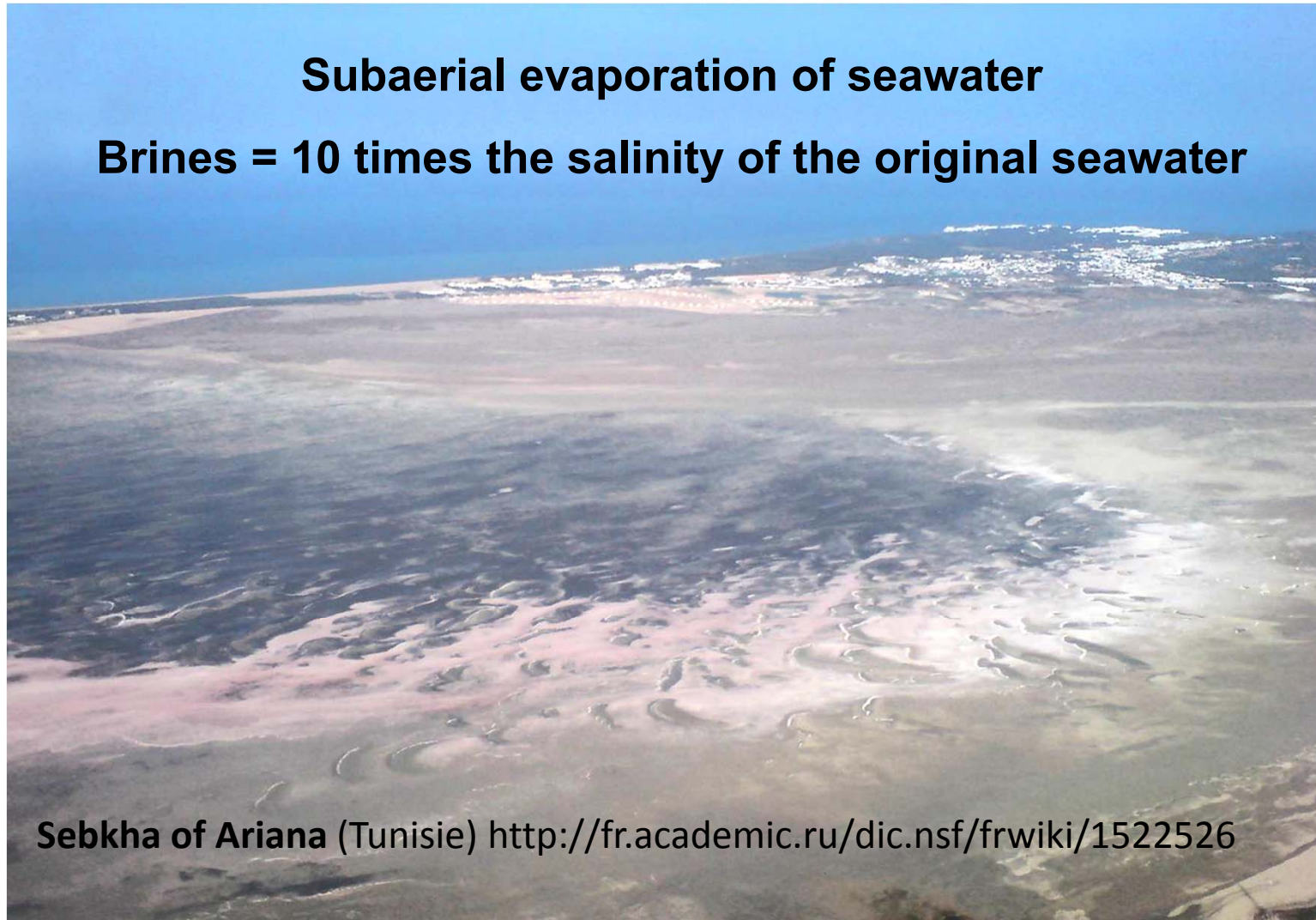


**B has an isotopic signature typical of brines from evaporation of seawater**  
**Majority of B + Mg in deposits from brines**

## I: Origin of the brines

**Subaerial evaporation of seawater**

**Brines = 10 times the salinity of the original seawater**

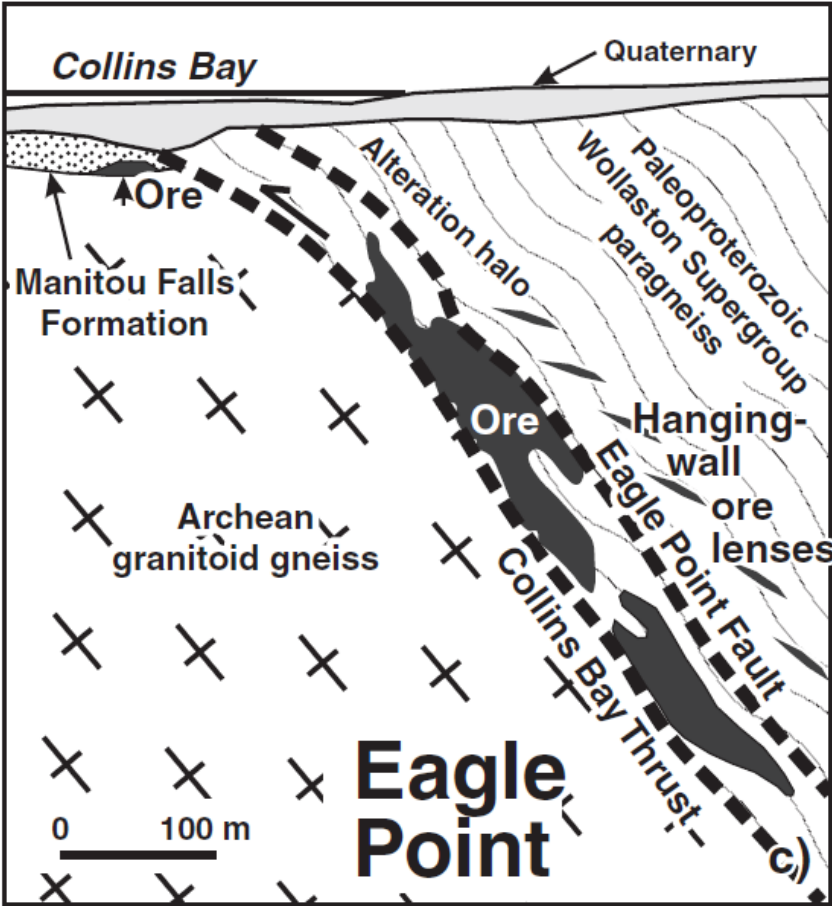


**Sebkhha of Ariana (Tunisie) <http://fr.academic.ru/dic.nsf/frwiki/1522526>**

## II: brine percolation in the basement and alteration

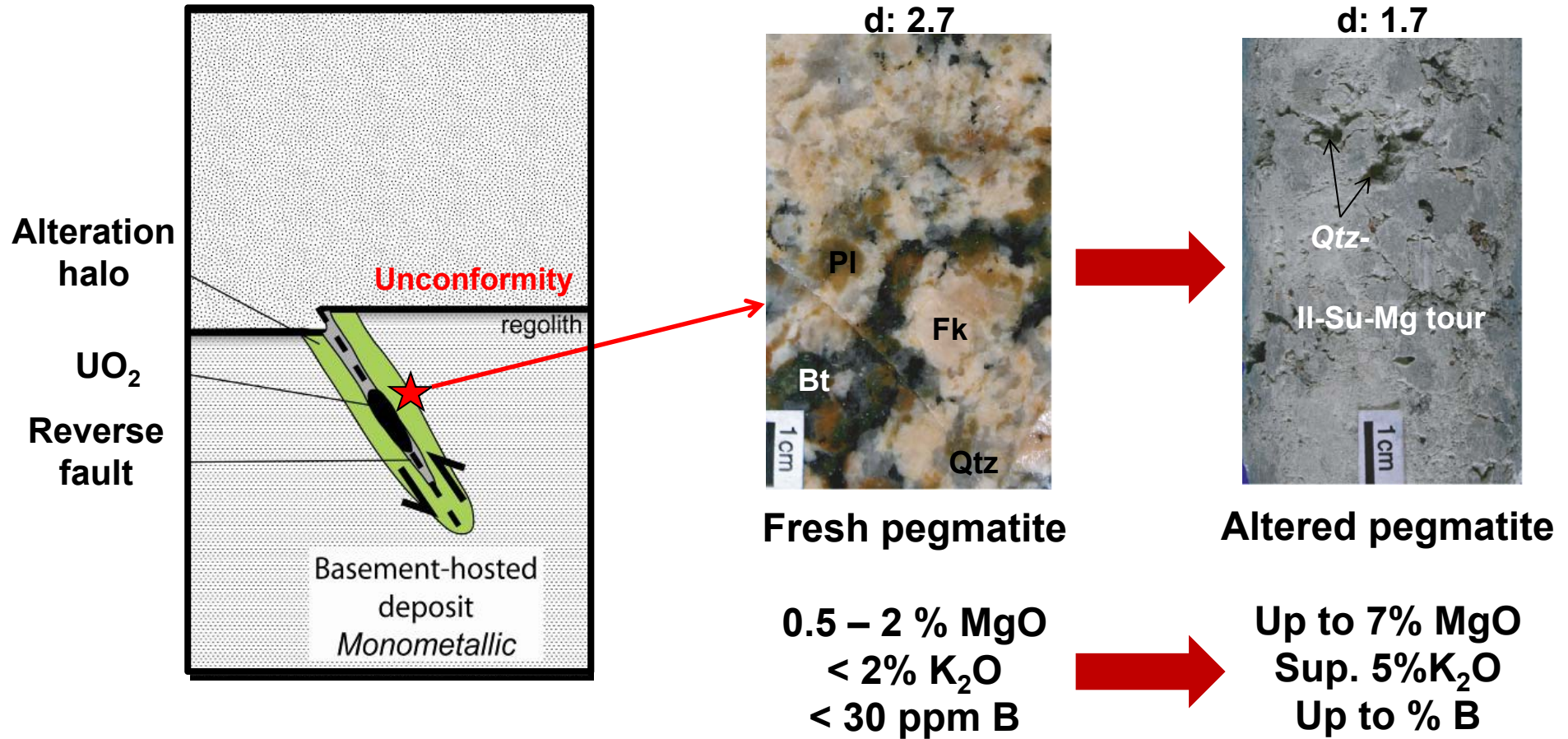
Unconformity

Basement



Jefferson et al. (2007)

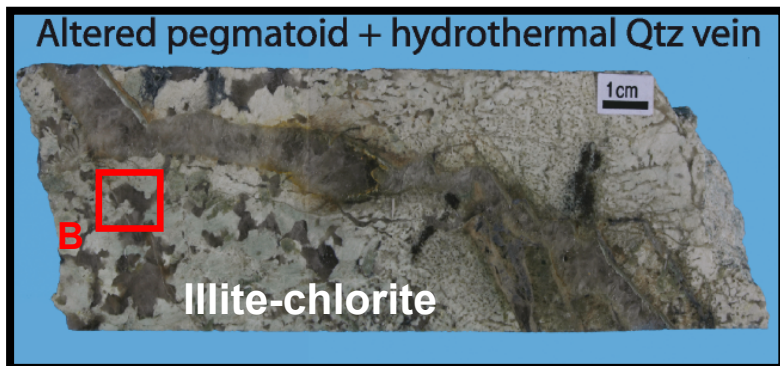
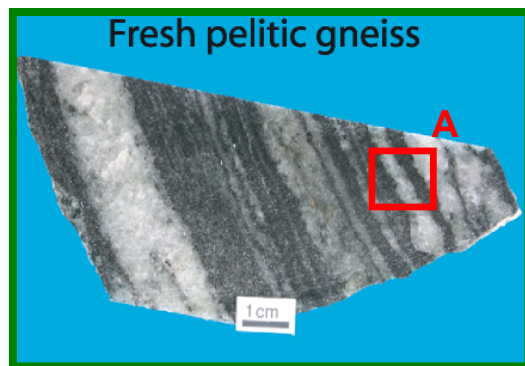
## II: brine percolation in the basement and alteration



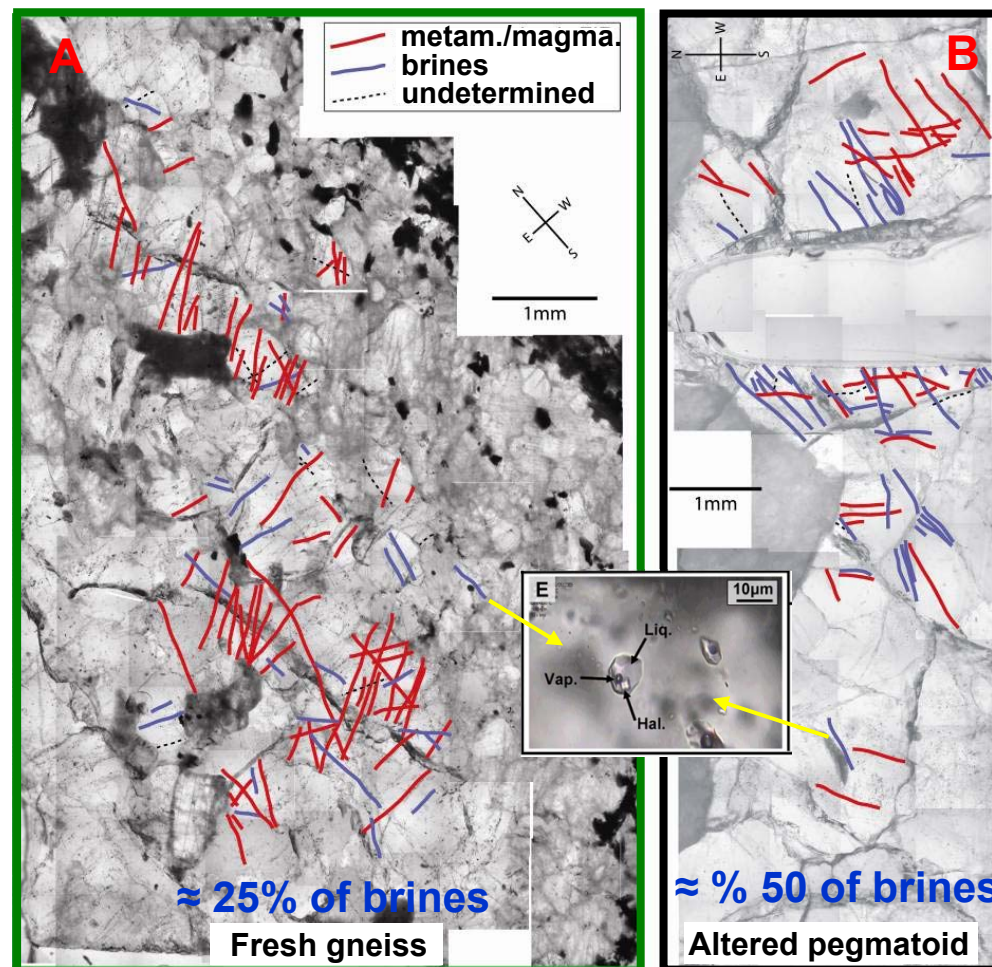
**How the brines percolated in the basement and developed the alteration and mineralization?**

## II: brine percolation in the basement

Technique: Fluid Inclusion Plane (FIP)  
nature and orientation in basement rocks

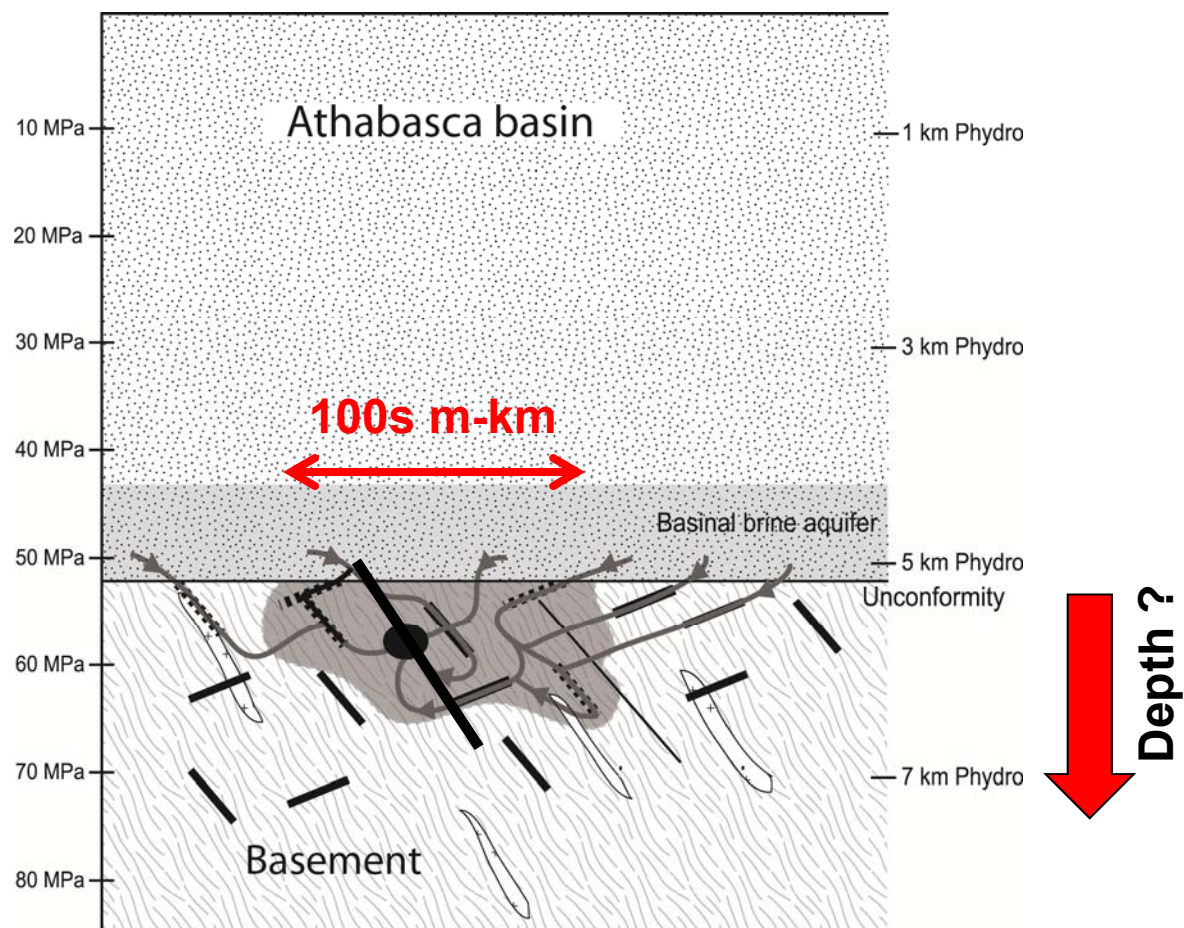


Basement rocks from P-Patch deposit



Mercadier et al. (2010), *Lithos*, 115, 121-136

## II: brine percolation in the basement



**Extensive brine percolation in the basement (400 m) via dense  $\mu$ fracturation networks during tectonic reactivation**



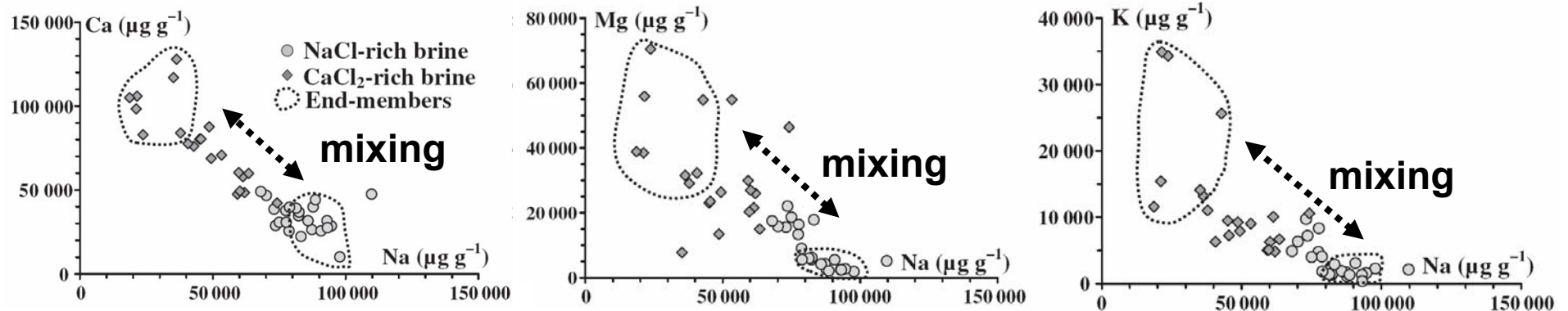
### III: brine modification via brine/basement interactions

Technique: LA-ICP-MS on individual fluid inclusion from hydrothermal quartz veins

Brine/basement rock interactions : major chemical and isotopic changes of the initial brines



2 brines: Cl-Na-Ca-Mg-K (NaCl-rich) and Cl-Ca-Mg-Na-K (CaCl<sub>2</sub>-rich) brines



Richard et al. (2010) , *Terra Nova*, 22, 303-308

Anti-correlation Na vs Ca-Mg-K-Sr-Ba

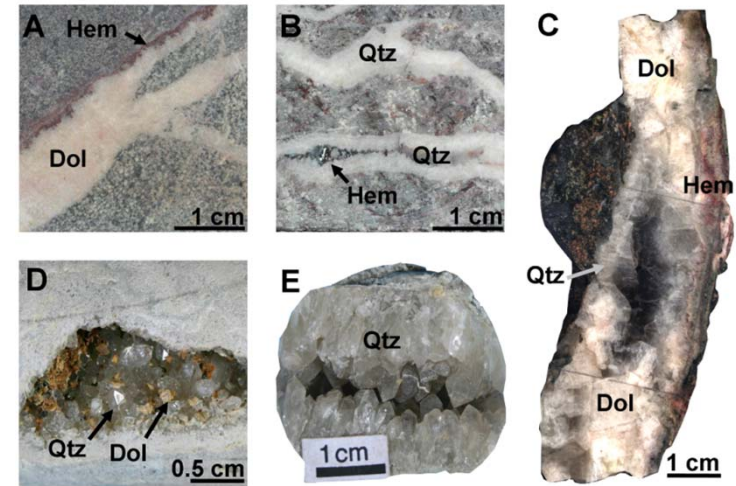
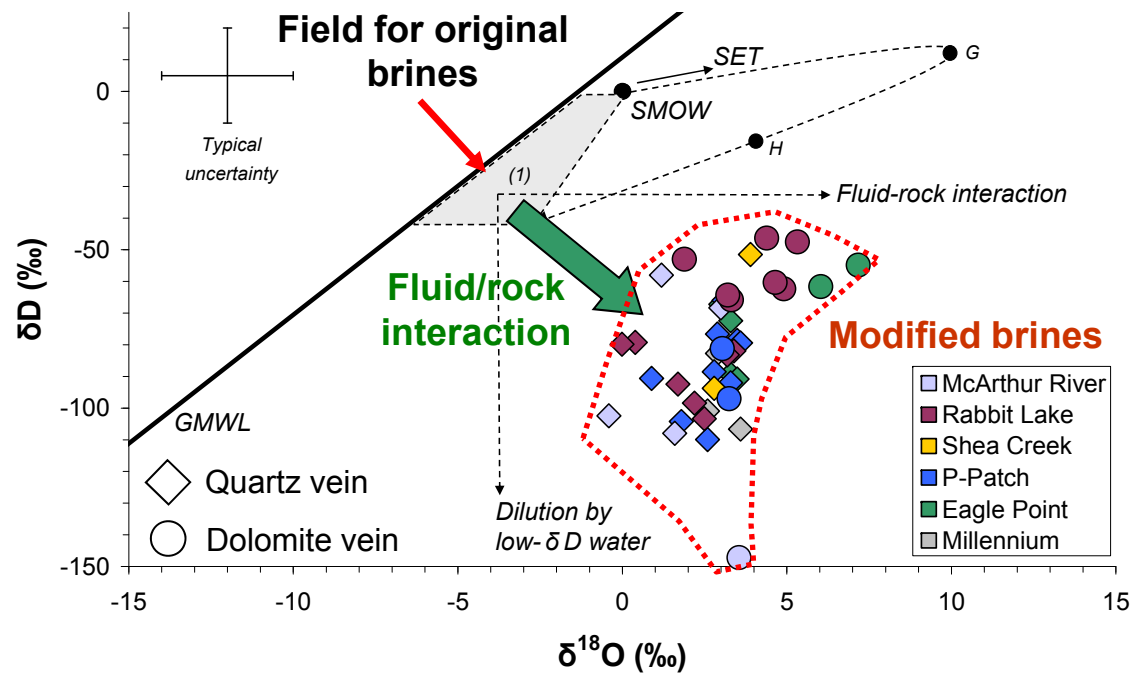
NaCl-rich: Mg + K depleted vs evaporated seawater : alteration halo

Ca enrichment: albitization?

### III: brine modification via brine/basement interactions

Technique: O and H isotope on quartz and dolomite veins

$T_{\text{fluid}}: 150 \pm 30^\circ\text{C}$  ( $\mu$ thermometry)



Quartz – dolomite veins

Richard et al. (2013), *Geochimica and Cosmochimica Acta*, 113, 38-59

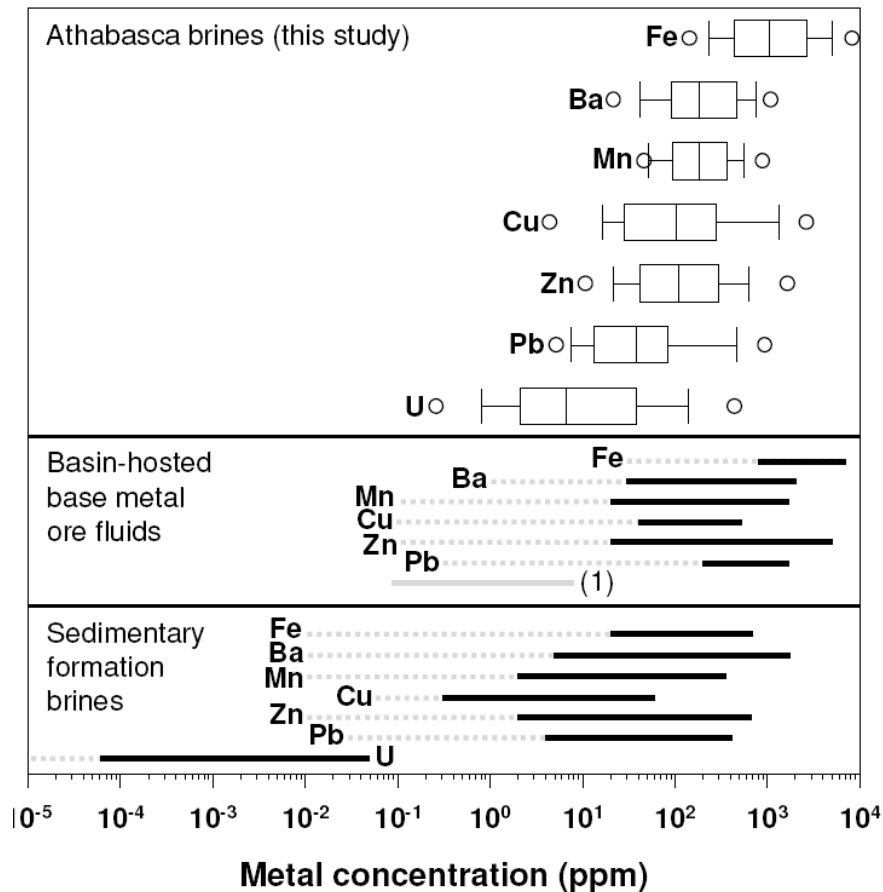
$\delta^{18}\text{O} > 0 \text{ ‰}$  : protracted fluid/basement rock interaction at low fluid-rock ratio

$\delta\text{D} \searrow$  : low  $\delta\text{D}$  water (radiolysis – bitumen synthesis)

### III: brine modification via brine/basement interactions

Technique: LA-ICP-MS on individual fluid inclusion from hydrothermal quartz veins

Metal uptake in the modified brines: Pb, Mn, Zn, Fe, Cu, U, Li, Sr



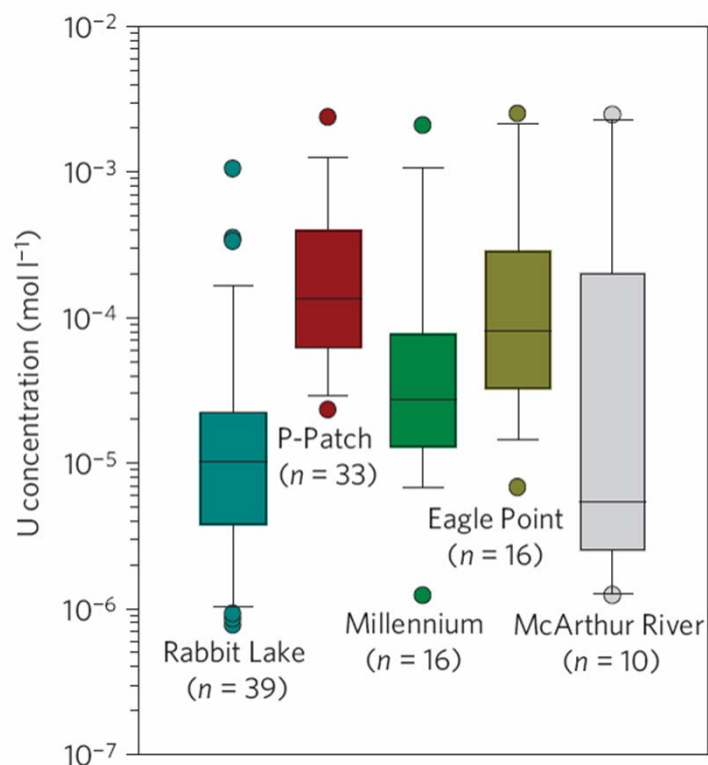
Strong metal enrichment compared to classical basinal brines and evaporated seawater

Correlation for the metals and similarity for deposits with basement metal source

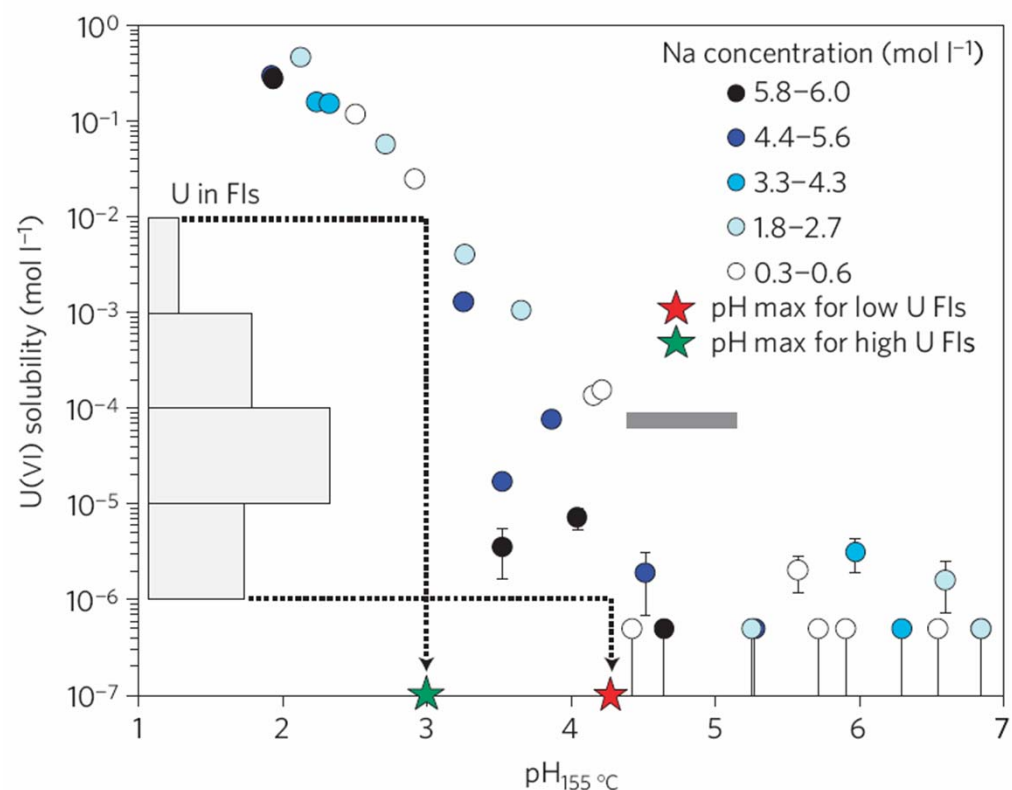
Copper belt, MVT, Zn-Pb...

## IV: Transport and deposition of U

Techniques: LA-ICP-MS on individual fluid inclusion + experimental work on U solubility in Na-Cl mixture analogous to ore forming brines

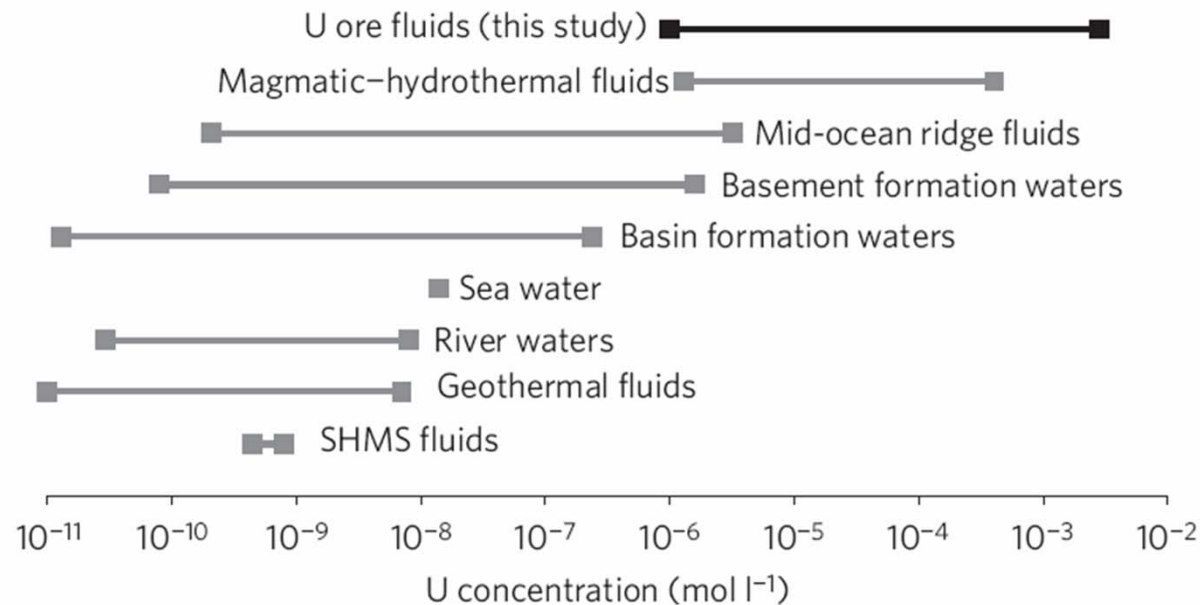


[U] in individual FI: 0.1 to 500 ppm



pH of ore forming brines: 2.5 - 4.5

## IV: Transport and deposition of U



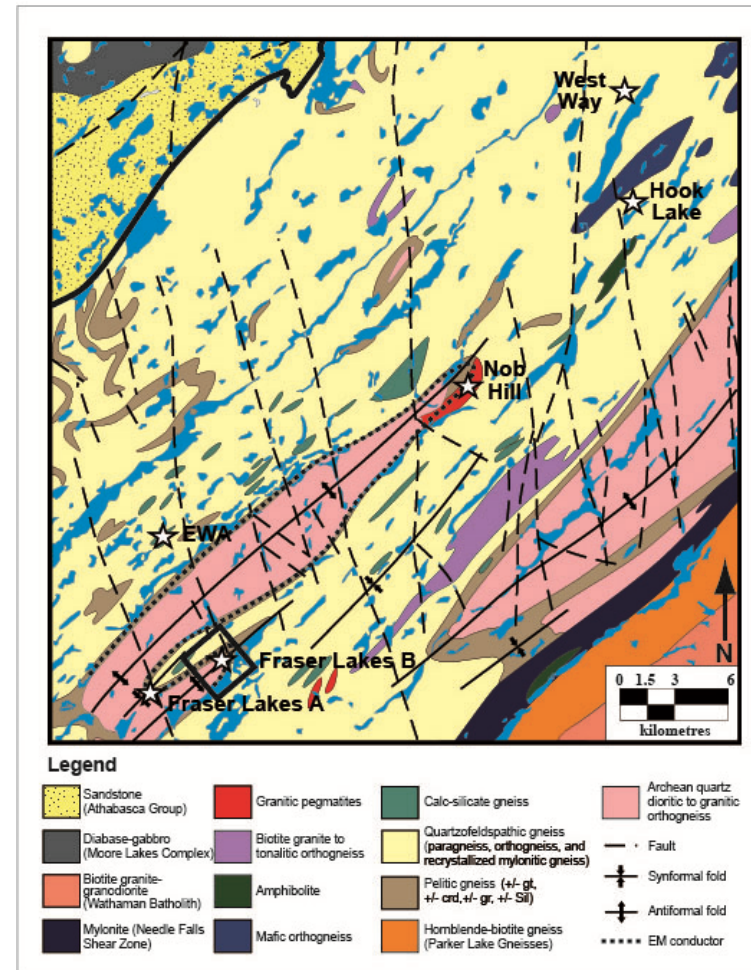
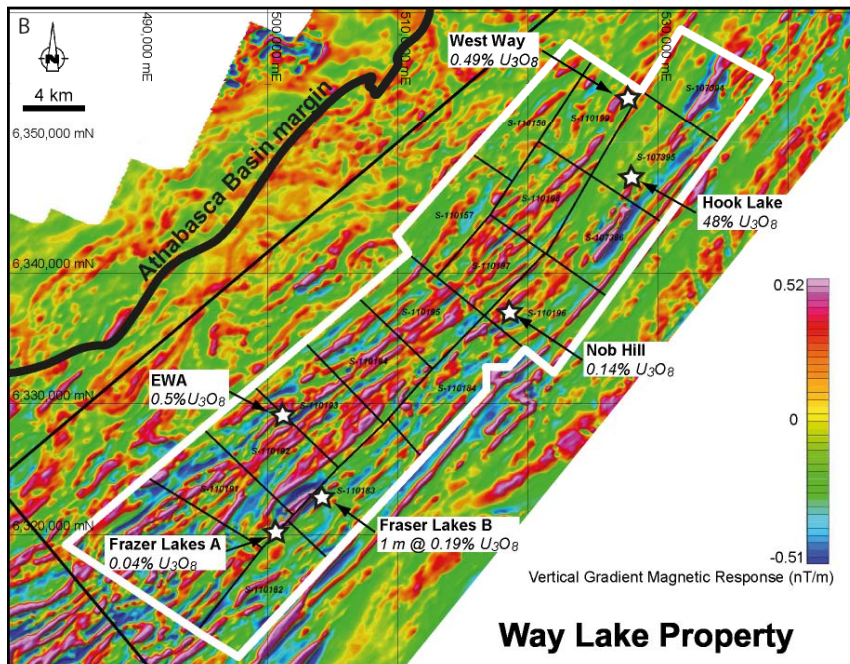
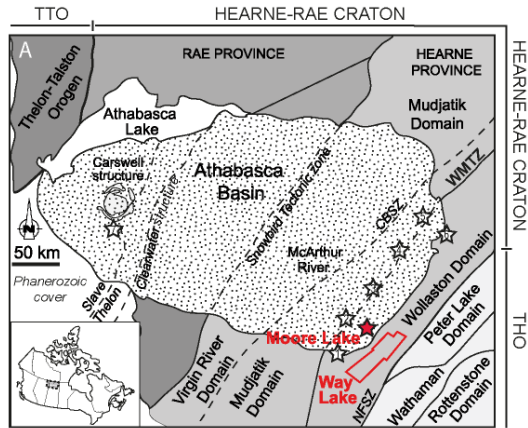
Richard et al. (2012) , *Nature Geoscience*, 5, 142-146

**Mineralizing brines from Athabasca = U richest crustal fluids thanks to:**  
**-oxidizing, acidic (pH: 2.5-4.5) and Cl-rich (6 mol/l) nature**  
**- U availability in the environment**

**High [U] : formation in a short period of time (0.1-1 Myr) as other world-class deposits of Pb-Zn and Au (Simmons et al., 2006; Wilkinson et al., 2009)**

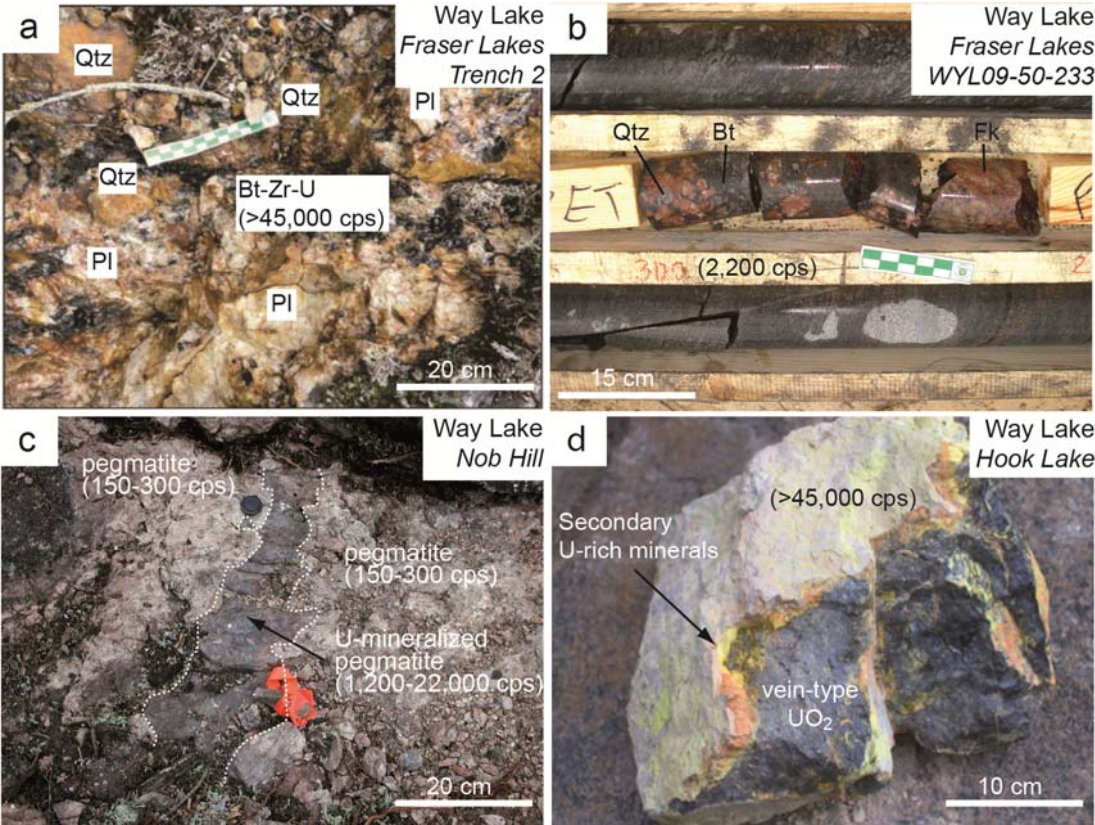
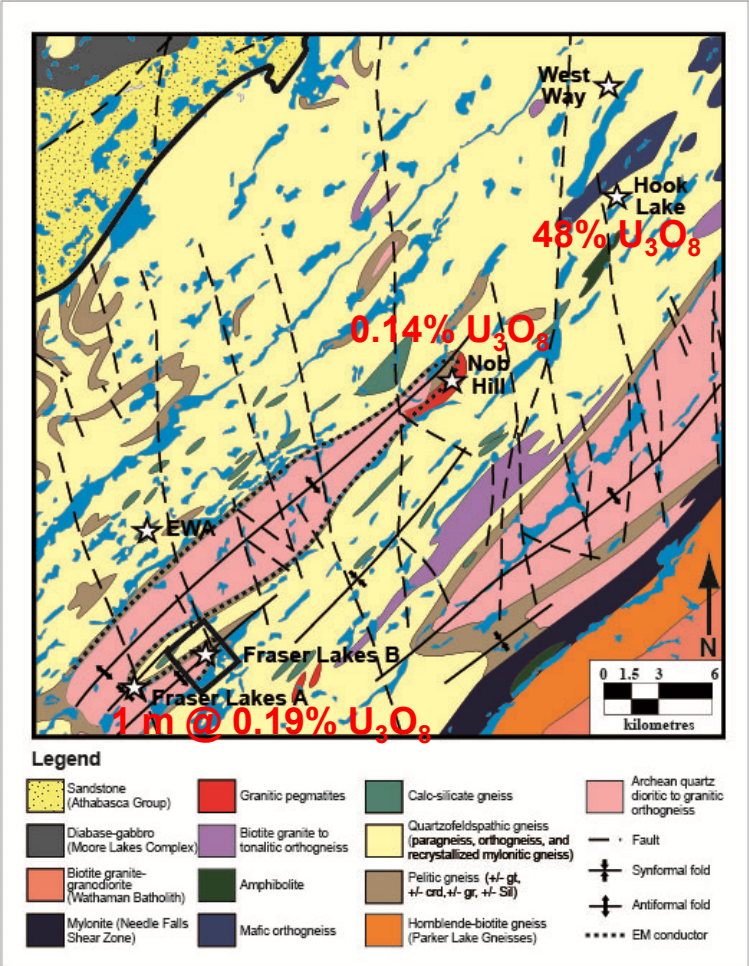
# V: Source(s) of uranium ?

## Basin vs. Basement, still in discussion



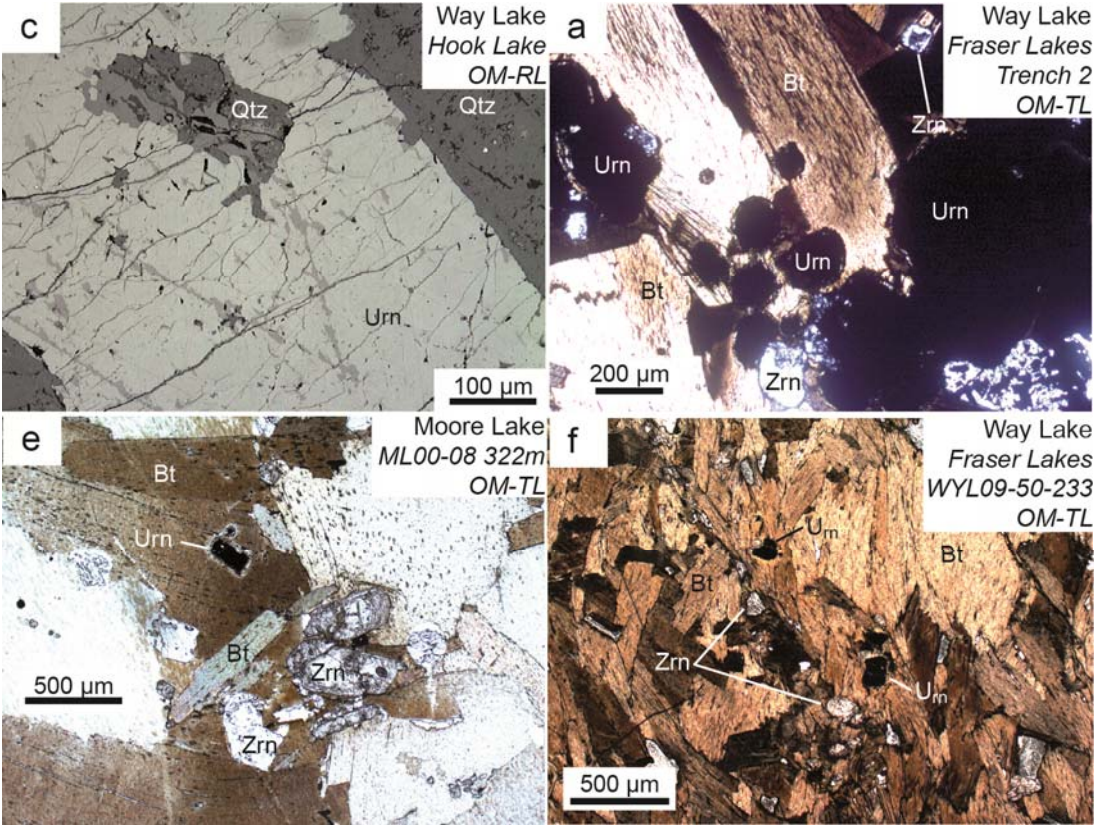
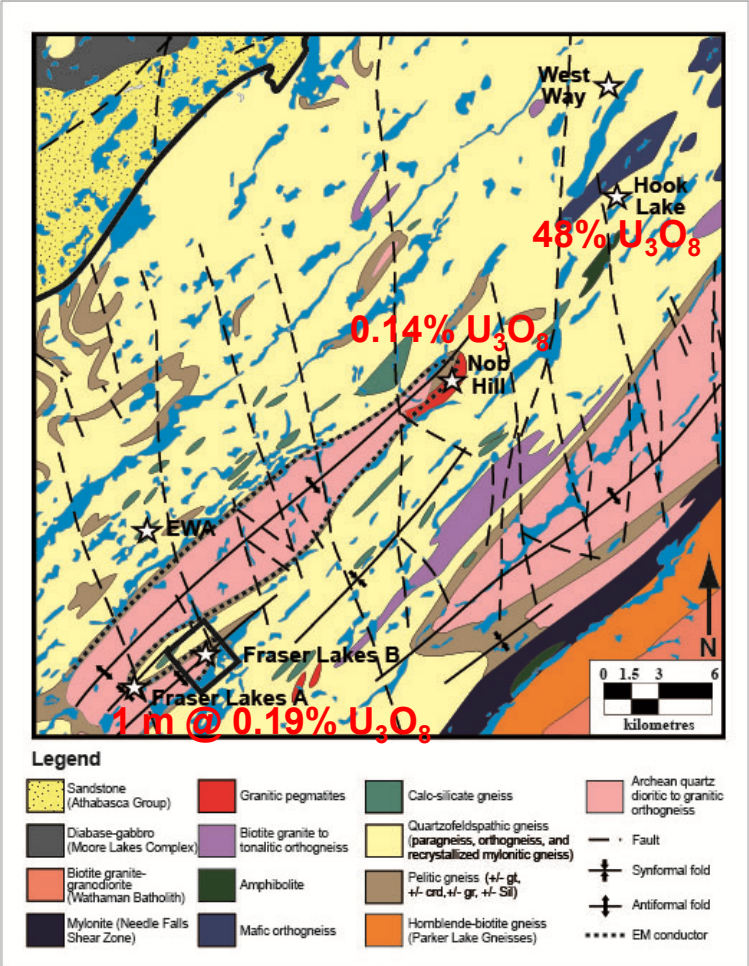
# V: Source(s) of uranium ?

# U occurrences in the basement



# V: Source(s) of uranium ?

## U occurrences in the basement

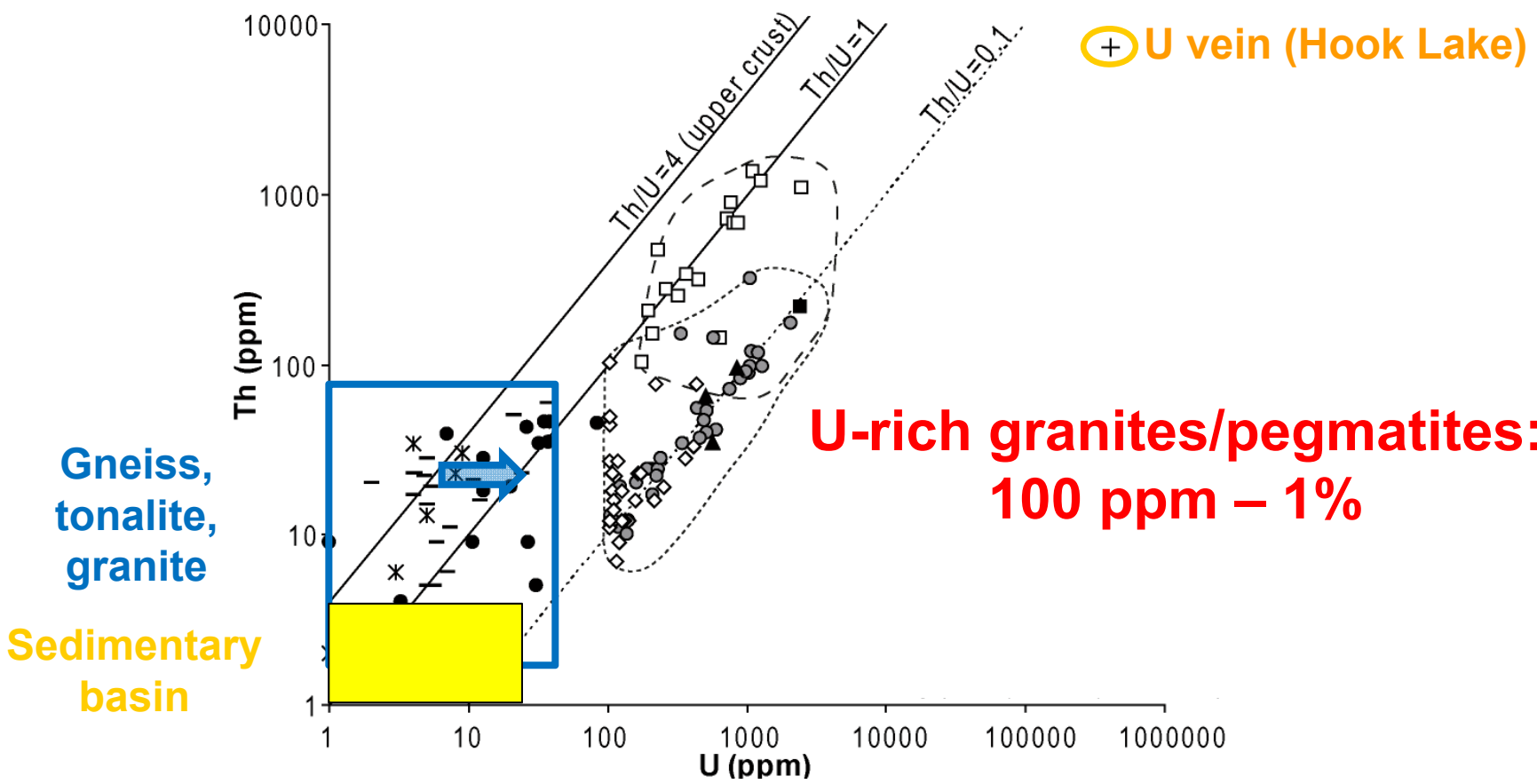


**Age: 1805-1750 Ma = Hudsonian  
ante Athabasca Basin**



# V: Source(s) of uranium ?

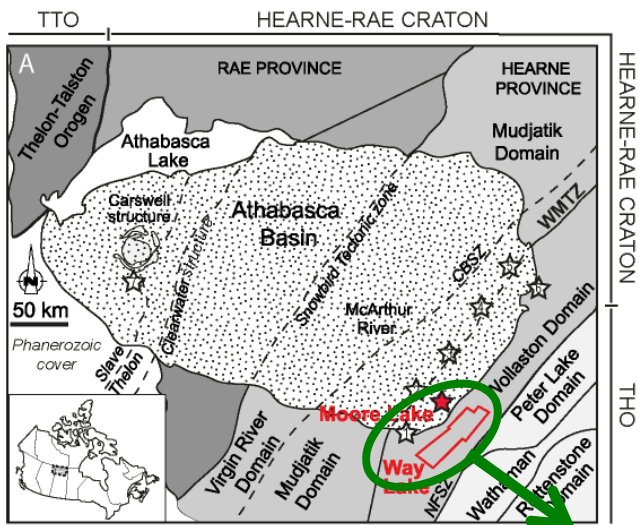
## U occurrences in the basement



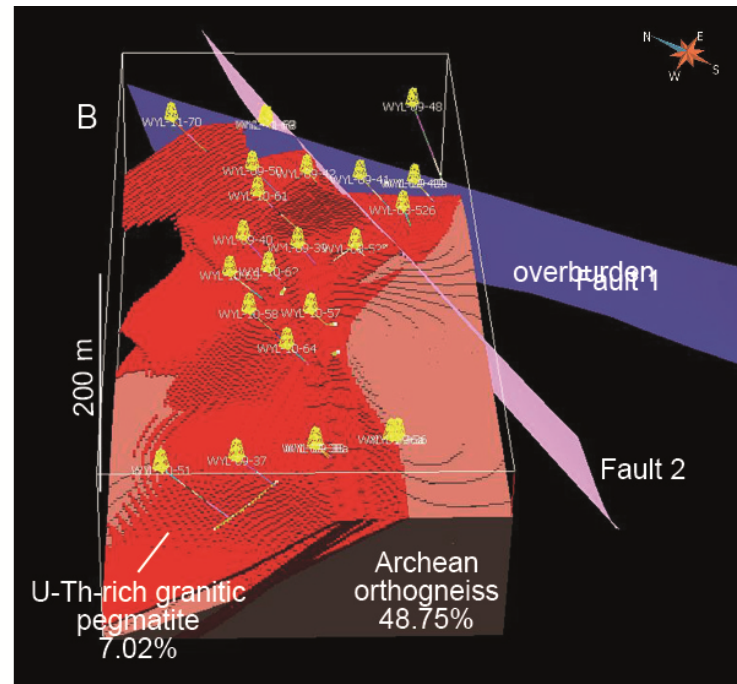
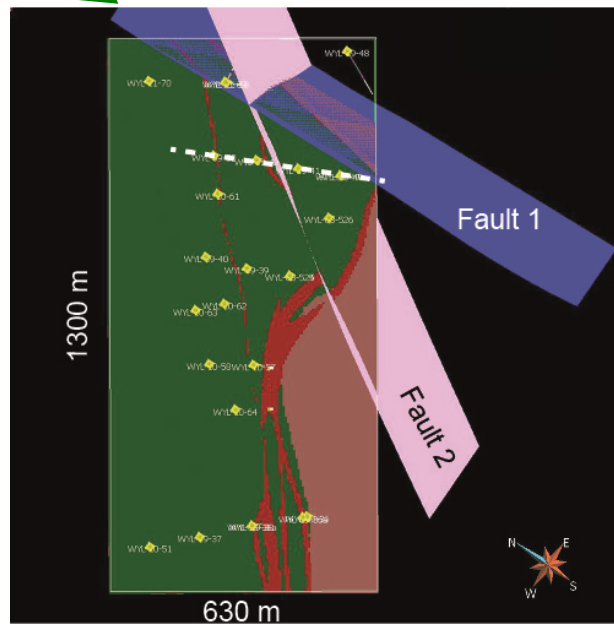
# V: Source(s) of uranium ?

**U-rich granites/pegmatites:  
An important lithology of the basement (7%)**

**3D modelization of the basement  
Gocad software**



Mercadier et al. (2013) , *Economic Geology*, 108, 913-933

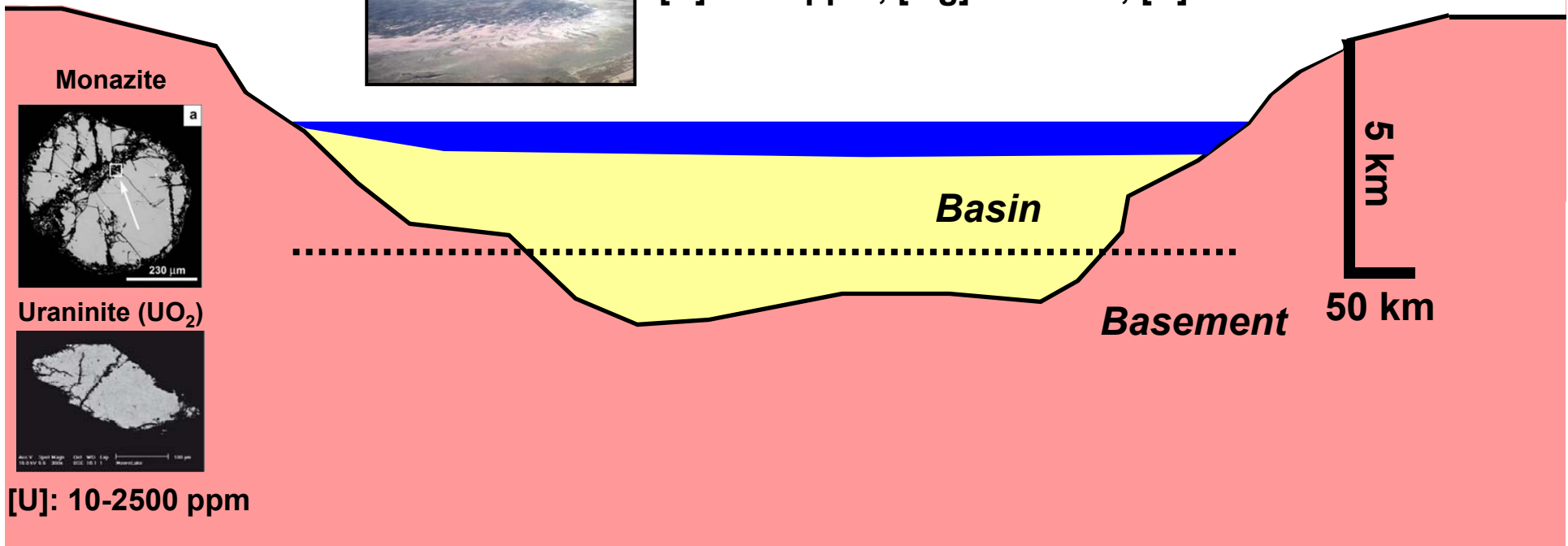


# A new model for unconformity-related U deposits

## Brine factory



Subaerial evaporation of seawater  
[U] < 0.2 ppm, [Mg] > 2 mol/l, [B] > 10<sup>-2</sup>mol/l

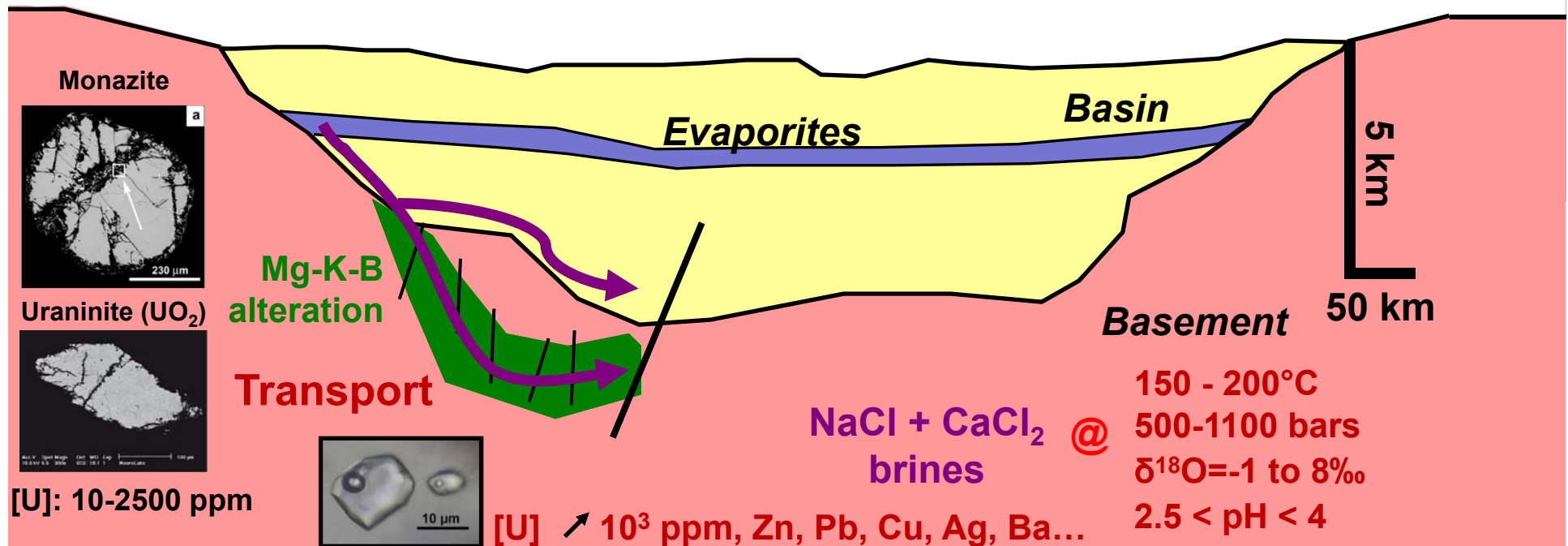


## U Source

# A new model for unconformity-related U deposits

**Brine percolation in the basement thanks to  $\mu$ fractures + faults during tectonic reactivation**

Chemical + isotopic modifications of the original brines to 2 end-members

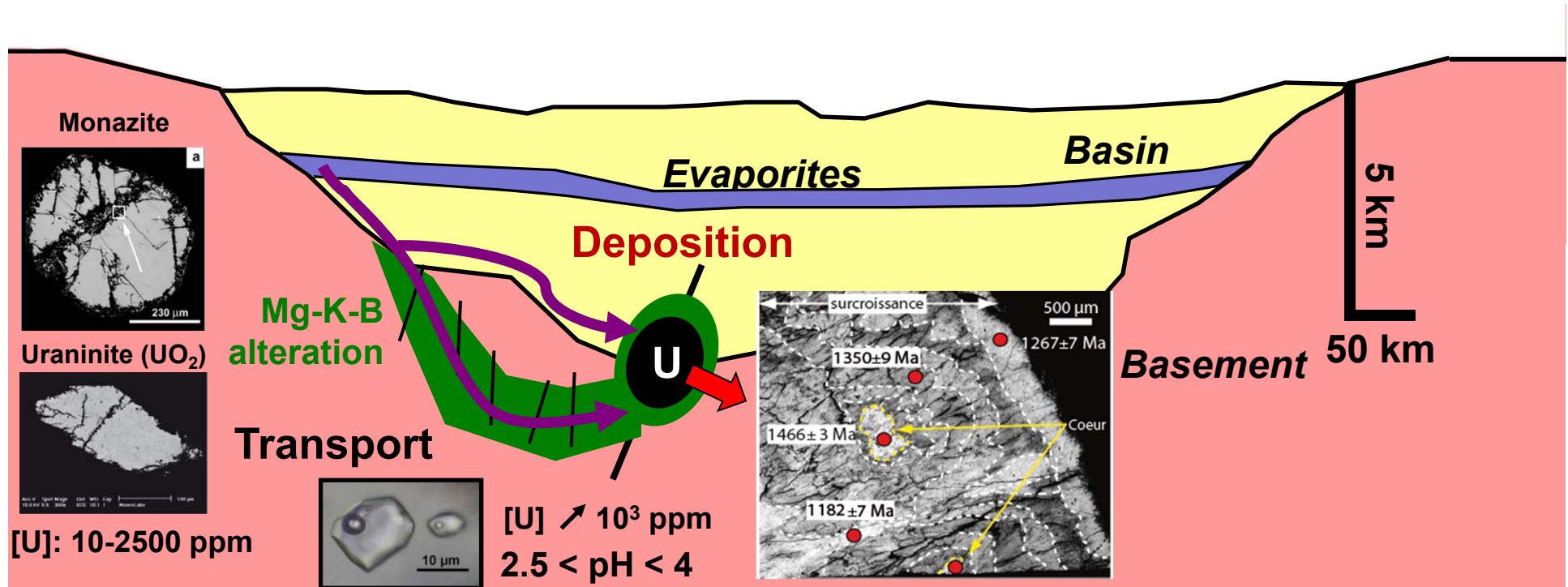


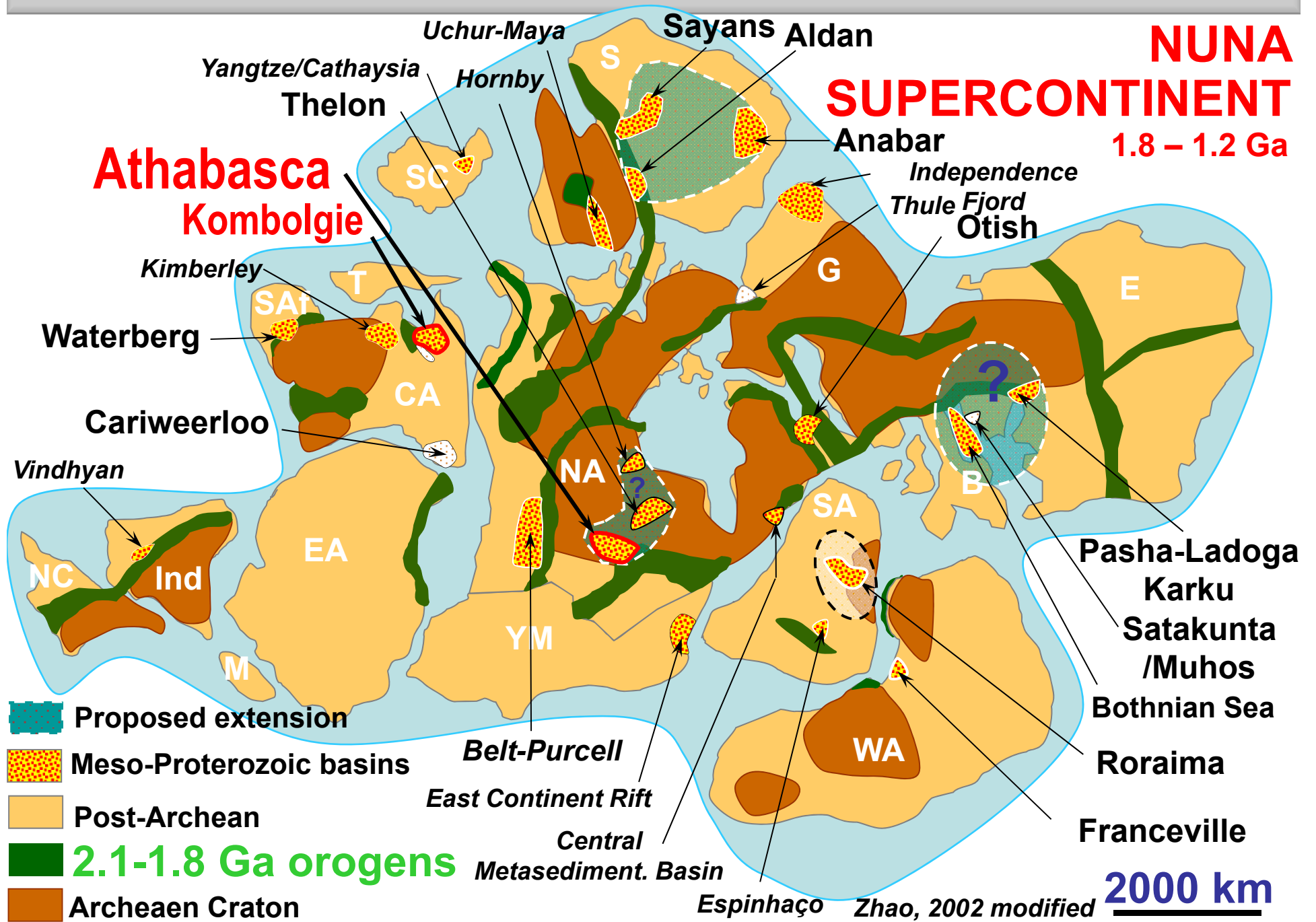
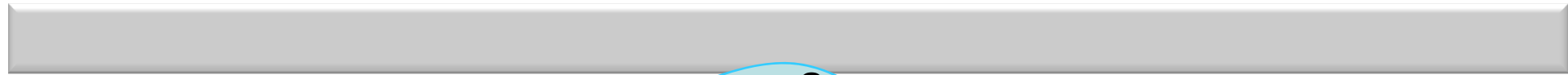
**metal uptake**

**Dissolution/alteration of accessory minerals + silicates due to low pH, oxydation and Cl availability**

# A new model for unconformity-related U deposits

**U deposition:** Brine mixing?  
Gas-brine mixing?  
Brine-mineral?





**SGA**  
**50<sup>TH</sup>**  
ANNIVERSARY MEETING

**NANCY FRANCE** | **24-27 AUGUST 2015**

**WELCOME BACK IN NANCY**  
FOR THE 13TH SGA BIENNIAL MEETING

**"MINERAL RESOURCES IN A SUSTAINABLE WORLD"**

[HTTP://SGA2015.UNIV-LORRAINE.FR](http://SGA2015.UNIV-LORRAINE.FR)

UNIVERSITÉ DE LORRAINE | RWTH AACHEN UNIVERSITY | KATHOLIEKE UNIVERSITEIT LEUVEN | Université de Liège | CNRS

## SESSION 6. Uranium symposium - in honour of Michel Cuney

This symposium welcomes papers dealing with uranium metallogeny and geochemistry. This session will illustrate the extreme diversity of uranium deposits and contributions could cover a large scale of geological environments (igneous, hydrothermal, sedimentary or superficial) and geological processes. Presentations of non-conventional deposits are welcomed. This session will be in honour of Michel Cuney who spent his entire scientific career to decipher the uranium geological cycle.

