



# Modern Mass Spec tools for food profiling to ensure authenticity and quality

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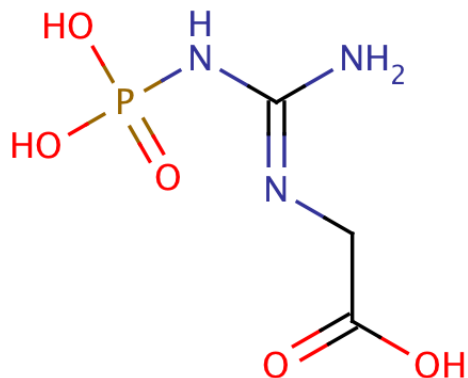
# Chemometric Multivariate Analysis

Goal: Detect observed phenotype differences at the chemical level

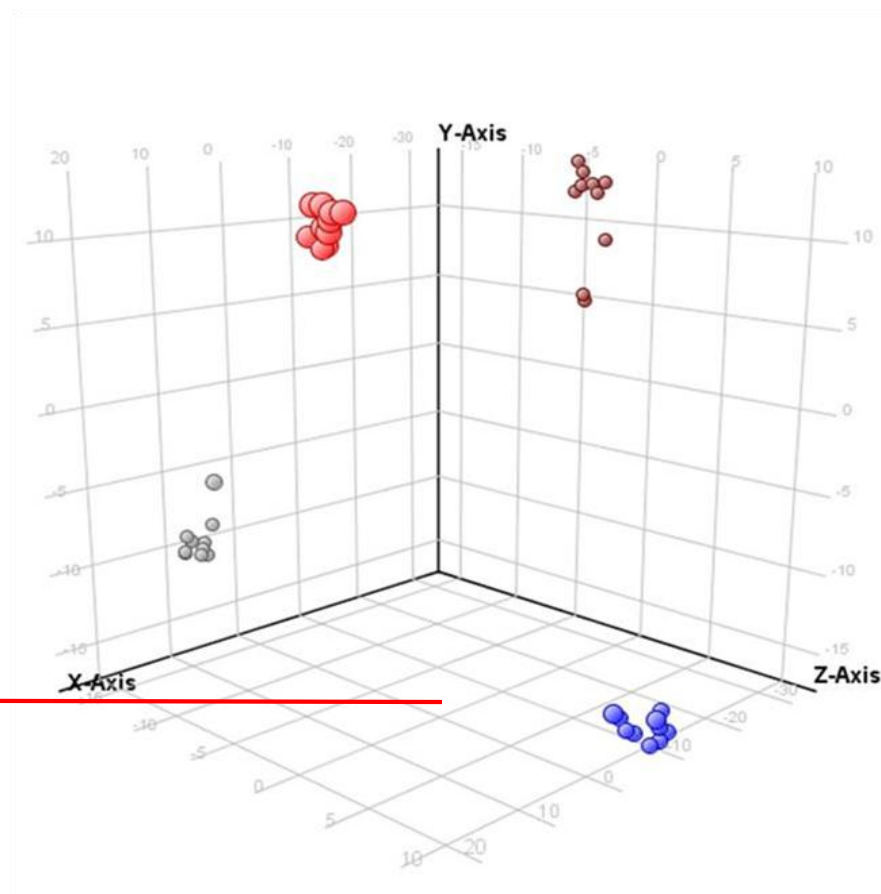
Multivariate analysis is a means to detect differences between phenotypes at the chemical level

Multivariate analysis can rank metabolite importance to phenotype

Does not explain underlying biology



Phosphoguanidinoacetate



GCMS

Separate &  
Detect



GC-QTOF/MSD  
GC-QQQ

Feature  
Finding  
Quantitate



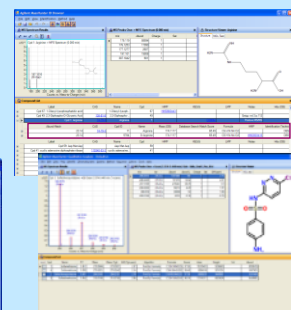
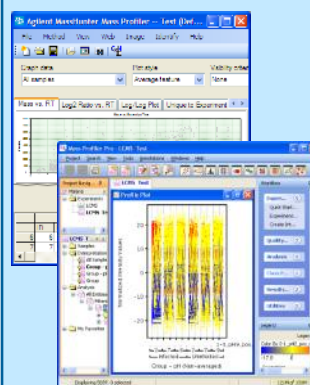
**MassHunter Qual**  
*AMDIS or Find by  
chromatographic  
deconvolution*

Alignment &  
Statistics

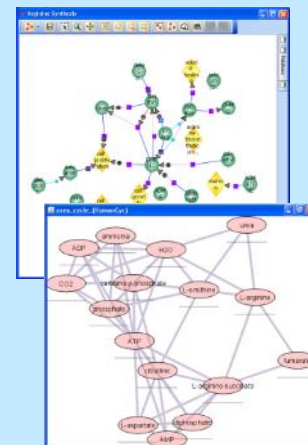
Identify

Pathways

**Mass Profiler  
(MP)**



**ID Browser**



**Pathway module  
Cytoscape**

**Mass Profiler  
Professional  
(MPP)**

LCMS



LC-TOF/QTOF  
LC-QQQ

**MassHunter Qual**  
*MFE,  
Find by Formula,  
Find by Ion*

LC/MS and GC/MS systems are already used to identify organic and inorganic contaminants in feed, food and other food stuff.



**Agilent Technologies**

# Chemometric – Un-Targeted or Targeted Analysis

## Un-Targeted - Looking for all metabolites and chemicals

- Data is acquired in full scan mode
- Metabolite tracking uses retention time and EI spectra
- Statistics used to find interesting metabolites (features)
- Features must be identified to make biological sense

## Targeted - Looking for known metabolites only

- Data is acquired in SIM (MS) or MRM (MS/MS) mode
- Absolute quantitation - Need external and internal standards
- Statistics used to confirm interesting metabolites



# Sample Analysis – GC/MS

## **GC/MS – Best for routine analysis**

### **Advantages**

- No ionization suppression
- Identification by EI searchable libraries

### **Disadvantages**

- Volatile analytes - derivatization required
- Molecular ion often missing



## **GC/MS/MS – Best for targeted**

### **Advantages**

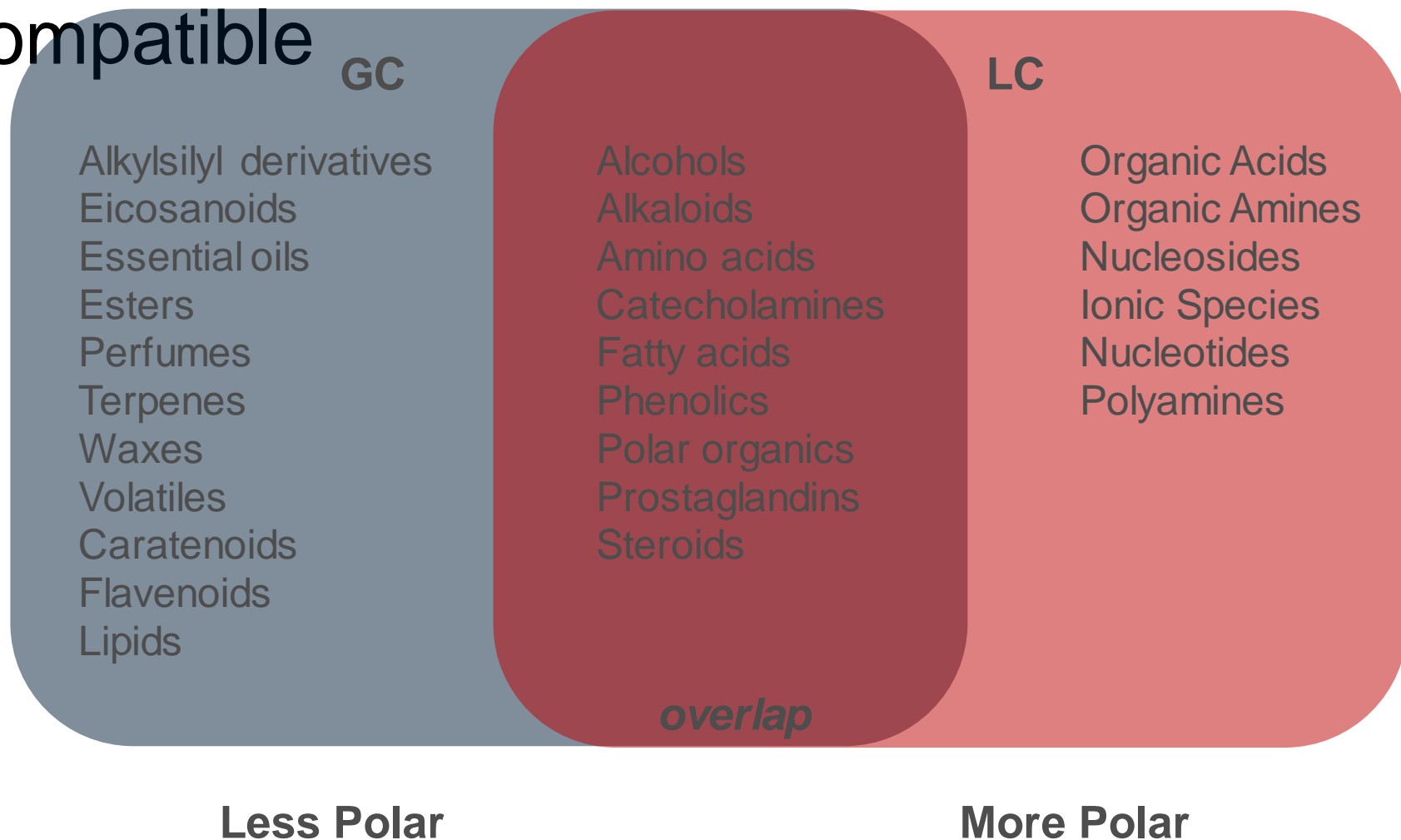
- Lower detection limits
- Greater selectivity

### **Disadvantages**

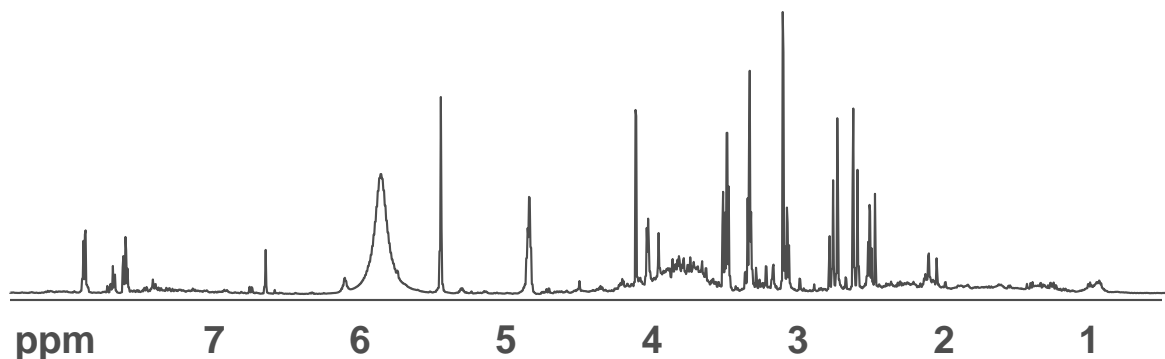
- Volatile analytes - derivatization required
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# Classes of chemicals and the analytical techniques with which they are most compatible

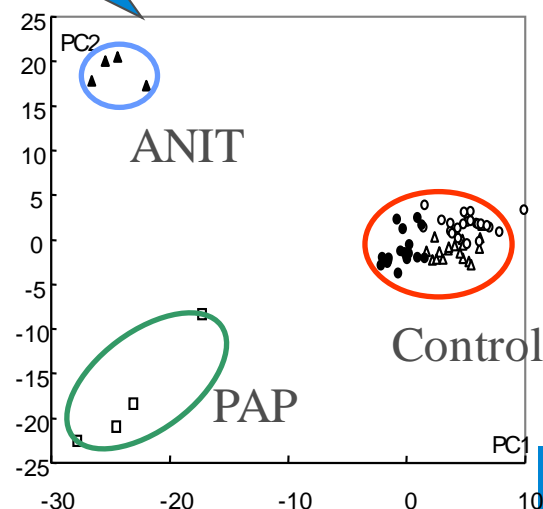
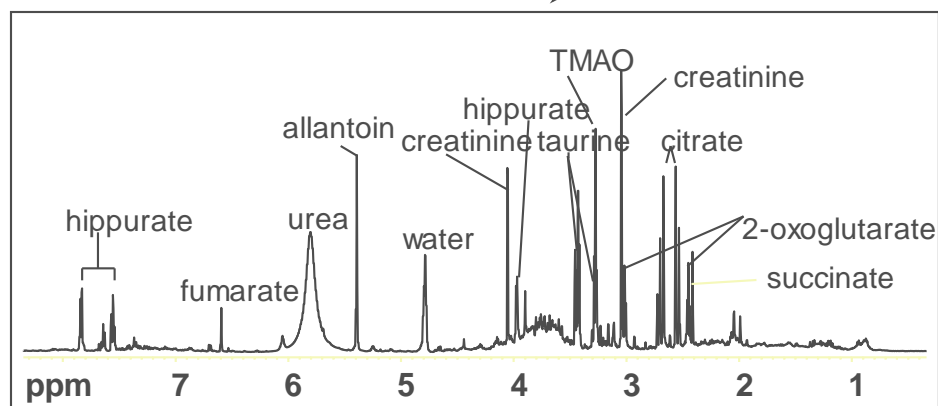


# 2 Routes to Chemometric



**Quantitative  
Methods**

**Chemometric (Pattern)  
Methods**





# Main Challenge





# **A CASE STUDY ON CHAMOMILE FLOWERS AND OILS**



# What is Chamomile?

**Chamomile: one name, different botanical species**



**German Chamomile**

*Matricaria chamomilla* L.

**Roman Chamomile**

*Chamaemelum nobile* (L.) All.

**Juhua**

*Chrysanthemum morifolium* Ramat.

**All chamomile species belong to the same *Asteraceae* (*Compositae*) family but different genera**



# Why Is Chamomile Important?

## Beneficial Properties

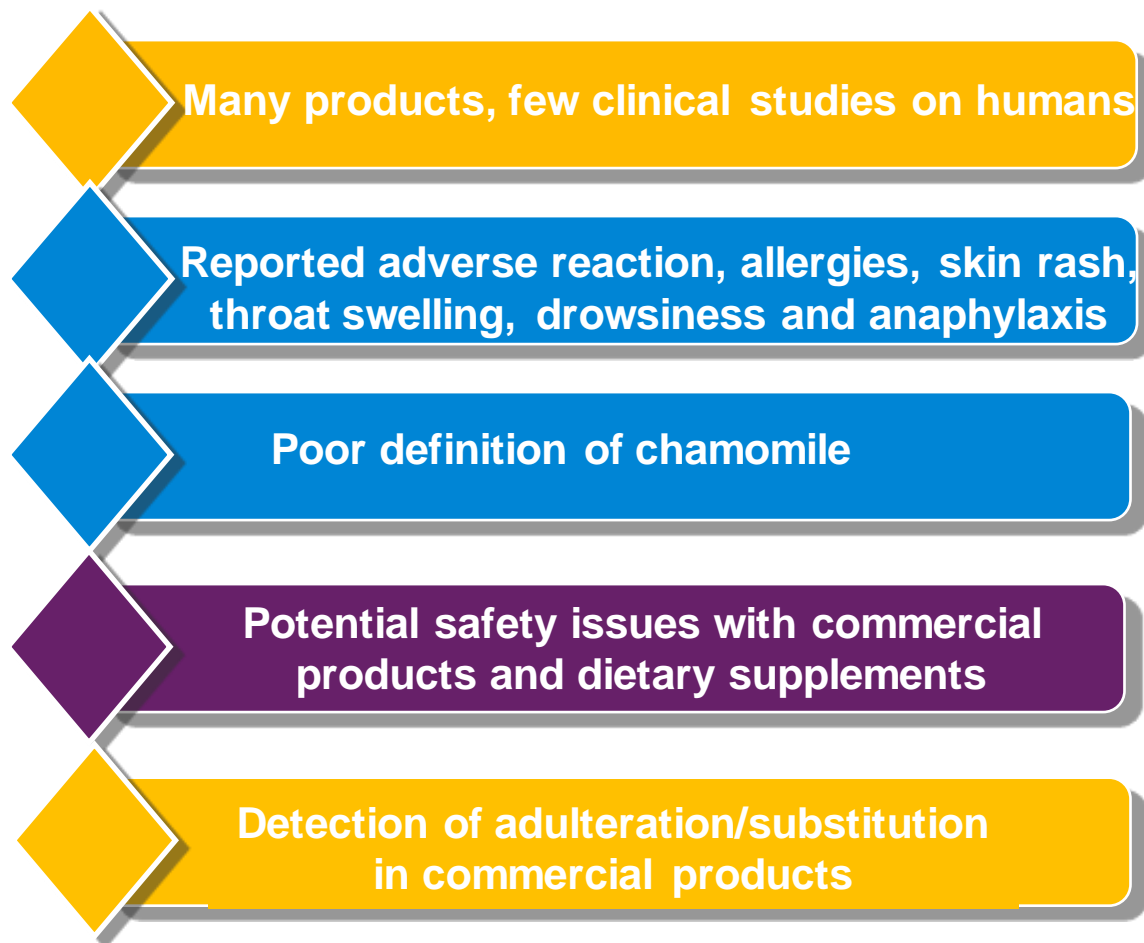
- Anti-inflammatory
- Hay fever
- Wound healing and burn relief
- Gastro-intestinal disorders
- Tooth ache, ear ache
- Ulcers
- Antibacterial
- Antifungal
- Infections
- Common cold
- Laryngitis
- Anxiety relief
- Sleep disorders
- Cardiovascular diseases
- Muscle spasm
- Rheumatic pain
- Arthritis

## Commercial Products

- Cosmetics
- Aromatherapy
- Teas
- Lotions
- Herbal Beer
- Gargles
- Shampoos



# Why Did We Study Chamomile?



# What Workflow Did We Use?

## Sample Preparation

Extract the volatile compounds by appropriate solvent

27 authenticated plants and 35 commercial products were extracted by hexane

11 essential oils were diluted in hexane

## GC/MS Analysis

Separate and detect compounds by the appropriate platform

## Peak Finding

Deconvolution  
Find and quantitate all compounds

## Statistical Analysis

Agilent MPP Software

Find meaningful differences in sample sets

Construct a Sample Class Prediction (SCP) model

## Prediction

Sample Class Prediction Model

Classify commercial products and oils

Identify markers for sub-group of chamomile



# How Did We Study Chamomile?

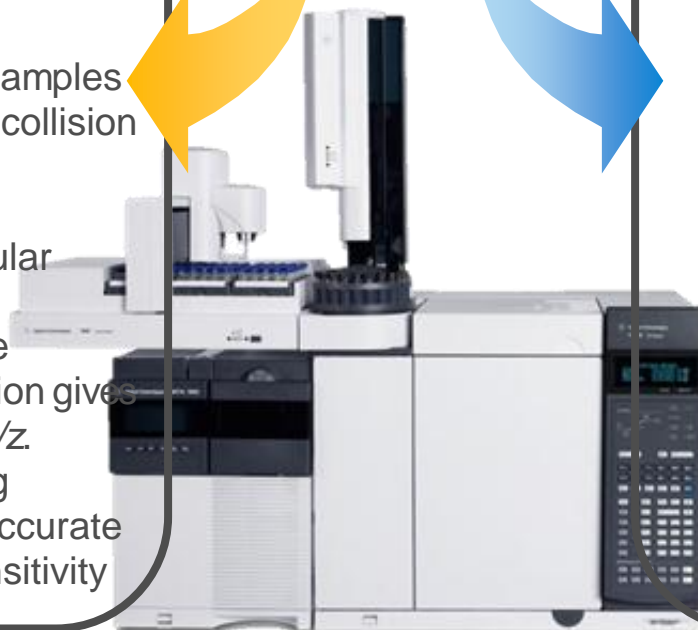
## GC/MS Analysis

### Advantages:

- GC resolves complex samples
- Ionization produced by collision with 70 eV electrons
- Fragmentation pattern characteristic of molecular structure
- Many libraries available
- GC/QToF instrumentation gives 2-5 ppm accuracy in  $m/z$ .
- Selected ion monitoring
- GC/QQQ allows very accurate quantitation at high sensitivity

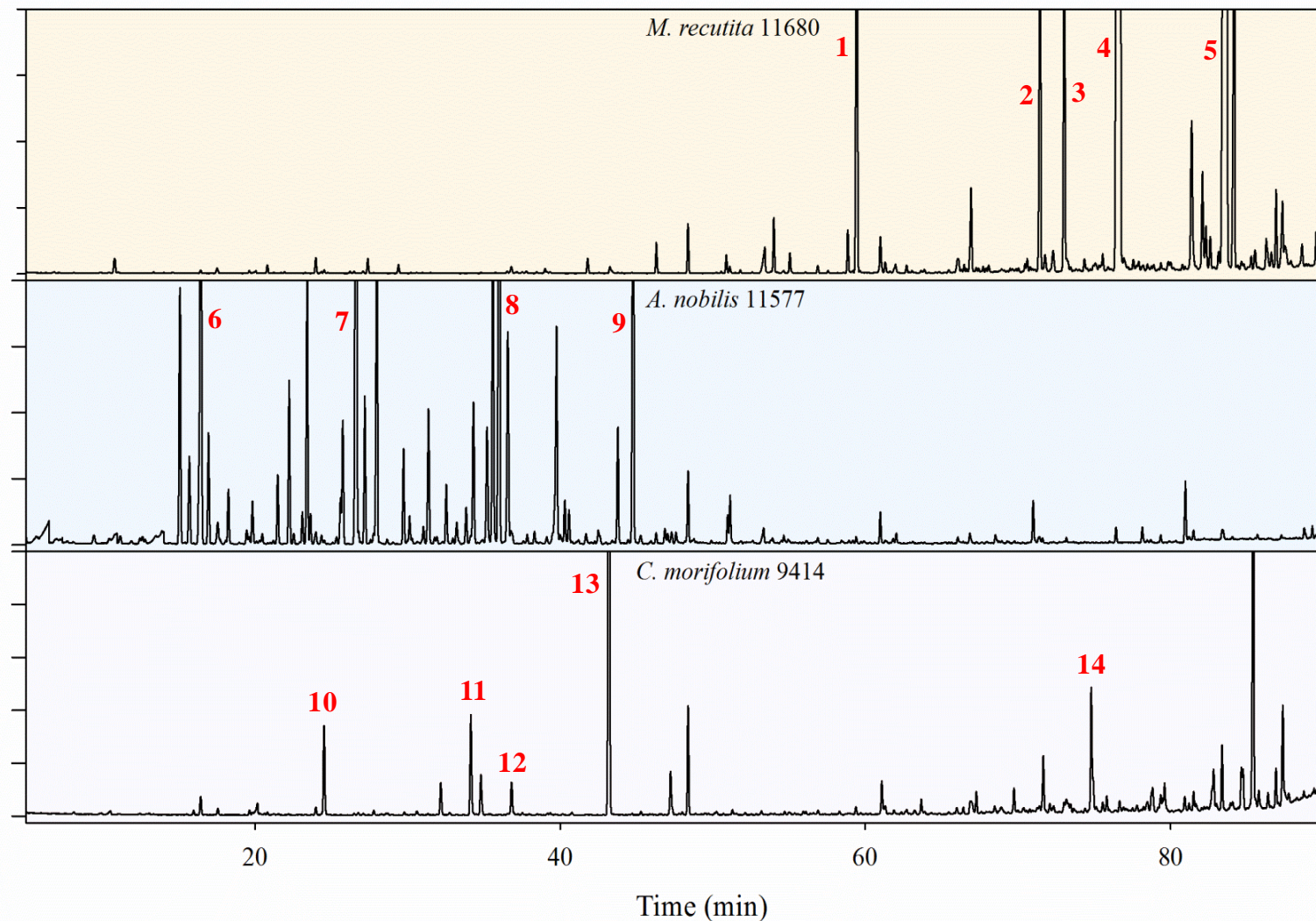
### Disadvantages:

- Sample must be volatile (300 °C)
- Complex and expensive instrumentation
- Fragmentation sometimes destroys molecular ion.



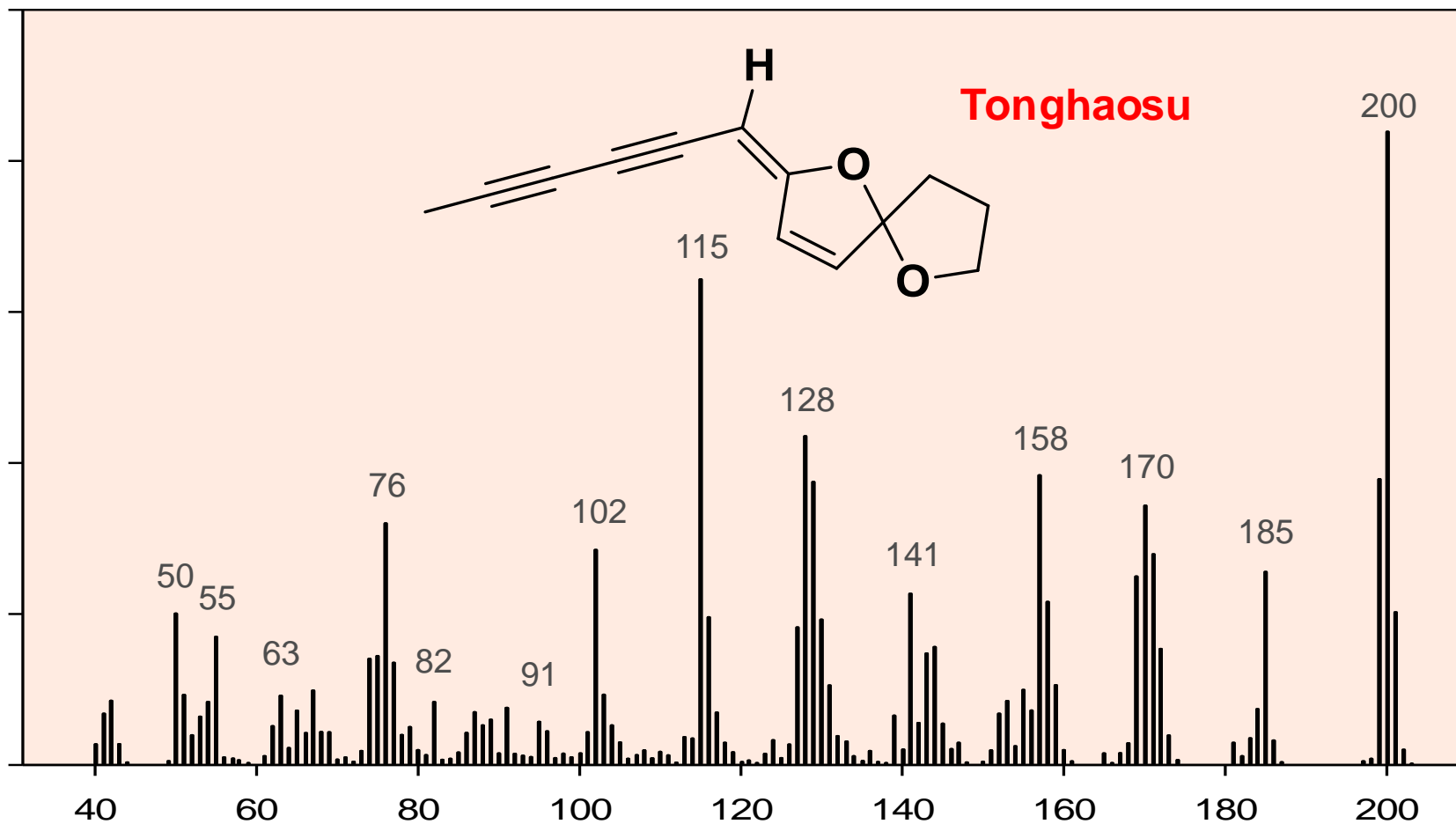


# What Can We Get From GC/MS?





# What Can We Get From GC/MS?



# How Did We Study Chamomile?

Preparation Data for Statistical Analysis Using *AMDIS*

Noise Analysis

Component Perception

Spectrum Deconvolution

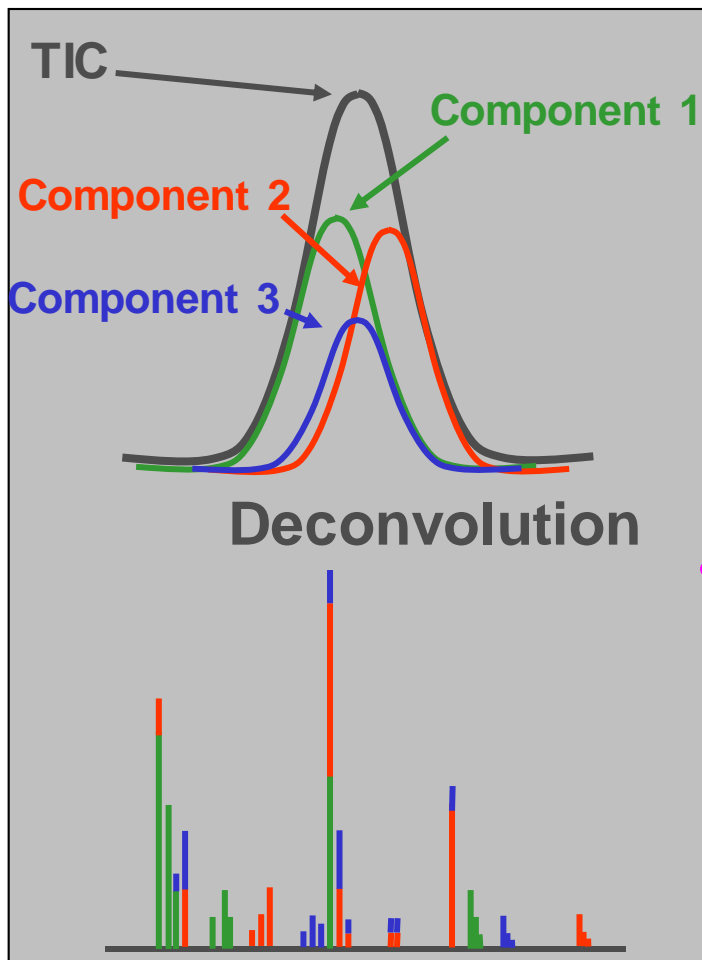
Compound Identification

Automated  
*M*ass Spectral  
*D*econvolution  
*I*dentification  
*S*ystem

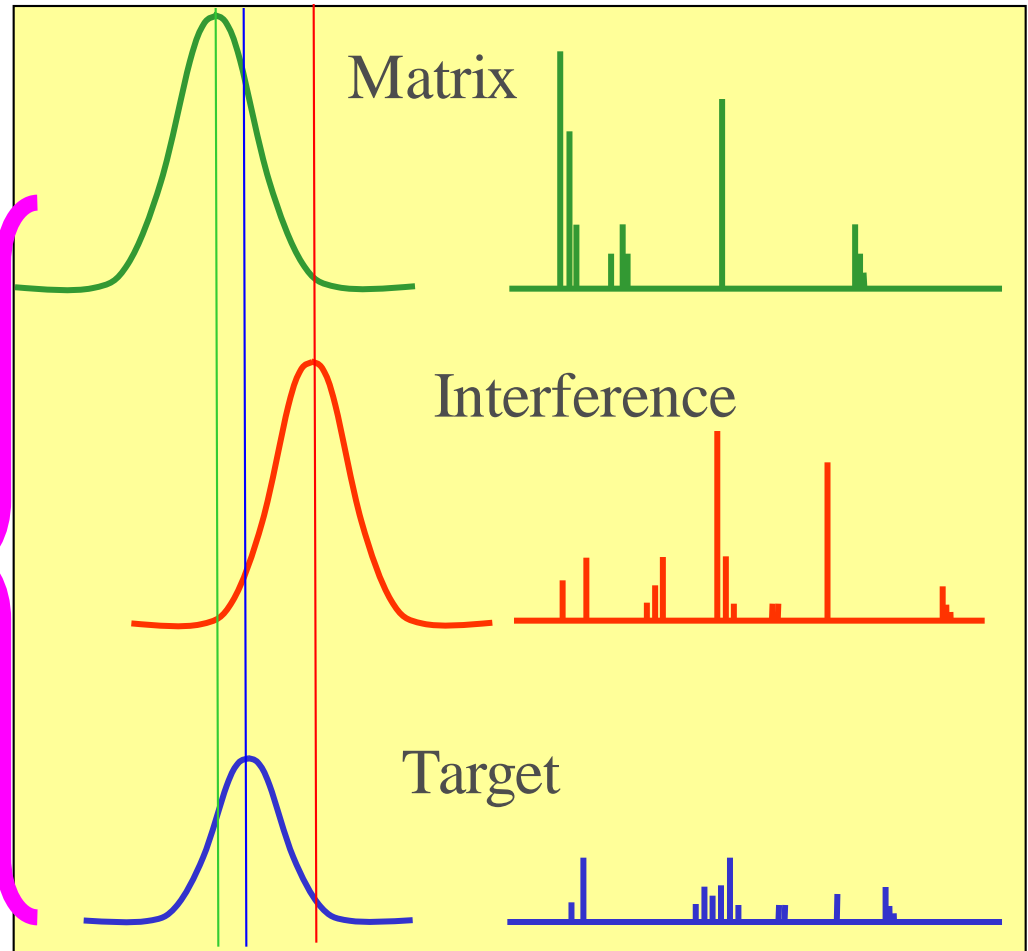


# Pulls Out Individual Components and Their Spectra

## TIC and Spectrum.

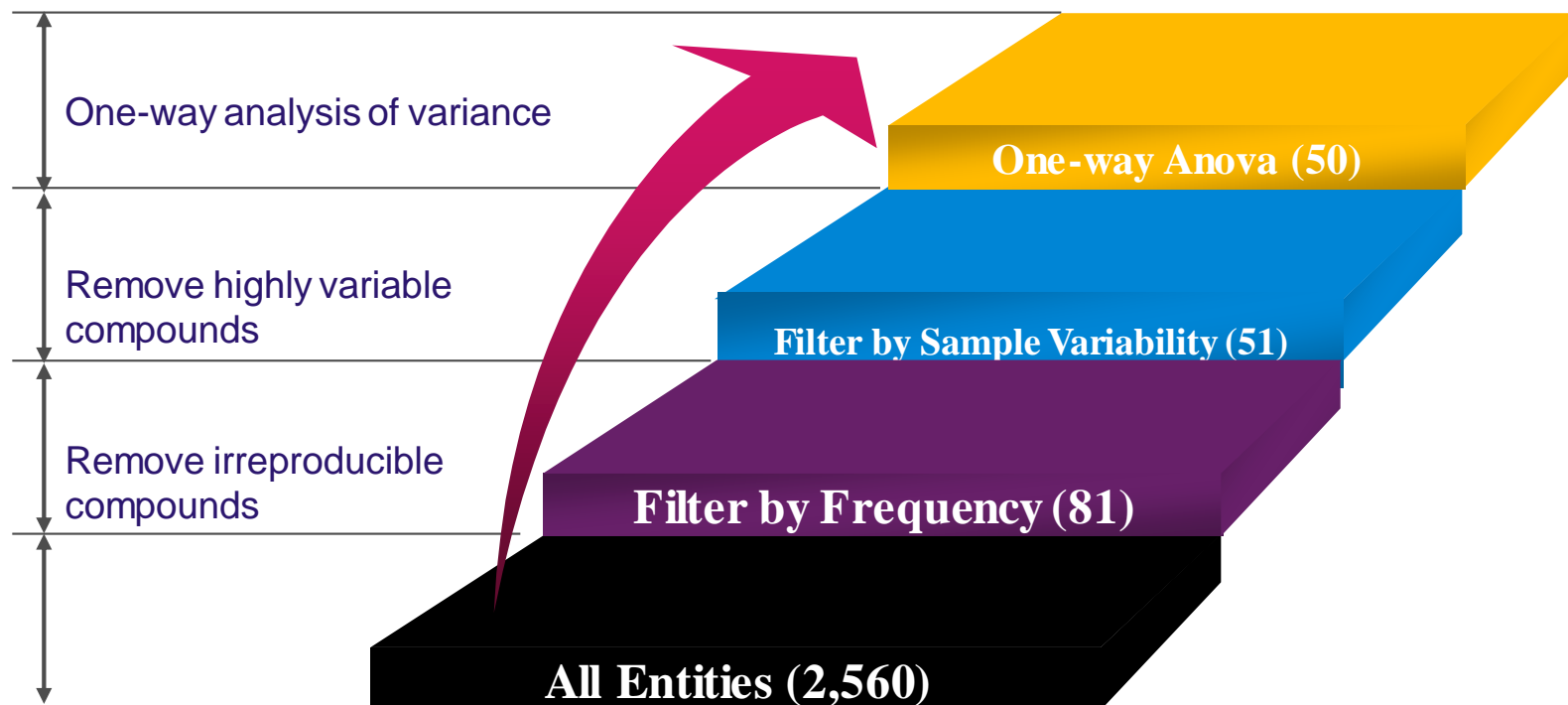


## Deconvoluted Peaks and Spectra.



# How Did We Study Chamomile?

**Find Meaningful Differences in Sample Sets Using Agilent *Mass Profiler Professional***

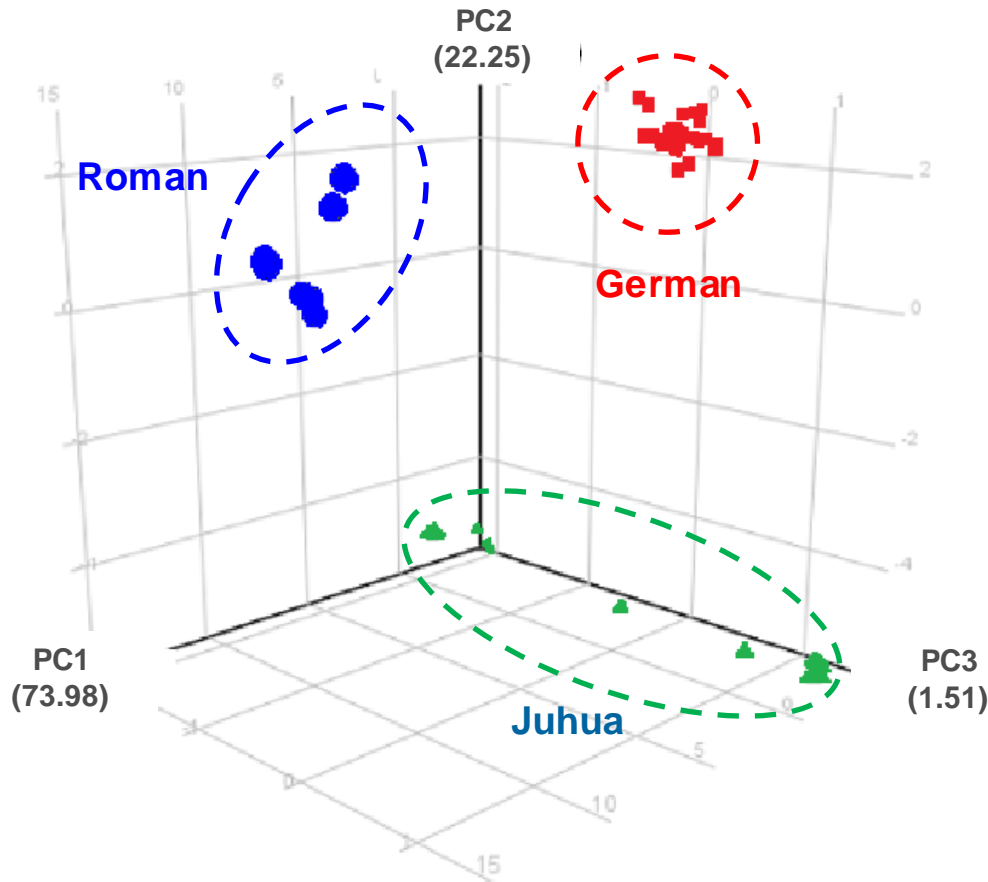


- To explore the most characteristic markers representing different chamomiles
- To reduce the dimensionality of the data



# How Did We Study Chamomile?

## Principal Component Analysis (PCA)



### Uses for:

- Data Visualization
- Data Reduction
- Data Classification
- Trend Analysis

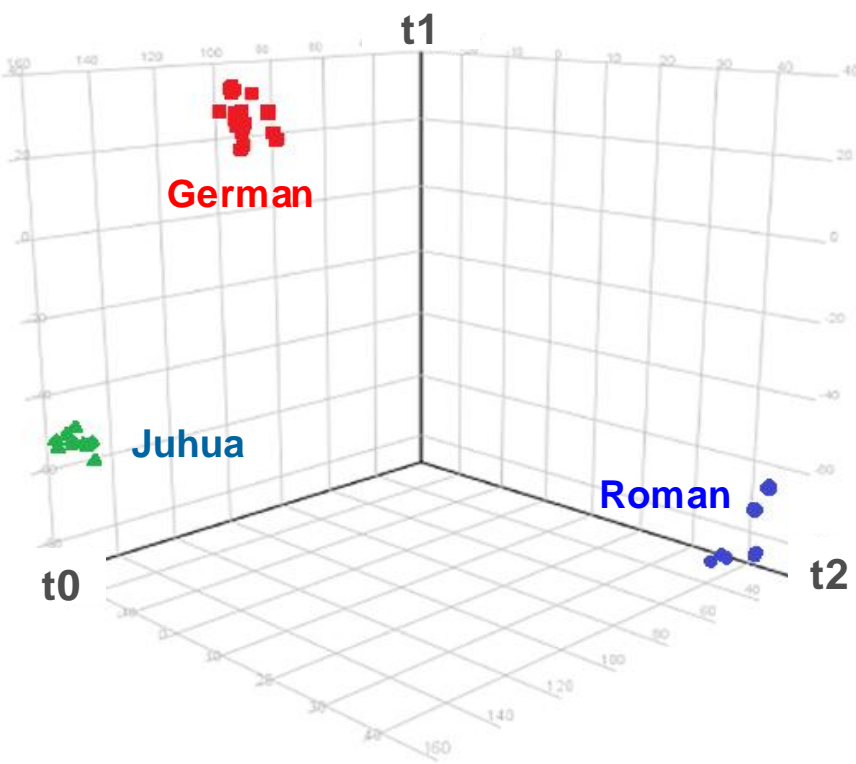
### Solved Problems in the Study:

- How many unique “sub-sets” are in the samples?
- How are they similar / different?
- What are the underlying factors that influence the samples?
- Which measurements are needed to differentiate?



# How Did We Study Chamomile?

Sample Class Prediction Model – Partial Least Squares Analysis (PLS-DA)



	German	Roman	Juhua	Accuracy (%)
Model Training				
German	15	0	0	100.0%
Roman	0	4	0	100.0%
Juhua	0	0	8	100.0%
Recognition Ability (%)				100.0%
Model Validation				
German	4	0	0	100.0%
Roman	0	4	0	100.0%
Juhua	0	0	4	100.0%
Prediction Ability (%)				100.0%

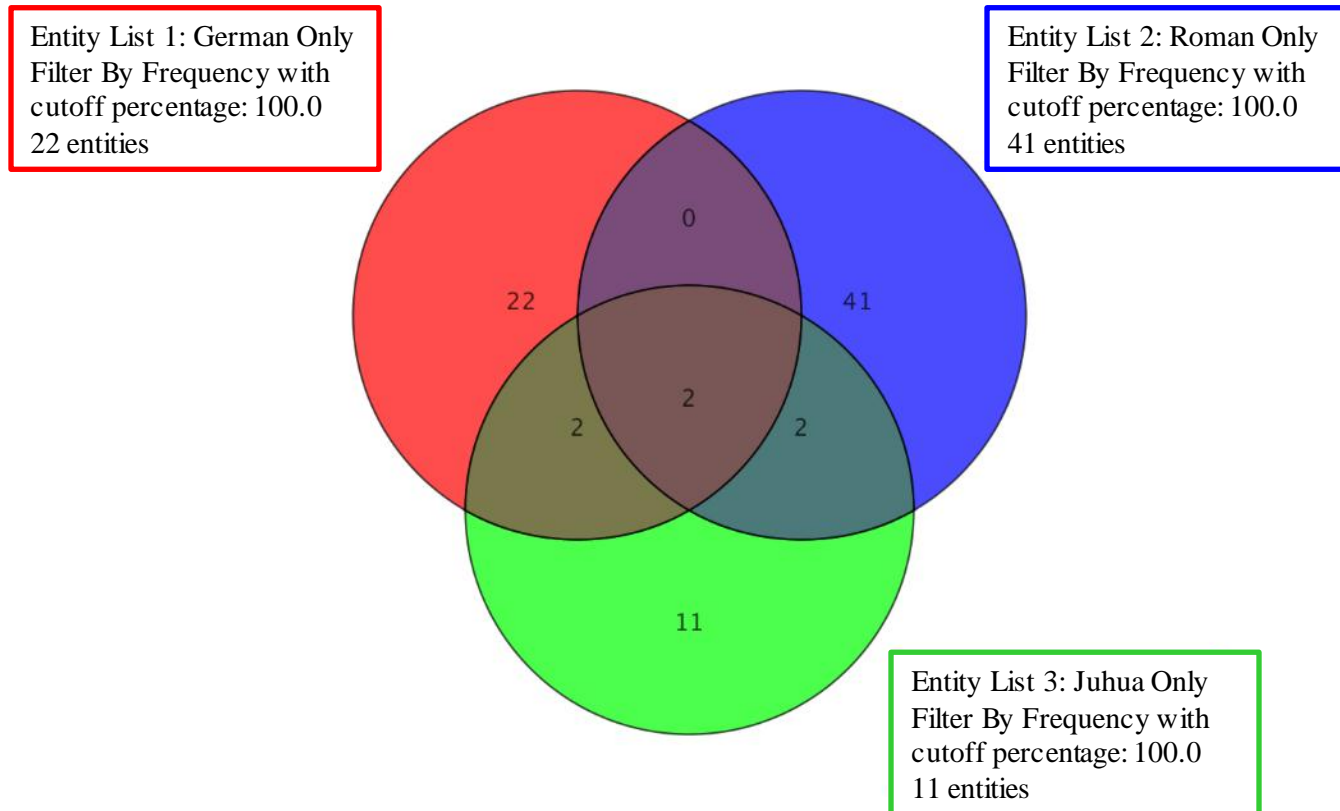
# How Did We Study Chamomile?

NCNPR Accession Code	Product Information from the Label	Predicted	Confidence Measure
<b>2061</b>	<b>Roman chamomile</b>	<b>German</b>	<b>0.47</b>
3670	Chamomile flower	German	0.92
<b>3998</b>	<b>Chamomile extracts</b>	<b>German</b>	<b>0.53</b>
4903	Chamomile powder	German	0.90
5770	Chamomile powder	German	0.93
9357	Chamomile flowers	German	0.82
9359	Chamomile flowers	German	0.84
9362	Chamomile flowers	German	0.84
9364	Chamomile flowers	German	0.92
9382	Chamomile Organic Tea (Leaves and flowers)	German	0.94
9383	Herbal Chamomile & Fruit Tea (Rosehips, chamomile, orange peel, lemon peel & lemon myrtle)	German	0.72
<b>9384</b>	<b>Chamomile Herb Tea</b>	<b>German</b>	<b>0.58</b>
9385	Organic Tea	German	0.81
9386	Carrington Tea-Chamomile	German	0.75
9387	Chamomile Herbal Tea	German	0.91
9388	Chamomile Herb Dietary Supplement	German	0.89
9389	Chamomile Herbal Tea	German	0.61
9390	Chamomile Herbal Tea	German	0.92
9391	Chamomile Herbal Tea	German	0.77
9393	Whole German Chamomile Flowers	German	0.87
9423	Chamomile Herbal Dietary Supplement	Juhua	0.83
9424	Chamomile Herbal Dietary Supplement	Juhua	0.84
<b>9425</b>	<b>Chamomile Herbal Dietary Supplement</b>	<b>Juhua</b>	<b>0.60</b>
9426	Chamomile Herbal Dietary Supplement	Juhua	0.86
9428	Chamomile Herbal Dietary Supplement	Juhua	0.82
9429	Chamomile Herbal Dietary Supplement	Juhua	0.81
9432	Chamomile Herbal Dietary Supplement	Juhua	0.99



# How Did We Study Chamomile?

## *Data Evaluation – Venn Diagram*



# How Did We Study Chamomile?

## Markers Identified From Venn Diagram

Entities		Tentative NIST Identification	Molecular Weight	CAS Number
m/z	t <sub>R</sub> (min)			
<i>Roman Chamomile</i>				
71.0	15.10	Isobutyric acid, isobutyl ester	144	97-85-8
71.0	23.42	Isobutyric acid, 2-methylbutyl ester	158	2445-69-4
55.0, 83.0	26.64	Butyl Butenoate <sup>a,b</sup>	156	54056-51-8
83.0	39.75	3-Methyl-2-butenic acid, 3-methylbut-2-enyl ester	168	299309
100.0	44.75	Hexyl Butenoate	324	60129-26-2
<i>German Chamomile</i>				
205.0	66.94	Spathulenol	220	77171-55-2
143.0	71.43	$\alpha$ -Bisabolol oxide B <sup>a,b,c</sup>	238	26184-88-3
93.0, 141.0	73.04	$\alpha$ -Bisabolol <sup>a,b,c</sup>	222	515-69-5
143.0	76.07	Bisabolol oxide A <sup>a,b,c</sup>	238	22567-36-8
128.0	83.70	<i>E</i> -1,6-Dioxaspiro[4.4]non-3-ene, 2-(2,4-hexadiynylidene)-	200	50257-98-2
<i>Juhua</i>				
95.0	36.82	Borneol	154	10385-78-1
132.0	61.06	$\alpha$ -Curcumene	202	644-30-4
91.0	67.27	Caryophyllene oxide	220	1139-30-6
105.0, 121.0	69.75	Alloaromadendrene oxide	220	156128
204.0	71.69	Eudesm-7(11)-en-4-ol	222	473-04-1



# A CASE STUDY ON CHARACTERIZATION OF OLIVE OIL

Authentication of “Extra Virgin” Purity



# Goals for Olive Oil Characterization Study

Initial screening with high-res TOF

Confirm Molecular Ion with PCI

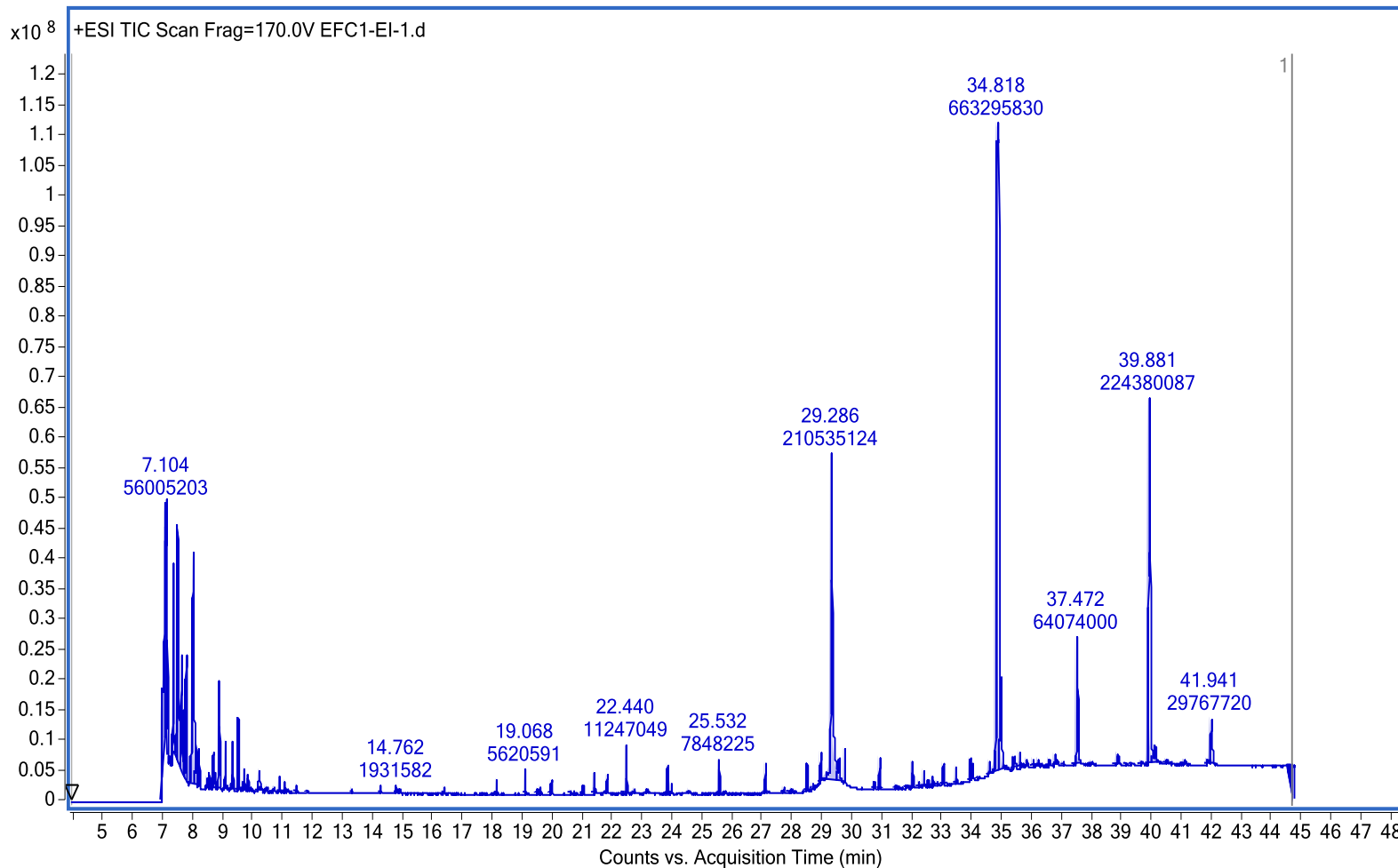
Mass Profiler Professional (MPP) analysis of data

- Generate statistical data on representative samples
- Build Class Prediction model based on sensory Pass/Fail information
- Test model on additional samples of known quality



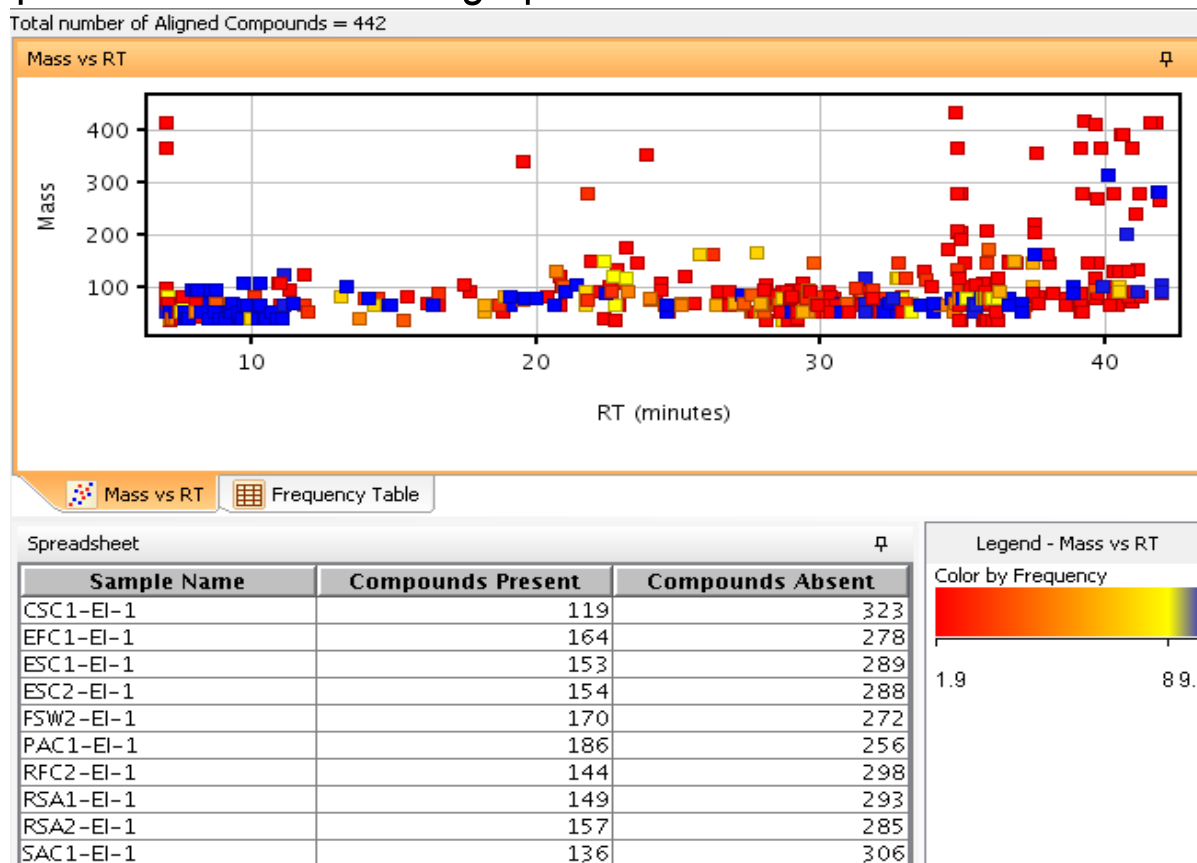
# Total Ion Current of Extra Virgin Olive Oil

About 150 peaks with relative area filter of 0.1% of largest peak



# Statistical Data on Olive Oil Samples

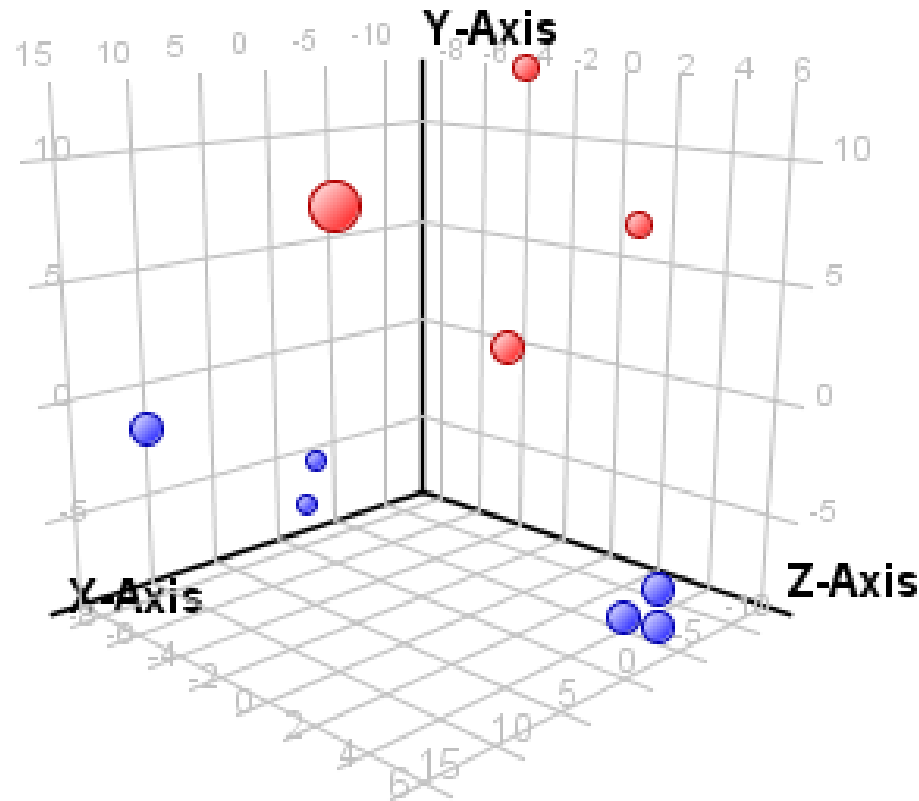
## 442 Unique Compounds from Chromatographic Deconvolution



*Most occurred only once or twice and were filtered out by MPP*  
*Each sample is defined by passed/failed sensory evaluation*



# Principle Component Analysis (PCA) Data Clusters

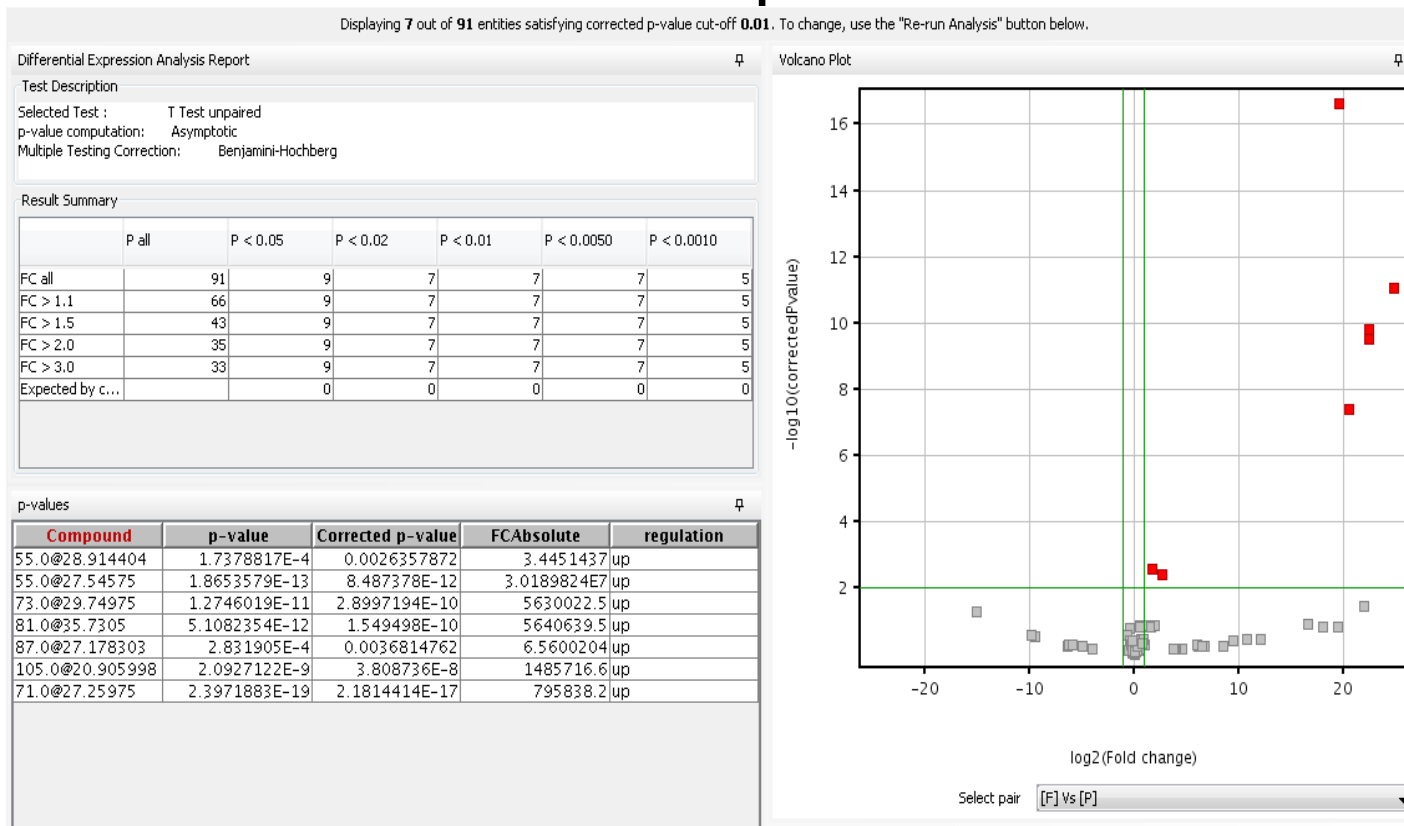


*Samples initially evaluated by sensory tests:*  
“Failed” marked red  
“Passed” marked blue





# Fold-Change Analysis Reveals Compounds Accumulated in “Failed” Samples



*The Volcano Plot on the right shows fold-change for each entity on the x-axis and significance on the y-axis.*



# Compounds Associated with “Failed” Sensory Test

Tentative NIST ID	NIST Match	Formula	CAS	Odor	Source
n-Hexadecanoic acid	789	C16H32O2	57-10-3	Faint Oily	Bedoukian Research
Octadecanoic acid, ethyl ester	703	C20H40O2	111-61-5	Waxy	The Good Scents Company
2,6,10,14,18,22-Tetracosahexaene, 2,6,10,15,19,23-hexamethyl-	831	C30H50	111-02-4	Floral	The Good Scents Company
α-Cubene	880	C15H24	17699-14-8	Herbal	The Good Scents Company
None: confirmed by PCI data as 2H-Pyran-2-one, tetrahydro-6-nonyl-	574??	C14H26O2	2721-22-4	Waxy	The Good Scents Company

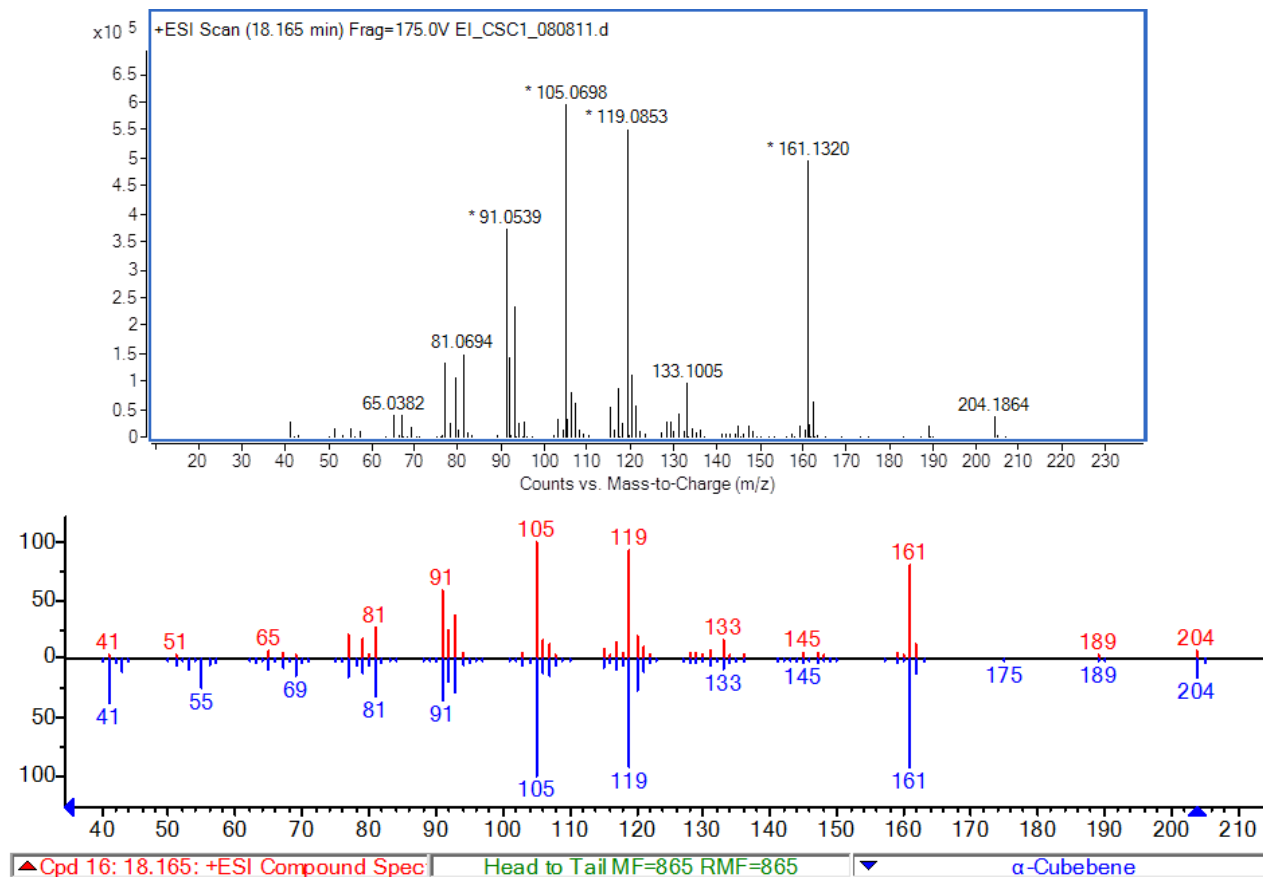
Lower NIST match factor indicates that tentative NIST ID may not be correct (actual compound not included in combined NIST Wiley database), but data gives a reasonable estimate of empirical formula

Tentative NIST ID	NIST Match	Formula	CAS	EI [M*]+			PCI [M+H]+		
				Calculated	Measured	Mass Error (PPM)	Calculated	Measured	Mass Error (PPM)
n-Hexadecanoic acid	789	C16H32O2	57-10-3	256.2397	256.2385	4.683115	257.2475	257.2470	1.9437
Octadecanoic acid, ethyl ester	703	C20H40O2	111-61-5	312.3023	312.3008	4.803039	313.3101	313.3091	3.1917
2,6,10,14,18,22-Tetracosahexaene, 2,6,10,15,19,23-hexamethyl-	831	C30H50	111-02-4	410.3907	410.39037	0.816295	411.3985	411.3987	0.4861
α-Cubene	880	C15H24	17699-14-8	204.1873	204.1883	4.897464	205.1951	205.1945	2.9240
None: confirmed by PCI data as 2H-Pyran-2-one, tetrahydro-6-nonyl-	574??	C14H26O2	2721-22-4	226.1927	ND	ND	227.2006	227.1987	8.3627

Compound with lowest NIST match also has highest deviation (ppm) for PCI



# EI Library Search of Un-regulated Compounds



*Commercial unit mass EI spectral libraries like Wiley and NIST can be searched using accurate mass EI TOF data to identify compounds.*

# PCI Confirmation of Molecular Ion

Allowed Species Limits Scoring

Mass and charge

Mass or m/z: 205.1945

Charge: 1

Charge carrier

Positive ions: H Negative ions: H

MS ion electron state: even electron

Elements and limits

Element	Minimum	Maximum
C	3	60
H	0	120
O	0	30
N	0	30
S	0	5
Cl	0	0
[13C]	0	0

Formula (M)	Score (MFG)	Mass	Mass (MFG)	m/z (Calc)	Diff (ppm)
C15 H24	100	204.1872	204.1878	205.1951	2.83
C10 H24 N2 O2	99.94	204.1872	204.1838	205.1911	-16.87
C9 H24 N4 O	99.7	204.1872	204.195	205.2023	38.14
C11 H24 O3	98.96	204.1872	204.1725	205.1798	-71.89
C7 H20 N6 O	98.55	204.1872	204.1699	205.1771	-85.04
C8 H24 N6	98.26	204.1872	204.2062	205.2135	93.16
C10 H24 N2 S	97.85	204.1872	204.166	205.1733	-103.85
C12 H28 O2	97.75	204.1872	204.2089	205.2162	106.31
C13 H20 N2	97.13	204.1872	204.1626	205.1699	-120.35
C8 H20 N4 O2	96.16	204.1872	204.1586	205.1659	-140.06
C11 H24 O S	95.11	204.1872	204.1548	205.1621	-158.86

Base formula (M)

C15 H24 O0

Species to calculate

☒ Positive ions ☐ Negative ions

- ☐ Neutral
- ☒ Radical
- ☒ +H
- ☐ +Na
- ☐ +K
- ☒ +C2H5
- ☒ +C3H5

Number of charges: 1

Species	Diff (ppm)	Defect
M <sup>+</sup>	204.1873	0.1873
(M+H) <sup>+</sup>	205.1951	2.81
(M+CH5) <sup>+</sup>	221.2264	0.2264
(M+C2H5) <sup>+</sup>	233.2264	0.2264
(M+C3H5) <sup>+</sup>	245.2264	0.2264

Mass and formula calculators used to determine empirical formula



# Compounds Associated with “Failed” Sensory Test

Tentative NIST ID	NIST Match	Formula	CAS	Odor	Source
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2,6,10,14,18,22-Tetracosahexaene, 2,6,10,15,19,23-hexamethyl-	831	C30H50	111-02-4	410.3907	410.39037	0.816295	411.3985	411.3987	0.4861
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Compound with lowest NIST match also has highest deviation (ppm) for PCI



# What Did We Learn?



**Chemometrics can be used to analyze large, complex (3-D) data sets MUCH faster than manual analysis**



**Mass Profiler Professional allows easy statistical analysis of MS data.**



**With AUTHENTICATED samples, an accurate sample class prediction model can be developed and verified**



**The SCP model can subsequently be used to analyze samples in an automated manner w/o reanalysis of the authenticated samples**



**Chemometric analysis can be used to identify potential markers for different type of samples**



# Thank you for your attention!

## Any Questions?



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