IAEA Analytical Quality in Nuclear Applications Series No. 61

# Certification of Polycyclic Aromatic Hydrocarbon Mass Fractions in IAEA-477 Sediment Sample



# CERTIFICATION OF POLYCYCLIC AROMATIC HYDROCARBON MASS FRACTIONS IN IAEA-477 SEDIMENT SAMPLE

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INTERNATIONAL ATOMIC ENERGY AGENCY VIENNA, 2020

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#### FOREWORD

The IAEA Environment Laboratories assist Member States in understanding, monitoring and protecting both the terrestrial and the marine environment. To assess the impact of land based and sea based pollution sources on marine coastal environments, it is imperative to ensure the quality and comparability of the analytical data generated by national and regional pollution monitoring programmes. For this purpose, the IAEA Environment Laboratories have been assisting national laboratories and regional laboratory networks through the Reference Products for Science and Trade programme since the early 1970s. This is accomplished through the production of certified reference materials, the provision of training in quality assurance, and the evaluation of measurement performances through worldwide and regional interlaboratory comparison exercises and proficiency tests.

This publication describes the production of the certified reference material IAEA-477 produced by the IAEA Environment Laboratories following ISO Standard ISO 17034:2016 and ISO/Guide 35:2017. This material is a sediment sample with certified mass fractions of polycyclic aromatic hydrocarbons. The assigned final values and their associated uncertainties were derived from the results provided by selected laboratories with demonstrated technical and quality competence, following the guidance provided in ISO/Guide 35:2017 and in the Guide to the Expression of Uncertainty in Measurement.

The material is intended to be used for quality control and assessment of method performance in the determination of mass fractions of polycyclic aromatic hydrocarbons, which are included in the group of priority substances in many environment monitoring programmes.

The IAEA is grateful to the Government of Monaco for the support provided to its Environment Laboratories and to the Permanent Mission of Australia to the United Nations in New York and its International Development Fund for their financial support under the IAEA's Peaceful Uses Initiative. The IAEA also acknowledges all the laboratories that participated in the characterization study of this reference material, and the Australian Nuclear Science and Technology Organisation and James Cook University for conducting the sampling mission. Planning of the sampling mission and collection of the sediment samples was performed by J. Daniell of James Cook University, and M. Johansen, E. Prentice, H. Heijnis and M. Corry of ANSTO. The IAEA officers responsible for this publication were I. Tolosa and R. Cassi of the IAEA Environment Laboratories.

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#### **1. INTRODUCTION**

The Marine Environmental Studies Laboratory (MESL) of the IAEA-EL provides assistance to Member States' laboratories to enhance the quality of the analytical measurement results for trace elements and organic contaminants in marine environmental samples. This is achieved through the production of CRMs, organization of Interlaboratory Comparisons and Proficiency Tests, and by conducting Training Courses on the analysis of contaminants in marine samples. This activity is undertaken in the framework of the IAEA-EL's Sub-Programme 2.4.1 "Reference Products for Science and Trade" and the Project 2.4.1.1 "Provision of Reference Products and Assurance of Laboratory Performance".

CRMs are valuable tools for analytical method development and validation to improve the measurements and quality control in analytical laboratories. More specifically, marine matrix CRMs are needed to ensure the reliability of the analytical measurements and ensure the use of high quality data as the basis for decision making in national or regional marine pollution monitoring programmes. Polycyclic aromatic hydrocarbons (PAHs) comprise a large group of more than several hundred chemical compounds containing two or more fused aromatic rings. They are produced during incomplete combustion of organic matter or released from oil spills. Because many of the PAHs compounds are known to be carcinogenic and genotoxic, they are included under the group of priority substances (PSs) in many environment monitoring programmes. The most important lists of monitored PAHs are a group of 16 PAHs listed by the U. S. Environmental Protection Agency (EPA), and the 15+1 European Union (EU) priority PAHs to be monitored in foods. While there are several CRMs certified for PAHs, there is still a noticeable lack of matrix CRMs, in particular in marine sediments, where the concentrations levels are in the low ng  $g^{-1}$  range. To meet this need, MESL has developed a CRM in a natural and remote sediment for the determination of PAHs in the environment monitoring programmes.

This publication describes and provides information on the sample preparation methodology followed and on the assignment of property values with their associated uncertainties for a number of polycyclic aromatic hydrocarbons in a marine sediment sample. Certification of the mass fractions was made for major polycyclic aromatic hydrocarbons (PAHs). The basic principles for evaluation of measurement uncertainty were followed according to the ISO Guide 35 [1] and the Guide to the Expression of Uncertainty in Measurement (GUM) [2], which combines the different uncertainties of characterization, heterogeneity and instability.

The CRM IAEA-477 was produced to satisfy the needs of laboratories to strengthen data quality assurance in the analysis of priority substances in marine samples. The low concentration levels of this CRM make it a suitable material for monitoring PAHs in marine environments where the concentration levels are relatively low and close to the method detection limits.

#### 2. METHODOLOGY

#### 2.1 COLLECTION AND PREPARATION OF THE MATERIAL

Two sediment samples were collected in Queensland region, Australia using a Van Veen type sampler. One site was sampled in Townsville Marina and the other in Townsville Ross River. Frozen samples were sent from Brisbane to IAEA laboratories in Monaco, and then sent for freeze-drying to LYOFAL (France) and to micronization at 20  $\mu$ m to AFT-Micro-Macinazione (France). Sediment was further  $\gamma$ -irradiated at doses ranging from 27 to 39.5 kGy at Steris (France). The final powdered composite sediment, about 15 kg, was homogenized by mixing it in a stainless-steel rotating drum for three weeks. Then, aliquots of about 50 grams were packaged into cleaned amber glass bottles with screw caps and labeled IAEA-477.

The particle size analysis was performed by laser diffraction light using a Mastersizer 2000 equipped with a Sirocco dispenser for a range of particles from 0.1 to 1000  $\mu$ m. The sample showed 100% of particulates below 20  $\mu$ m (Figure 1).



Fig. 1. Particle size distribution of IAEA-477.

#### 2.2. SELECTION OF LABORATORIES FOR THE CHARACTERIZATION STUDY

The selection of participating laboratories was based on the results they have provided during previous ILCs and characterization exercises for the same compounds in sediment. Only laboratories that provided reliable and satisfactory data by applying valid quality control and quality assurance procedures were invited to participate in this characterization study.

One bottle of sediment sample was sent to each participating laboratory, accompanied by an information sheet and a reporting form to include the data results for polycyclic aromatic

hydrocarbons (PAHs). The participating laboratories were requested to analyse six aliquots using their usual technique, and the reported results had to be accompanied with the description of the method used together with the applied quality control procedures, including results for the organic contaminants in a CRM with a matrix similar to the candidate reference material.

The list of laboratories participating in the characterization study is provided on page 83.

#### 2.3. HOMOGENEITY ASSESSMENT

In order to establish the degree of homogeneity of the reference material with respect to the properties of interest, both within- and between-unit homogeneity have to be evaluated to ensure that all distributed units are the same within the stated uncertainty [1].

Homogeneity test was performed by IAEA-EL in Monaco after the bottling of the sample material. The between-bottle homogeneity of the material was assessed by determining the concentration of selected parent PAHs in duplicate analysis from 10 bottles (5 g sample intake) randomly selected during the whole bottling process of the bulk dry powder. The analytic method used included microwave extraction with hexane/methylene chloride (1:1), sulphur removal by copper and fractionation by solid phase extraction (SPE) to isolate PAHs from aliphatic hydrocarbons. The final measurements of PAHs were performed by gas chromatography coupled to mass spectrometry using selected ion monitoring (GC-MS-SIM) under "quasi" repeatability conditions and in a randomized order to be able to separate an analytical drift from a trend in the filling sequence.

#### 2.4. CHARACTERIZATION

Characterization refers to the process of assigning a property value that can be reliably assessed when its value is confirmed by several laboratories working independently and using different methods, for each of which the accuracy has been well established [3]. The material was initially analysed in the IAEA-EL in Monaco. The final characterization was based on the results provided by selected laboratories with demonstrated technical and quality competence. Although having a formal accreditation was not mandatory, all laboratories participating in the characterization campaign provided their results with their method validation data in accordance with the guidelines of ISO 17025 [4].

As it is shown in Table 1, the characterization and the value assignment approach for the PAH analytes included a combination of results derived from analyses using different solvents and extraction techniques, a variety of different cleanup and fractionation procedures and diverse instrumental detection techniques. Consequently, 12 independent datasets were obtained for PAHs.

YTICAL METHODOLOGIES USED FOR THE CERTIFICATION OF PAHS IN CRM IAEA-477	tion Procedure Solvent Cleanup Fractionation Equipment Chromatographic Column	Aicrowave         DCM         SPE         Silica/Alumina         GC-MS         5% Phenyl DMPsiloxane & Select PAH	Aicrowave Acetone/n-Hexane GPC <sup>2</sup> Silica GC-MS/MS 5% Phenyl DMPsiloxane	ASE n-Hexane/DCM None None HPLC-FLD C18	Sohxlet DCM Silica Silica GC-MS 5% Phenyl DMPsiloxane	Sohxlet n-Hexane/ DCM SPE Silica GC-MS 5% Phenyl DMPsiloxane	ASE n-Hexane/ DCM SPE Silica GC-MS 50% Phenyl DMPsiloxane	IP hot LE <sup>1</sup> Toluene/Acetone None None GC-HRMS 5% Phenyl DMPsiloxane	Aicrowave Acetone/n-Hexane Adsorption Column Silica GC-MSMS 5% Phenyl DMPsiloxane	Sohxlet Methanol/ DCM Saponification Silica GC-MS 5% Phenyl DMPsiloxane	ASE n-Hexane/ DCM Silica Silica GC-MS 5% Phenyl DMPsiloxane	ASE Acetone/DCM None None GC-MS 5% Phenyl DMPsiloxane	
I. ANALYTICAL METH	<b>Extraction Procedure</b>	Microwave	Microwave	ASE	Sohxlet	Sohxlet	ASE	HP hot LE <sup>1</sup>	Microwave	Sohxlet	ASE	ASE	
TABLE 1.	Lab	1	2	б	4	5	9	L	6	10	11	12	<b>c</b>

<sup>1</sup> High performance hot liquid extraction 

<sup>2</sup> Gel permeation chromatography

The extraction procedures of the PAHs were performed through accelerated solvent extraction (ASE), microwave, Soxhlet, high performance hot liquid extraction by using different solvents mixtures, including dichloromethane (DCM), hexane, acetone, toluene and methanol. After further cleanup with copper/sorbents and/or fractionation with different adsorbents, PAHs were characterized using four different analytical techniques as summarized in Table 2 and Figure 2: gas chromatography/mass spectrometry (GC-MS), gas chromatography/high resolution mass spectrometry (GC-HRMS), high performance liquid chromatography/fluorescence detector (HPLC–FLD), and gas chromatography coupled to tandem mass spectrometry. A reversed-phase C18 column was used in the HPLC and three GC stationary phases of different selectivity, e.g. the 5 % phenyl-substituted methylpolysyloxane phase, the 50% phenyl-substituted methylpolysiloxane phase.

Method code	Instrumental technique
GC-MS	Gas chromatography/mass spectrometry
GC-HRMS	Gas chromatography/high resolution mass spectrometry
HPLC-FLD	High performance liquid chromatography/fluorescence detector
GC-MS/MS	Gas chromatography coupled to tandem mass spectrometry





*Fig. 2. Analytical methods used for the characterization of PAHs in the IAEA-477 sediment sample. Abbreviations used to describe the instrumental techniques are given in Table 1.* 

#### 2.5. MOISTURE CONTENT

Participants were required to determine the moisture content of the lyophilized material received by drying an aliquot to a constant mass at 105°C. The moisture content of the lyophilized material, as the average of the mean value of 9 participants was found to be  $0.9 \pm 0.4\%$ .

#### **3. RESULTS AND DISCUSSION**

#### 3.1. RESULTS OF THE HOMOGENEITY STUDY

#### 3.1.1. Between-unit and within-unit homogeneity

To assess the uncertainty contribution related to the inhomogeneity, 10 bottle units (about 3% of the total batch) of sediment samples were analysed for selected PAHs under repeatability conditions. Each bottle unit was extracted and analysed in duplicate, resulting in 2 independent data values by bottle unit.

Grubbs-tests at 95% and 99% confidence levels were performed to identify potentially outlying individual results as well as outlying bottle means. Few individual results and unit means were detected as outliers at 95% and 99% confidence levels. One technical reason for outlying results was identified for the high molecular weight (HMW) PAHs due to a blockage of a SPE cleanup column, which required the use of the vacuum pump to collect the HMW PAHs. In this context, the individual results of the outlying bottle means were rejected, whereas all other individual results following a normal distribution, were retained for evaluating the between-unit homogeneity.

Quantification of between-unit homogeneity was estimated according to ISO Guide 35 [1] by analysis of variance (ANOVA) which can separate the between-unit variation ( $s_{bb}$ ) from the within-unit variation ( $s_{wb}$ ). The latter is equivalent to the method repeatability if the individual aliquots are representative for the whole unit. ANOVA allows the calculation of within-unit standard deviation  $s_{wb}$  using Eq. (1) and also between-units standard deviation  $s_{bb}$ , using Eq. (2):

$$s_{wb} = u_{wb} = \sqrt{MS_{wb}} \tag{1}$$

$$s_{bb} = u_{bb} = \sqrt{\frac{MS_{bb} - MS_{wb}}{n}} \tag{2}$$

 $s_{bb}$  and  $s_{wb}$  are estimates of the true standard deviations and are therefore subject to random fluctuations. In some cases, the mean square between groups ( $MS_{bb}$ ) can be smaller than the mean squares within groups ( $MS_{wb}$ ), resulting in negative arguments under the square root used for the estimation of the between-unit variation. In this case,  $u^*_{bb}$ , the maximum between-unit

variability that could be masked by method repeatability, was calculated as described by Linsinger et al. [5] through Eq. (3):

$$u_{bb}^* = \frac{s_{wb}}{\sqrt{n}} \sqrt[4]{\frac{2}{\nu_{MSwb}}}$$
(3)

Where: n is the number of replicate sub-samples per bottle; and  $v_{MSwb}$  is the degrees of freedom of  $MS_{wb}$ .

As presented in Table 3, the between-unit variations (*s*<sub>bb</sub> and *u*<sub>bb</sub>\*) for most of the selected PAHs were between 2.1 and 5% small enough to ensure the homogeneity of the material at 5 g sample size. Only benzo[e]pyrene exhibited a relatively higher value (7.8%), probably related to the higher measurement variability of this HMW PAH at such low concentration level. The withinunit variation (s<sub>wb</sub>) derived from the ANOVA calculation was higher than the typical method repeatability (s<sub>meas</sub>: 1-5%) derived from the analyses of six replicates from the same bottle ( $s_{mes} = \frac{s}{\sqrt{6}}$ ), and also higher than the average  $s_{wbp}$  derived from the replicate data provided by the participant laboratories. Taking in account the close agreement among the replicate analysis provided by the laboratory participants, we might conclude that the within-unit homogeneity fits the purpose of this CRM.

The uncertainty  $u_{hom}$  associated with inhomogeneity of the material was estimated according to the ISO Guide 35 [1] by Eq. 4:

$$u_{hom} = \sqrt{u_{wb}^2 + u_{bb}^2} \tag{4}$$

where the  $u_{wb}$  was derived from the participant laboratories data and the  $u_{bb}$  was taken as the maximum values of the between-unit variations ( $s_{bb}$  and  $u_{bb}^*$ ). As it is presented in Table 3, the uncertainty contribution related to inhomogeneity was estimated to range from 3.3 to 8.4 %, and, thus, we set the uncertainty associated with inhomogeneity at 8 % for all PAH analytes.

		ANOVA		METHOD	PART.	
Compounds	Swb	Sbb	u <sub>bb</sub> *	Smeas (%)	Swbp	u <sub>hom</sub>
	%	%	%	%	%	%
Napthalene	8.3	-	4.1	4	4.2±2.0	5.8
Phenanthrene	5.1	5	2.5	3	2.6±1.7	5.6
Pyrene	5.5	2.1	2.7	1	2.4±1.3	3.6
Chrysene+triphenylene	5.1	-	2.5	1	2.1±1.0	3.3
Benzo[e]Pyrene	15.7	-	7.8	5	3.0±1.9	8.4
Perylene	6.6	3.4	3.3	5	3.1±1.8	4.6

TABLE 3. THE ESTIMATE OF INHOMOGENEITY CONTRIBUTIONS TO THE TOTAL UNCERTAINTY FOR THE SELECTED PARENT PAHS COMPOUNDS AND REPEATABILITY OF THE METHOD

#### 3.2. STABILITY STUDY

Stability information is important to determine the presence of any potential degradation of the analytes during sample transport to the customers (short-term stability) as well as conditions for storage (long-term stability) of the material. Time, temperature, moisture and radiation are usually the more pertinent parameters affecting stability of the samples. The influence of light was minimized by using amber bottles, which were stored in the dark and moisture was minimized by the freeze-drying preparation of the material. Consequently, only the influence of time and temperature are usually investigated by using an isochronous stability design.

As previous results on PAHs stability studies did not show any significant trend of degradation over the timeframe at different temperatures: +20°C and +40°C [6], no special precautions regarding temperature control during transport was taken. This approach is supported by the chemical nature of the PAHs which owe a high chemical stability and persistence. Therefore, no additional uncertainty with respect to instability due to transport was taken into account and the uncertainty associated with short term stability under transport conditions is taken as zero.

#### 3.2.1. Long-term stability study

Long-term stability evaluation aims to determine if the certified values of the analyte(s) remain valid during the lifetime of the certified reference material. Judging from similar materials issued from our IAEA laboratory and other organizations such as NIST, the expiry date of the CRMs was given to be 5-10 years after the date of certification if they were stored at temperatures less than 25-30°C and away from direct sunlight [7, 8]. As no measurements of long-term instability has been performed in this CRM, the uncertainty for long-term stability has been set as the intrinsic variability of the method s<sub>meas</sub>, which was set at 3% (phenantrene from Table 3). This is based on the experience on previous IAEA organic reference material where changes for long-term stability have not been detected, and therefore the uncertainty associated to the long-term stability is considered lower than the variability of the method. The CRM will be monitored at regular intervals and changes, which may occur, will be reflected in an updated certificate of analysis provided to the users.

#### 3.3 DETERMINATION OF ASSIGNED VALUES AND UNCERTAINTIES

The data provided by the participant laboratories was first checked for their validity based on the fully documented method which it was tested by using an appropriated CRM for PAHs or participating in interlaboratory comparisons. Values given as below limit of detection or below limit of quantification were excluded for the statistical evaluation, excepting the values of naphthalene from laboratory 3 and anthracene from laboratories 1 and 3, due to the close values to the other reported values. Also, all data set of Acenapthene and Dibenzo[a,h]anthracene were retained due to the consistent dataset. Based on this criterion, the following datasets were rejected:

Lab. 1: 1-Methylnaphthalene, 2-Methylnaphthalene and Naphthalene

#### Lab 2: 1-Methylnaphthalenes, 2-Methylnaphthalene

Most of the participating laboratories applying the GC used the common 5% phenylmethylpolysiloxane phase where several important PAH isomers are not completely resolved, i.e. Chrysene and Triphenylene, Benzo[b]fluoranthene and Benzo[*j*]fluoranthene. Therefore, for those laboratories that achieved the separation of these isomers by using a 50% phenylmehtylpolysiloxane or a Select PAH column (lab 1 and lab 6), or using a brand new 5% phenylmehtylpolysiloxane (lab 7), their individual values were combined and included in the

final data set for the statistical evaluation as Chrysene (+Triphenylene) and Benzo[b+j] fluoranthene.

The characterization datasets accepted based on the technical reasons resulted in 2 to 12 measurement results for the target PAH compounds. As each participant used different extraction techniques/solvents followed by their own fractionation procedures and GC or HPLC separation, the data set is expected to provide a certain scatter. Therefore, different statistical tests were carried out to confirm that the single values provided by the participant laboratories follow a common mother distribution and are compliant with the certification requirements. All accepted sets of results were submitted to the following statistical tests: Grubbs and Dixon's test to detect outliers with respect to the mean and Kolmogorov-SmirNov's test for the normality test. All data sets were normally distributed, and outliers were found for Acenapthylene (lab 6 and 7), C2-Phen/Anth (lab 10 and 13), C3-Chrysenes (lab. 13), Benzo[b+j]fluoranthene (lab 6 and 10) and Dibenzo[a,h]anthracene (lab 6 and 7). All data were retained for statistical analysis, except for the 2 outliers of Benzo[b+j]fluoranthene and the outlier at 99% of acenapthylene.

The medians, unweighted mean of the means and robust mean were calculated and compared (Table 4). No significant differences were observed for the major PAHs compounds and the reference values obtained with the unweighted mean of the means was considered the most reliable estimates of the property values of the selected PAHs.

#### TABLE 4. COMPARISON OF DIFFERENT MEANS

	No. Results	Mean of the	Median	Robust		
	accepted	means µg Kg <sup>-1</sup>	μg Kg <sup>-1</sup>	mean µg Kg <sup>-1</sup>	outliers 95%	outliers 99%
Naphthalene	9	4.50	4.34	4.42	0	0
2-Methylnaphthalene	5	2.42	2.50	2.44	0	0
1-Methylnaphthalene	5	1.41	1.44	1.41	0	0
C2- Naphthalenes	6	8.03	6.67	7.66	0	0
C3- Naphthalenes	6	5.16	5.19	5.05	0	0
Biphenyl	3	2.00	2.10	2.00	0	0
Acenaphthylene	9	0.71	0.71	0.71	1	1
Fluorene	10	0.80	0.64	0.68	0	0
Acenaphthene	8	0.33	0.24	0.26	0	0
C1-Fluorenes	3	2.74	2.20	2.28	0	0
C2-Fluorenes	3	3.76	3.09	3.22	0	0
C3-Fluorenes	3	4.29	5.45	5.24	0	0
Dibenzothiophene	6	0.55	0.52	0.54	0	0
C1-Dibenzothiophenes	4	2.49	2.05	2.17	0	0
C2-Dibenzothiophenes	4	7.62	6.31	7.26	0	0
C3-Dibenzothiophenes	3	7.53	6.89	7.53	0	0
Phenanthrene	12	4.55	4.38	4.45	0	0
Anthracene	10	1.31	1.29	1.26	0	0
1-Methylphenanthrene	5	1.51	1.30	1.47	0	0
2-Methylphenanthrene	6	2.41	2.47	2.40	0	0
C1- Phen/Anth	6	7.58	7.37	7.58	0	0
C2- Phen/Anth	6	8.58	8.37	8.41	2	0
C3- Phen/Anth	6	4.91	4.62	4.91	0	0
C4- Phen/Anth	4	2.41	2.44	2.42	0	0
Fluoranthene	12	6.65	6.22	6.61	0	0
Pyrene	12	6.17	5.63	6.15	0	0
1-MetylPyrene	2	0.71	0.71		0	0
C1-Fluoranthenes/pyrenes	5	4.42	4.40	4.45	0	0
C2-Fluoranthenes/pyrenes	4	3.49	3.13	3.35	0	0
C3-Fluoranthenes/pyrenes	2	2.10	2.10		0	0
Benz[a]anthracene	12	2.43	2.29	2.40	0	0
Chrysene	4	4.00	3.37	3.66	0	0
Chrysene (+ Triphenylene)	10	3.53	3.21	3.37	0	0
Triphenylene	2	0.81	0.81		0	0
C1-Chrysenes	5	4.13	4.60	4.28	0	0
C2-Chrysenes	5	3.52	3.90	3.52	0	0
C3-Chrysenes	3	3.18	1.55	1.58	1	1
Benzo[b]fluoranthene	4	5.76	4.43	4.75	0	0
Benzo[j]fluoranthene	3	2.16	1.59	1.87	0	0
Benzo[ $b+i$ ]fluoranthene	9	5.00	4.68	4.98	2	1

10010 (00110)						
	No.	Mean	Median	Robust		
	Results	of the mea	ns	mean	outliers	outliers
		µg Kg-1	μg Kg <sup>-1</sup>	μg Kg <sup>-1</sup>	95%	99%
Benzo[k]fluoranthene	10	2.09	1.75	2.04	0	0
Benzo[a]fluoranthene	2	0.70	0.70		0	0
Benzo[e]pyrene	9	3.95	2.95	3.22	0	0
Benzo[a]pyrene	12	2.87	2.81	2.80	0	0
Indeno[1,2,3-c,d]pyrene	12	2.65	2.51	2.65	0	0
Dibenz[a,h]anthracene	9	0.43	0.43	0.42	2	0
Benzo[g,h,i]perylene	11	2.98	3.04	2.98	0	0
Perylene	9	35.64	31.57	33.85	0	0

Table 4 (cont)

The uncertainties associated with the assigned property values were evaluated according to ISO Guide 35 [1]. The relative combined uncertainty of the assigned property value of the CRM involved combining the standard uncertainties associated with the characterization  $(u_{char})$ , homogeneity  $(u_{hom})$ , short term stability  $(u_{short})$  and long-term stability  $(u_{stab})$ . Because the uncertainty component derived from the short-term stability was insignificant and assumed to be zero, the final expanded uncertainty was a combination of the other three different contributions using the law of propagation of uncertainty as shown in Eq. (5):

$$U = k \times \sqrt{u_{char}^2 + u_{stab}^2 + u_{hom}^2}$$
(5)

Where k is the coverage factor of 2, representing a confidence level of 95%  $u_{hom}$  was evaluated as described in section 3.1.1, and set at 8% for all PAH analytes  $u_{stab}$  was evaluated as described in section 3.2.1 and set at 3% for all PAH analytes.  $u_{char}$  was evaluated as described in ISO 35 [1] using Eq. (6):

$$u_{char} = \frac{s}{\sqrt{p}} \tag{6}$$

Where: *s* is the standard deviation and p is the number of data sets accepted.

The final assigned values derived by the mean of the means are shown in Table 5 together with their individual uncertainty associated to the characterization and final uncertainty budget.

		Mean of the		U total
	No. Results	means	$u_{ m char}$	(k=2)
		µg Kg⁻¹	%	%
Naphthalene	9	4.50	18	40
2-Methylnaphthalene	5	2.42	12	30
1-Methylnaphthalene	5	1.41	13	32
C2- Naphthalenes	6	8.03	23	50
C3- Naphthalenes	6	5.16	20	44
Biphenyl	3	2.00	16	36
Acenaphthylene	9	0.71	14	33
Fluorene	10	0.80	18	40
Acenaphthene	8	0.33	31	64
C1-Fluorenes	3	2.74	24	50
C2-Fluorenes	3	3.76	26	55
C3-Fluorenes	3	4.29	33	68
Dibenzothiophene	6	0.55	13	32
C1-Dibenzothiophenes	4	2.49	27	56
C2-Dibenzothiophenes	4	7.62	37	77
C3-Dibenzothiophenes	3	7.53	23	48
Phenanthrene	12	4.55	12	29
Anthracene	10	1.31	16	36
1-Methylphenanthrene	5	1.51	18	39
2-Methylphenanthrene	6	2.41	15	35
C1- Phen/Anth	6	7.58	13	30
C2- Phen/Anth	6	8.58	16	37
C3- Phen/Anth	6	4.91	21	46
C4- Phen/Anth	4	2.41	10	27
Fluoranthene	12	6.65	9.2	25
Pvrene	12	6.17	9.4	25
1-MetvlPvrene	2	0.71		
C1-Fluoranthenes/pyrenes	5	4.42	23	49
C2-Fluoranthenes/pyrenes	4	3.49	33	69
C3-Fluoranthenes/pyrenes	2	2.10		
Benz[ $a$ ]anthracene	12	2.43	7.7	23
Chrysene	4	4.00	32	<b>6</b> 7
Chrysene (+ Triphenylene)	10	3.53	8.7	24
Triphenvlene	2	0.81		
C1-Chrysenes	5	4.13	13	32
C2-Chrysenes	5	3.52	19	42
C3-Chrysenes	3	3.18	52	105
Benzo[ <i>b</i> ]fluoranthene	4	5.76	32	67
Benzo[ <i>i</i> ]f]uoranthene	3	2.16	43	88
Benzo[ $b+i$ ]fluoranthene	11	5 00	83	24
Renzo[k]fluoranthene	10	2.00	11	27 28

TABLE 5. MEAN OF THE M	MEAN, CHARACTERIZATION	NUNCERTAINTY AND	O TOTAL UNCERTAINTY
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Table	5.	(cont.)
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	No. Results	Mean of the means	$u_{ m char}$	U total (k=2)
		μg Kg <sup>-1</sup>	%	%
Benzo[a]fluoranthene	2	0.70		
Benzo[e]pyrene	9	3.95	14	33
Benzo[a]pyrene	12	2.87	9.0	25
Indeno[1,2,3-c,d]pyrene	12	2.65	8.9	25
Dibenz[ <i>a</i> , <i>h</i> ]anthracene	9	0.43	9.2	25
Benzo[g,h,i]perylene	11	2.98	7.0	22
Perylene	9	35.64	22	47

The results for the mass fractions of the PAHs as reported by the participants in this characterization are presented in Appendix I and II. The laboratory means are plotted together with the mean of the means denoted by a bold line, while the dashed lines represent mean  $\pm$  expanded uncertainty (*k*=2) in all figures (as calculated in Eq. 5). The error bars represent the expanded uncertainty of the participant calculated as  $2 \ge \frac{s}{\sqrt{n}}$  where *s* is the standard deviation and *n* is the number of measurements reported by the participant. In this context, it should be noted that this expanded uncertainty as a measurement of the experimental standard deviation of the mean only reflect the repeatability of the participant method, neglecting other key contributors to the uncertainty as it is described within the GUM guide [2]. Reasonable and comparability estimates of uncertainty in organic contaminant analysis is still a priority to be achieved in accordance to the GUM guide.

As shown previously in Table 1, methods using different independent analytical techniques (GC-MS; GC-MS/MS, GC-HRMS and HPLC-FLD) with different extraction and purification procedures were used for the characterization of the material. A good agreement was observed for results obtained with different methods which confirms the absence of any significant method bias and demonstrates the identity of the analyte.

The mean of the means of the laboratory were assigned as certified values, for those compounds where the assigned value was derived from at least 5 datasets and its relative expanded uncertainties was less than 40 % of the assigned value. These criteria were fulfilled for 19 PAHs: Naphthalene, 2-Methylnaphthalene, 1-Methylnaphthalene, Phenantrene, Anthracene, 1-Methylphenanthrene, 2-Methylphenanthrene, C1-Phenanthrenes/ Anthracenes, Fluoranthene, Pyrene, Benz[*a*]anthracene, Chrysene (+Triphenylene), C1-Chrysenes, Benzo[k]fluoranthene, Benzo[b+j]fluoranthene; Benzo[*e*]pyrene, Benzo[*a*]pyrene, Indeno[1,2,3-c,d]pyrene, Benzo[g,h,i]perylene. The certified values are presented in Table 6, together with their expanded uncertainty.

Compounds that did not fulfill the criteria of certification are considered information values. They include C2-Naphthalenes, C3-Naphthalenes, Biphenyl, Acenaphthylene, Fluorene,

Acenaphthene, C1-Fluorenes, C2-Fluorenes, C3-Fluorenes, Dibenzothiophene, C1-Dibenzothiophenes, C2-Dibenzothiophenes, C3-Dibenzothiophenes, C2-Phen/Anth, C3-Phen/Anth, C4-Phen/Anth, 1-Metylpyrene, C1-Fluoranthenes/pyrenes, C2-C2-Chrysenes, C3-Fluoranthenes/pyrenes, Triphenylene, Fluoranthenes/pyrenes, C3-Benzo[*b*]fluoranthene, Benzo[*j*/fluoranthene, Benzo[*a*]fluoranthene, Chrysenes, Dibenz[a,h]anthracene, Perylene. Table 7, shows the information values for PAHs together with the expanded uncertainty for the compounds that could be calculated.

Compound	Unit	Certified value <sup>1</sup>	$U(k=2)^{2}$
Naphthalene	μg kg <sup>-1</sup>	4.5	1.8
2-Methylnaphthalene	µg kg⁻¹	2.4	0.7
1-Methylnaphthalene	μg kg <sup>-1</sup>	1.4	0.5
Phenanthrene	μg kg <sup>-1</sup>	4.6	1.3
Anthracene	μg kg <sup>-1</sup>	1.3	0.5
1-Methylphenanthrene	μg kg <sup>-1</sup>	1.5	0.6
2-Methylphenanthrene	μg kg <sup>-1</sup>	2.4	0.9
C1- Phen/Anth	μg kg <sup>-1</sup>	7.6	2.3
Fluoranthene	μg kg <sup>-1</sup>	6.6	1.7
Pyrene	μg kg <sup>-1</sup>	6.2	1.6
Benz[a]anthracene	μg kg <sup>-1</sup>	2.4	0.6
Chrysene (+ Triphenylene)	μg kg <sup>-1</sup>	3.5	0.9
C1-Chrysenes	μg kg <sup>-1</sup>	4.1	1.3
Benzo[ <i>b</i> + <i>j</i> ]fluoranthene	μg kg <sup>-1</sup>	5.0	1.2
Benzo[k]fluoranthene	μg kg <sup>-1</sup>	2.1	0.6
Benzo[e]pyrene	μg kg <sup>-1</sup>	3.9	1.3
Benzo[a]pyrene	μg kg <sup>-1</sup>	2.9	0.7
Indeno[1,2,3-c,d]pyrene	μg kg <sup>-1</sup>	2.6	0.7
Benzo[g,h,i]perylene	μg kg <sup>-1</sup>	3.0	0.7

TABLE 6. CERTIFED VALUES FOR PAHS MASS FRACTIONS AND THEIR EXPANDED UNCERTAINTY (k=2) IN THE IAEA-477 SEDIMENT SAMPLE

<sup>1</sup> The value is the mean of the mean of the accepted sets of data, each set being obtained by different laboratory. The certified values are reported on dry mass basis and are traceable to the SI.

<sup>2</sup> Expanded uncertainty evaluated according to ISO Guide 35 [1] with a coverage factor k=2 estimated in accordance with the JCGM 100:2008 Evaluation of measurement data – Guide to the expression of uncertainty in measurement [2], corresponding to the level of confidence of about 95%.

Compound	Unit	Information value <sup>1</sup>	U ( <i>k</i> =2) <sup>2</sup>
C2- Naphthalenes	µg kg-1	8.0	4.0
C3- Naphthalenes	μg kg <sup>-1</sup>	5.2	2.3
Biphenyl	μg kg <sup>-1</sup>	2.0	0.7
Acenaphthylene	μg kg <sup>-1</sup>	0.7	0.3
Fluorene	μg kg-1	0.8	0.3
Acenaphthene	μg kg-1	0.3	0.2
C1-Fluorenes	μg kg-1	2.7	1.4
C2-Fluorenes	μg kg-1	3.8	2.1
C3-Fluorenes	μg kg <sup>-1</sup>	4.3	2.9
Dibenzothiophene	µg kg⁻¹	0.6	0.2
C1-Dibenzothiophenes	µg kg⁻¹	2.5	1.4
C2-Dibenzothiophenes	µg kg⁻¹	7.6	5.9
C3-Dibenzothiophenes	μg kg-1	7.5	3.7
C2- Phen/Anth	μg kg <sup>-1</sup>	8.6	3.2
C3- Phen/Anth	μg kg <sup>-1</sup>	4.9	2.3
C4- Phen/Anth	μg kg <sup>-1</sup>	2.4	0.6
1-Metylpyrene	μg kg <sup>-1</sup>	0.7	
C1-Fluoranthenes/pyrenes	μg kg <sup>-1</sup>	4.4	2.2
C2-Fluoranthenes/pyrenes	μg kg <sup>-1</sup>	3.5	2.4
C3-Fluoranthenes/pyrenes	μg kg <sup>-1</sup>	2.1	
Chrysene	μg kg <sup>-1</sup>	4.0	2.7
Triphenylene	μg kg <sup>-1</sup>	0.8	
C2-Chrysenes	μg kg <sup>-1</sup>	3.5	1.5
C3-Chrysenes	μg kg <sup>-1</sup>	1.6	
Benzo[b]fluoranthene	μg kg <sup>-1</sup>	5.8	3.8
Benzo[j]fluoranthene	μg kg <sup>-1</sup>	2.2	1.9
Benzo[a]fluoranthene	μg kg <sup>-1</sup>	0.7	
Dibenz[ <i>a</i> , <i>h</i> ]anthracene	μg kg <sup>-1</sup>	0.4	0.1
Perylene	μg kg <sup>-1</sup>	35.6	16.9

TABLE 7. INFORMATION VALUES FOR PAHs MASS FRACTIONS AND THEIR EXPANDED UNCERTAINTY (k=2) IN THE IAEA-477 SEDIMENT SAMPLE

<sup>1</sup> The value is the mean of the mean of the accepted sets of data, except for C3-Chrysenes where the value is the robust mean. Each set of data is being obtained by different laboratory. The information values are reported on dry mass basis and are traceable to the SI.

<sup>2</sup> Expanded uncertainty evaluated according to ISO Guide 35 [1] with a coverage factor k=2 estimated in accordance with the JCGM 100:2008 Evaluation of measurement data – Guide to the expression of uncertainty in measurement [2], corresponding to the level of confidence of about 95%.

#### 4. METROLOGICAL TRACEABILITY AND COMMUTABILITY

Metrological traceability is defined as the property of a measurement result where the result is related to a reference through a documented unbroken chain of calibrations, each contributing to the measurement uncertainty [9]. The information on the calibration standards, surrogate standards and matrix standard reference materials (CRMs) standards are summarized in Appendix III. The methods used by all participating laboratories were validated by using matrix standard reference materials (CRMs) from NIST (SRM1941b, SRM1944), IAEA (IAEA-408, IAEA-459, IAEA-383) and materials characterized by QUASIMEME proficiency tests (QPH094MS). Their reported values are based on calibration standard solutions of known purity, issued by accredited commercial companies with documented unbroken chain of calibrations. Consequently, the assigned values derived from combining the individual results are traceable to International System of Units (SI). Furthermore, the agreement between the results generated by different analytical methodologies ensures the comparability of the measurement results and validates the identity of the measurand.

Commutability is a property of an RM, demonstrated by the closeness of agreement between the relation among the measurement results for a stated quantity in this material, obtained according to two given measurement procedures, and the relation obtained among the measurement results for other specified materials [10].

A material is said to be commutable when the measurand in the routine test samples behaves similar as in the CRM with respect to the different measurement procedures. This implies that the application of the procedures to the CRM would produce the same quantitative value as normal routine samples containing the same concentration of the measurand. In this respect, IAEA-477 is a natural marine sediment sample and their analytical behaviour should be the same as for a routine sample of dried sediment samples. The agreement between the data acquired with different analytical procedures for the IAEA-477 characterization study endorses the absence of any significant method bias and shows commutability of the material for all certified organic compounds.

#### **5. CONCLUSIONS**

The combination of different data sets from at least two different analytical techniques has allowed the assignment of certified concentrations for 19 PAHs following the recommendation of ISO Guide 35. The extensive characterization with very low concentration levels and associated uncertainties will make CRM 477 a valuable sediment reference material for use in the validation of analytical methods for the determination of polycyclic aromatic hydrocarbons included as priority substances (PSs), within the environmental monitoring programmes.

# APPENDIX I: CHARACTERIZATION RESULTS: ASSIGNED VALUES OF PAHS

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
2	2.93	0.23	0.500	GC/MSMS	IAEA-408
3	1.23	0.22	0.500	HPLC-FLD	QPH094MS
4	4.95	0.19	0.950	GC/MS	NIST 1941b
5	1.97	0.05	0.100	GC/MS	IAEA 459
6	6.53	0.64	0.500	GC/MS	NIST 1941b
9	4.34	0.40	0.128	GC/MSMS	IAEA-459
10	3.25	0.18	0.160	GC/MS	NIST 1941b
11	8.82	0.79	2.000	GC/MS	IAEA-383
13	6.49	0.50	0.050	GC/MS	IAEA-459
		Re	esults not used for	the assignment value	
1	0.87	0.09	5.000	GC/MS	NIST 1941b
7	<4.9		7.000	GC/HRMS	IAEA-459

TABLE 8. NAPHTHALENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ ):



FIG. 3 Laboratory results used to calculate the assignment mass fraction of Naphthalene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
4	2.74	0.15	0.110	GC/MS	NIST 1941b
5	1.41	0.07	0.050	GC/MS	IAEA 459
6	2.28	0.28	0.500	GC/MS	NIST 1941b
10	2.50	0.30	0.080	GC/MS	NIST 1941b
13	3.18	0.39	0.015	GC/MS	IAEA-459
		R	esults not used for	the assignment value	
1	0.55	0.07	0.700	GC/MS	NIST 1941b
2	0.83	0.07	0.170	GC/MSMS	IAEA-408
7	<3.9		0.130	GC/HRMS	IAEA-459



FIG. 4. Laboratory results used to calculate the assignment mass fraction of 2-Methylnaphthalene in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 10. 1-	-METHYLNAP	HTHALENE RESU	LTS REPORTED	BY PARTICIE	ANTS (µg kg <sup>-1</sup>	)
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
4	1.44	0.14	0.140	GC/MS	NIST 1941b
5	0.83	0.07	0.050	GC/MS	IAEA 459
6	1.17	0.21	0.500	GC/MS	NIST 1941b
10	1.92	0.16	0.060	GC/MS	IAEA-459
13	1.69	0.19	0.014	GC/MS	NIST 1941b
		Re	sults not used for t	he assignment value	
1	0.27	0.04	0.700	GC/MS	NIST 1941b
2	0.36	0.06	0.100	GC/MSMS	IAEA-408
7	<1.9		0.130	GC/HRMS	IAEA-459



FIG. 5. Laboratory results used to calculate the assignment mass fraction of 1-Methylnaphthalene in IAEA-477 ( $\mu g k g^{-1}$ ).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	2.85	0.14	0.700	GC/MS	NIST 1941b
2	5.41	0.10	0.100	GC/MSMS	IAEA-408
3	2.57	0.17	0.500	HPLC-FLD	QPH094MS
4	2.63	0.07	0.430	GC/MS	NIST 1941b
5	3.97	0.13	0.050	GC/MS	IAEA 459
6	8.37	0.64	0.500	GC/MS	NIST 1941b
7	2.72	0.06	0.020	GC/HRMS	IAEA-459
9	6.64	0.23	0.018	GC/MSMS	IAEA-459
10	4.80	0.05	0.090	GC/MS	NIST 1941b
11	3.18	0.42	0.400	GC/MS	IAEA-383
12	5.41	0.38	0.800	GC/MS	IAEA-383
13	6.04	0.42	0.014	GC/MS	IAEA-459

### TABLE 11. PHENANTHRENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )



FIG. 6. Laboratory results used to calculate the assignment mass fraction of Phenanthrene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	0.72	0.05	0.700	GC/MS	NIST 1941b
2	1.41	0.04	0.120	GC/MSMS	IAEA-408
3	0.83	0.03	0.500	HPLC-FLD	QPH094MS
5	0.79	0.03	0.050	GC/MS	IAEA 459
6	2.68	0.18	0.500	GC/MS	NIST 1941b
7	0.56	0.04	0.020	GC/HRMS	IAEA-459
9	1.89	0.10	0.020	GC/MSMS	IAEA-459
10	1.28	0.08	0.090	GC/MS	NIST 1941b
11	1.67	0.16	0.400	GC/MS	IAEA-383
13	1.30	0.22	0.030	GC/MS	IAEA-459
		Re	sults not used for t	the assignment value	
4	<loq< td=""><td></td><td>0.280</td><td>GC/MS</td><td>NIST 1941b</td></loq<>		0.280	GC/MS	NIST 1941b
12	<1.44		0.800	GC/MS	IAEA-383
6 7 9 10 11 13 4 12	2.68 0.56 1.89 1.28 1.67 1.30 <loq &lt;1.44</loq 	0.18 0.04 0.10 0.08 0.16 0.22 Re	0.500 0.020 0.090 0.400 0.030 sults not used for 1 0.280 0.800	GC/MS GC/HRMS GC/MSMS GC/MS GC/MS GC/MS the assignment value GC/MS GC/MS	NIST 1941b IAEA-459 IAEA-459 NIST 1941b IAEA-383 IAEA-459 NIST 1941b IAEA-383

### TABLE 12. ANTHRACENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-}1})$



FIG. 7. Laboratory results used to calculate the assignment mass fraction of Anthracene in IAEA-477 (µg kg<sup>-1</sup>).

TABLE 13. 1-METHYLPHENANTHRENE RESU	TS REPORTED BY PARTICIPANTS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	1.14	0.05	0.700	GC/MS	NIST 1941b
4	1.30	0.10	0.290	GC/MS	NIST 1941b
6	2.37	0.08	0.500	GC/MS	NIST 1941b
7	0.89	0.05	0.020	GC/HRMS	IAEA-459
13	1.85	0.06	0.014	GC/MS	IAEA-459



*FIG. 8. Laboratory results used to calculate the assignment mass fraction of 1-Methylphenanthrene in IAEA-477* ( $\mu g k g^{-1}$ ).
TABLE 14. 2-METHYLPHENANTHRENE RESULTS REPORTED BY PARTICIPANTS (µg kg <sup>-1</sup>	')
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	1.46	0.05	0.700	GC/MS	NIST 1941b
2	2.91	0.21	0.450	GC/MSMS	IAEA-408
4	2.06	0.15	0.110	GC/MS	NIST 1941b
6	3.70	0.31	0.500	GC/MS	NIST 1941b
7	1.43	0.07		GC/HRMS	IAEA-459
13	2.87	0.14	0.014	GC/MS	IAEA-459



*FIG. 9. Laboratory results used to calculate the assignment mass fraction of* 2-Methylphenanthrene *in IAEA-477* ( $\mu$ g kg<sup>-1</sup>).

TABLE 15. C1	- PHEN/ANTH	RESULTS RI	EPORTED BY	PARTICIPAN	TS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	5.23	0.21	0.700	GC/MS	NIST 1941b
2	7.50	0.49	0.450	GC/MSMS	IAEA-408
4	7.24	0.13	0.910	GC/MS	NIST 1941b
7	4.93	0.17	0.080	GC/HRMS	IAEA-459
10	10.50	0.70	0.110	GC/MS	NIST 1941b
13	10.08	0.47	0.014	GC/MS	IAEA-459



FIG. 10. Laboratory results used to calculate the assignment mass fraction of C1-Phen/Anth in IAEA-477 ( $\mu g k g^{-1}$ ).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	4.95	0.21	0.300	GC/MS	NIST 1941b
2	7.88	0.14	0.110	GC/MSMS	IAEA-408
3	5.10	0.21	0.500	HPLC-FLD	QPH094MS
4	3.65	0.18	0.640	GC/MS	NIST 1941b
5	5.21	0.09	0.050	GC/MS	IAEA 459
6	10.25	0.88	0.500	GC/MS	NIST 1941b
7	4.72	0.11	0.010	GC/HRMS	IAEA-459
9	9.63	0.47	0.006	GC/MSMS	IAEA-459
10	7.23	0.10	0.080	GC/MS	NIST 1941b
11	7.88	0.44	0.400	GC/MS	IAEA-383
12	5.20	0.30	0.740	GC/MS	IAEA-383
13	8.16	0.32	0.024	GC/MS	IAEA-459

# TABLE 16. FLUORANTHENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-}1})$



FIG. 11. Laboratory results used to calculate the assignment mass fraction of Fluoranthene in IAEA-477 ( $\mu g k g^{-1}$ ).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	4.58	0.18	0.300	GC/MS	NIST 1941b
2	6.27	0.20	0.210	GC/MSMS	IAEA-408
3	4.47	0.20	0.500	HPLC-FLD	QPH094MS
4	3.94	0.17	0.460	GC/MS	NIST 1941b
5	4.28	0.16	0.050	GC/MS	IAEA 459
6	9.52	1.07	0.500	GC/MS	NIST 1941b
7	4.35	0.12	0.010	GC/HRMS	IAEA-459
9	8.98	0.48	0.006	GC/MSMS	IAEA-459
10	6.93	0.10	0.080	GC/MS	NIST 1941b
11	7.65	0.59	0.400	GC/MS	IAEA-383
12	5.00	0.29	0.950	GC/MS	IAEA-383
13	8.01	0.23	0.023	GC/MS	IAEA-459

# TABLE 17. PYRENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )



FIG. 12. Laboratory results used to calculate the assignment mass fraction of Pyrene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	1.81	0.07	0.300	GC/MS	NIST 1941b
2	2.39	0.07	0.310	GC/MSMS	IAEA-408
3	1.47	0.10	0.500	HPLC-FLD	QPH094MS
4	1.83	0.11	0.430	GC/MS	NIST 1941b
5	2.13	0.11	0.100	GC/MS	IAEA 459
6	3.25	0.33	0.500	GC/MS	NIST 1941b
7	2.02	0.08	0.020	GC/HRMS	IAEA-459
9	2.74	0.09	0.008	GC/MSMS	IAEA-459
10	3.47	0.10	0.100	GC/MS	NIST 1941b
11	3.32	0.24	0.400	GC/MS	IAEA-383
12	2.18	0.16	0.230	GC/MS	IAEA-383
13	2.50	0.21	0.020	GC/MS	IAEA-459

# TABLE 18. BENZ[A]ANTHRACENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )



FIG. 13. Laboratory results used to calculate the assignment mass fraction of Benz(a) anthracene in IAEA-477 (µg kg<sup>-1</sup>).

TABLE 19. CHRYSENE (-	+ TRIPHENYLENE	) RESULTS REPORTED	BY PARTICIPAN	√TS (µg	kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	5.07	0.18	0.300	GC/MS	NIST 1941b
2	3.72	0.12	0.080	GC/MSMS	IAEA-408
4	2.66	0.17	0.440	GC/MS	NIST 1941b
5	2.61	0.14	0.100	GC/MS	IAEA 459
7	3.29	0.12	0.020	GC-HRMS	IAEA-459
9	3.12	0.14	0.010	GC/MSMS	IAEA-459
10	5.15	0.07	0.090	GC/MS	NIST 1941b
11	2.65	0.18	0.400	GC/MS	IAEA-383
12	2.90	0.18	1.260	GC/MS	IAEA-383
13	4.16	0.12	0.011	GC/MS	IAEA-459



FIG. 14. Laboratory results used to calculate the assignment mass fraction of Chrysene (+ Triphenylene) in IAEA-477 ( $\mu$ g kg<sup>-1</sup>).

TABLE 20. C1-	-CHRYSENES	RESULTS	REPORTED	BY PARTIC	CIPANTS (	(µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	3.35	0.16	0.300	GC/MS	NIST 1941b
4	2.35	0.15	0.170	GC/MS	NIST 1941b
7	5.27	0.37	0.360	GC/HRMS	IAEA-459
10	5.10	0.30	0.120	GC/MS	NIST 1941b
13	4.60	0.35	0.011	GC/MS	IAEA-459



FIG. 15. Laboratory results used to calculate the assignment mass fraction of C1-Chrysenes in IAEA-477 ( $\mu$ g kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	4.68	0.16	0.700	GC/MS	NIST 1941b
2	6.60	0.15	0.310	GC/MSMS	IAEA-408
4	3.79	0.22	0.520	GC/MS	NIST 1941b
5	3.52	0.21	0.100	GC/MS	IAEA 459
7	4.32	0.14	0.010	GC/HRMS	IAEA-459
9	6.78	0.07	0.016	GC/MSMS	IAEA-459
11	4.10	0.21	0.400	GC/MS	IAEA-383
12	6.28	0.48	2.110	GC/MS	IAEA-383
13	4.93	0.80	0.018	GC/MS	IAEA-459
		Re	sults not used for t	the assignment value	
6	15.10	1.70	0.500	GC/MS	NIST 1941b
10	10.35	0.27	0.190	GC/MS	NIST 1941b

#### TABLE 21. BENZO[B+J] FLUORANTHENE RESULTS REPORTED BY PARTICIPANTS (µg kg<sup>-1</sup>)



FIG. 16. Laboratory results used to calculate the assignment mass fraction of Benzo(b+j) fluoranthene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC			
1	1.40	0.12	0.700	GC/MS	NIST 1941b			
2	1.80	0.06	0.300	GC/MSMS	IAEA-408			
3	1.15	0.09	0.500	HPLC-FLD	QPH094MS			
5	2.86	0.16	0.100	GC/MS	IAEA 459			
6	3.15	0.31	0.500	GC/MS	NIST 1941b			
7	1.70	0.05	0.010	GC/HRMS	IAEA-459			
9	1.67	0.06	0.010	GC/MSMS	IAEA-459			
10	2.97	0.12	0.180	GC/MS	NIST 1941b			
11	1.47	0.13	0.400	GC/MS	IAEA-383			
13	2.78	0.42	0.016	GC/MS	IAEA-459			
	Results not used for the assignment value							
4	<loq< td=""><td></td><td>0.770</td><td>GC/MS</td><td>NIST 1941b</td></loq<>		0.770	GC/MS	NIST 1941b			

#### TABLE 22. BENZO[K]FLUORANTHENE RESULTS REPORTED BY PARTICIPANTS (µg kg-1)



FIG. 17. Laboratory results used to calculate the assignment mass fraction of Benzo(k) fluoranthene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	2.56	0.08	0.300	GC/MS	NIST 1941b
2	3.87	0.07	0.350	GC/MSMS	IAEA-408
4	2.95	0.15	0.460	GC/MS	NIST 1941b
6	6.98	0.42	0.500	GC/MS	NIST 1941b
7	2.80	0.12	0.010	GC/HRMS	IAEA-459
10	6.13	0.18	0.210	GC/MS	NIST 1941b
11	2.75	0.37	0.400	GC/MS	IAEA-383
12	2.64	0.19	0.440	GC/MS	IAEA-383
13	4.82	0.52	0.012	GC/MS	IAEA-459

#### TABLE 23. BENZO[E]PYRENE RESULTS REPORTED BY PARTICIPANTS (µg kg<sup>-1</sup>)



FIG. 18. Laboratory results used to calculate the assignment mass fraction of Benzo(e) pyrene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	2.39	0.10	0.700	GC/MS	NIST 1941b
2	3.13	0.05	0.350	GC/MSMS	IAEA-408
3	1.89	0.09	0.500	HPLC-FLD	QPH094MS
4	1.62	0.06	0.340	GC/MS	NIST 1941b
5	2.62	0.07	0.100	GC/MS	IAEA 459
6	4.45	0.41	0.500	GC/MS	NIST 1941b
7	2.55	0.10	0.010	GC/HRMS	IAEA-459
9	3.12	0.05	0.013	GC/MSMS	IAEA-459
10	4.32	0.11	0.230	GC/MS	NIST 1941b
11	1.92	0.14	0.400	GC/MS	IAEA-383
12	3.00	0.19	0.880	GC/MS	IAEA-383
13	3.42	0.64	0.019	GC/MS	IAEA-459

# TABLE 24. BENZO[A]PYRENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )



FIG. 19. Laboratory results used to calculate the assignment mass fraction of Benzo(a) pyrene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	2.43	0.11	0.700	GC/MS	NIST 1941b
2	3.32	0.07	0.330	GC/MSMS	IAEA-408
3	1.77	0.17	0.500	HPLC-FLD	QPH094MS
4	1.66	0.01	0.510	GC/MS	NIST 1941b
5	3.11	0.21	0.100	GC/MS	IAEA 459
6	3.77	0.48	0.500	GC/MS	NIST 1941b
7	3.12	0.17	0.020	GC/HRMS	IAEA-459
9	2.54	0.06	0.007	GC/MSMS	IAEA-459
10	2.48	0.10	0.150	GC/MS	NIST 1941b
11	1.67	0.25	0.400	GC/MS	IAEA-383
12	1.90	0.20	0.730	GC/MS	IAEA-383
13	4.02	0.51	0.038	GC/MS	IAEA-459

# TABLE 25. INDENO[1,2,3-C,D]PYRENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )



FIG. 20. Laboratory results used to calculate the assignment mass fraction of Indeno[1,2,3-c,d]pyrene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC		
1	2.69	0.12	0.700	GC/MS	NIST 1941b		
2	4.12	0.05	0.050	GC/MSMS	IAEA-408		
3	2.44	0.09	0.500	HPLC-FLD	QPH094MS		
4	1.89	0.08	0.350	GC/MS	NIST 1941b		
5	3.04	0.25	0.100	GC/MS	IAEA 459		
6	2.58	0.54	0.500	GC/MS	NIST 1941b		
7	3.38	0.19	0.020	GC/HRMS	IAEA-459		
9	3.05	0.13	0.006	GC/MSMS	IAEA-459		
10	3.58	0.06	0.260	GC/MS	NIST 1941b		
11	2.22	0.31	0.400	GC/MS	IAEA-383		
13	3.81	0.51	0.070	GC/MS	IAEA-459		
	Results not used for the assignment value						
12	<1.40		0.820	GC/MS	IAEA-383		

# TABLE 26. BENZO(G,H,I)PERYLENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )



FIG. 21. Laboratory results used to calculate the assignment mass fraction of Benzo(g,h,i) perylene in IAEA-477 (µg kg<sup>-1</sup>).

# APPENDIX II: CHARACTERIZATION RESULTS: INFORMATION VALUES OF PAHS

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	2.82	0.48	0.700	GC/MS	NIST 1941b
2	5.31	0.28	0.070	GC/MSMS	IAEA-408
4	8.03	0.13	0.120	GC/MS	NIST 1941b
7	12.0	1.47	0.350	GC/HRMS	IAEA-459
10	5.17	0.11	0.120	GC/MS	NIST 1941b
13	14.8	1.81	0.017	GC/MS	IAEA-459

#### TABLE 27. C2- NAPHTHALENES RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )



FIG. 22. Laboratory results used to calculate the information mass fraction of C2- Naphthalenes in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 28. C3- NAPHTHALENES RESULTS	S REPORTED BY PARTICIPANTS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	2.82	0.63	0.700	GC/MS	NIST 1941b
2	2.47	0.30	0.250	GC/MSMS	IAEA-408
4	5.62	0.29	0.120	GC/MS	NIST 1941b
7	4.77	0.52	0.380	GC/HRMS	IAEA-459
10	5.95	0.66	0.100	GC/MS	NIST 1941b
13	9.31	1.84	0.017	GC/MS	IAEA-459



FIG. 23. Laboratory results used to calculate the information mass fraction of C3- Naphthalenes in IAEA-477 ( $\mu$ g kg<sup>-1</sup>).

# TABLE 29. BIPHENYL RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-1}})$

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
6	2.48	0.18	0.500	GC/MS	NIST 1941b
7	1.42	0.06	0.130	GC/HRMS	IAEA-459
13	2.10	0.25	0.026	GC/MS	IAEA-459



FIG. 24. Laboratory results used to calculate the information mass fraction of Biphenyl in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC		
1	0.41	0.03	0.700	GC/MS	NIST 1941b		
2	0.38	0.05	0.170	GC/MSMS	IAEA-408		
5	0.47	0.01	0.050	GC/MS	IAEA 459		
6	1.19	0.15	0.500	GC/MS	NIST 1941b		
9	0.62	0.07	0.024	GC/MSMS	IAEA-459		
10	0.80	0.16	0.130	GC/MS	NIST 1941b		
11	0.82	0.13	0.400	GC/MS	IAEA-383		
13	0.96	0.08	0.022	GC/MS	IAEA-459		
Results not used for the assignment value							
4	<loq< td=""><td></td><td>0.630</td><td>GC/MS</td><td>NIST 1941b</td></loq<>		0.630	GC/MS	NIST 1941b		
7	4.77	0.52	0.130	GC/HRMS	IAEA-459		

# TABLE 30. ACENAPHTHYLENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-1}}$ )



*FIG. 25. Laboratory results used to calculate the information mass fraction of Acenaphthylene in IAEA-477* ( $\mu$ g kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC			
1	0.52	0.03	0.700	GC/MS	NIST 1941b			
2	0.69	0.05	0.180	GC/MSMS	IAEA-408			
3	0.26	0.07	0.500	HPLC-FLD	QPH094MS			
5	0.51	0.01	0.050	GC/MS	IAEA 459			
6	1.62	0.32	0.500	GC/MS	NIST 1941b			
7	0.39	0.01	0.110	GC/HRMS	IAEA-459			
9	1.43	0.07	0.037	GC/MSMS	IAEA-459			
10	0.58	0.03	0.110	GC/MS	NIST 1941b			
11	1.13	0.26	0.400	GC/MS	IAEA-383			
13	0.89	0.10	0.039	GC/MS	IAEA-459			
	Results not used for the assignment value							
4	<loq< td=""><td></td><td>0.460</td><td>GC/MS</td><td>NIST 1941b</td></loq<>		0.460	GC/MS	NIST 1941b			
12	<2.15		1.300	GC/MS	IAEA-383			
	<u>&lt;2.15</u>		1.300	GC/MS	IAEA-383			

# TABLE 31. FLUORENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-1}}$ )



FIG. 26. Laboratory results used to calculate the information mass fraction of Fluorene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	0.11	0.02	0.700	GC/MS	NIST 1941b
2	0.09	0.02	0.210	GC/MSMS	IAEA-408
3	0.05	0.02	0.500	HPLC-FLD	QPH094MS
5	0.25	0.02	0.050	GC/MS	IAEA 459
6	0.56	0.03	0.500	GC/MS	NIST 1941b
10	0.43	0.04	0.130	GC/MS	NIST 1941b
11	0.88	0.03	0.400	GC/MS	IAEA-383
13	0.23	0.04	0.033	GC/MS	IAEA-459
		Res	sults not used for th	ne assignment value	
4	<loq< td=""><td></td><td>0.280</td><td>GC/MS</td><td>NIST 1941b</td></loq<>		0.280	GC/MS	NIST 1941b
7	<1.3		0.220	GC/HRMS	IAEA-459
9	<1.518		0.455	GC/MSMS	IAEA-459

# TABLE 32. ACENAPHTHENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-1}})$



FIG. 27. Laboratory results used to calculate the information mass fraction of Acenaphthene in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 33. C1-FLUORENES RESULTS REPORTED	BY PARTICIPANTS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
4	2.00	0.12	0.140	GC/MS	NIST 1941b
7	2.20	0.15	0.330	GC/HRMS	IAEA-459
13	4.03	0.39	0.039	GC/MS	IAEA-459



FIG. 28. Laboratory results used to calculate the information mass fraction of C1-Fluorenes in IAEA-477 ( $\mu$ g kg<sup>-1</sup>).

TABLE 34. C2-FLUORENES	RESULTS REPORTED	BY PARTICIPANTS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
4	3.09	0.19	0.140	GC/MS	NIST 1941b
7	2.77	0.08	0.650	GC/HRMS	IAEA-459
13	5.42	0.46	0.039	GC/MS	IAEA-459



FIG. 29. Laboratory results used to calculate the information mass fraction of C2-Fluorenes in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 35. C3-FLUORENES	RESULTS REPORTED	BY PARTICIPANTS (	(µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
4	5.45	0.65	0.140	GC/MS	NIST 1941b
7	1.47	0.15	0.650	GC/HRMS	IAEA-459
13	5.94	0.63	0.039	GC/MS	IAEA-459



FIG. 30. Laboratory results used to calculate the information mass fraction of C3-Fluorenes in IAEA-477 ( $\mu g k g^{-1}$ ).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
2	0.33	0.04	0.300	GC/MSMS	IAEA-408
4	0.43	0.02	0.130	GC/MS	NIST 1944
6	0.56	0.05	0.500	GC/MS	NIST 1941b
7	0.66	0.02	0.110	GC/HRMS	IAEA-459
10	0.47	0.04	0.070	GC/MS	NIST 1941b
13	0.83	0.11	0.014	GC/MS	IAEA-459

# TABLE 36. DIBENZOTHIOPHENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{\text{-}1}$ )



FIG. 31. Laboratory results used to calculate the information mass fraction of Dibenzothiophene in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 37. C1-DIBENZOTHIOPHENES RESULT	S REPORTED BY PARTICIPANTS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
4	2.41	0.09	0.240	GC/MS	NIST 1941b
7	4.38	0.22	0.270	GC/HRMS	IAEA-459
10	1.47	0.13	0.070	GC/MS	NIST 1941b
13	1.69	0.14	0.014	GC/MS	IAEA-459
	c				



FIG. 32. Laboratory results used to calculate the information mass fraction of C1-Dibenzothiophenes in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 38. C2-DIBENZOTHIOPHENES RESULT	S REPORTED BY PARTICIPANTS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
4	9.40	0.28	0.240	GC/MS	NIST 1941b
7	14.90	1.70	0.540	GC/HRMS	IAEA-459
10	2.95	0.37	0.070	GC/MS	NIST 1941b
13	3.21	0.19	0.014	GC/MS	IAEA-459
	C				



FIG. 33. Laboratory results used to calculate the information mass fraction of C2-Dibenzothiophenes in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 39. C3-DIBENZOTHIOPHENES RESULTS REPORTED BY PARTICIPANTS (μ	g kg-	<sup>1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
4	6.89	0.13	0.240	GC/MS	NIST 1941b
7	10.75	0.92	0.540	GC/HRMS	IAEA-459
13	4.94	0.21	0.014	GC/MS	IAEA-459



FIG. 34. Laboratory results used to calculate the information mass fraction of C3-Dibenzothiophenes in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 40. C2	- PHEN/ANTH	RESULTS	REPORTED	BY PARTICI	PANTS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	7.05	0.35	0.700	GC/MS	NIST 1941b
2	7.93	0.33	0.290	GC/MSMS	IAEA-408
4	9.93	0.40	0.070	GC/MS	NIST 1941b
7	8.82	1.36	0.350	GC/HRMS	IAEA-459
10	14.08	1.33	0.110	GC/MS	NIST 1941b
13	3.70	0.19	0.014	GC/MS	IAEA-459



FIG. 35. Laboratory results used to calculate the information mass fraction of C2- Phen/Anth in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 41. C3-	- PHEN/ANTH	RESULTS R	EPORTED B	BY PARTICIPAN	TS (µg k	:g <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	4.22	0.23	0.700	GC/MS	NIST 1941b
2	2.69	0.11	0.090	GC/MSMS	IAEA-408
4	8.14	0.15	0.080	GC/MS	NIST 1941b
7	5.02	0.14	0.310	GC/HRMS	IAEA-459
10	7.62	0.91	0.120	GC/MS	NIST 1941b
13	1.76	0.12	0.014	GC/MS	IAEA-459



FIG. 36. Laboratory results used to calculate the information mass fraction of C3- Phen/Anth in IAEA-477 ( $\mu$ g kg<sup>-1</sup>).

TABLE 42. C4	- PHEN/ANTH	<b>RESULTS R</b>	EPORTED E	BY PARTICIP.	ANTS	(µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	2.21	0.14	0.700	GC/MS	NIST 1941b
4	2.94	0.18	0.080	GC/MS	NIST 1941b
7	1.83	0.16	0.310	GC/HRMS	IAEA-459
13	2.68	0.08	0.014	GC/MS	IAEA-459



FIG. 37. Laboratory results used to calculate the information mass fraction of C4- Phen/Anth in IAEA-477 ( $\mu$ g kg<sup>-1</sup>).

<b>ΓABLE 43. 1-METYLPYRENE RESU</b>	LTS REPORTED	BY PARTICIPANTS (	μg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
7	0.48	0.03	0.020	GC/HRMS	IAEA-459
13	0.93	0.09	0.023	GC/MS	IAEA-459

# TABLE 44. C1-FLUORANTHENES/PYRENES RESULTS REPORTED BY PARTICIPANTS (µg kg<sup>-1</sup>)

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	3.52	0.16	0.300	GC/MS	NIST 1941b
4	1.17	0.05	0.640	GC/MS	NIST 1941b
7	4.40	0.15	0.240	GC/HRMS	IAEA-459
10	5.78	0.27	0.100	GC/MS	NIST 1941b
13	7.21	0.35	0.023	GC/MS	IAEA-459



FIG. 38. Laboratory results used to calculate the information mass fraction of C1-Fluoranthenes/pyrenes in IAEA-477 ( $\mu$ g kg<sup>-1</sup>).

TABLE 45. C2-FLUORANTHENES/PYRENES	RESULTS REPORTED	Η BY PARTICIPANTS (με	g kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	3.72	0.18	0.300	GC/MS	NIST 1941b
4	1.11	0.03	0.640	GC/MS	NIST 1941b
7	2.53	0.27	0.240	GC/HRMS	IAEA-459
10	6.60	0.50	0.110	GC/MS	NIST 1941b
	c				



FIG. 39. Laboratory results used to calculate the information mass fraction of C2-Fluoranthenes/pyrenes in IAEA-477 ( $\mu$ g kg<sup>-1</sup>).

#### TABLE 46. C3-FLUORANTHENES/PYRENES RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	2.67	0.08	0.300	GC/MS	NIST 1941b
7	1.53	0.17	0.240	GC/HRMS	IAEA-459

TABLE 47. CHRYSEN	E RESULTS REPORTE	D BY PARTICIPANTS (µg kg <sup>-1</sup> )
-------------------	-------------------	--

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	4.23	0.15	0.300	GC/MS	NIST 1941b
3	1.71	0.16	0.500	HPLC-FLD	QPH094MS
6	7.55	0.53	0.500	GC/MS	NIST 1941b
7	2.52	0.10	0.020	GC/HRMS	IAEA-459



FIG. 40. Laboratory results used to calculate the information mass fraction of Chrysene in IAEA-477 (µg kg<sup>-1</sup>).

#### TABLE 48. TRIPHENYLENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-}1}$ )

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	0.84	0.04	0.300	GC/MS	NIST 1941b
7	0.78	0.04	0.020	GC/HRMS	IAEA-459

\*Calculated as: 2 x  $\frac{s}{\sqrt{n}}$  where s is the standard deviation and n is the number of measurements reported by participants.

# TABLE 49. C2-CHRYSENES RESULTS REPORTED BY PARTICIPANTS ( $\mu g \ kg^{-1}$ )

LAB Code	Mean	Uncertainty* Detection limit		Instrumentation	(C)RM use for QC
1	4.06	0.11	0.300	GC/MS	NIST 1941b
4	1.50	0.05	0.170	GC/MS	NIST 1941b
7	2.65	0.21	0.360	GC/HRMS	IAEA-459
10	3.90	0.25	0.120	GC/MS	NIST 1941b
13	5.50	1.78	0.011	GC/MS	IAEA-459

\*Calculated as: 2 x  $\frac{s}{\sqrt{n}}$  where s is the standard deviation and n is the number of measurements reported by participants.



FIG. 41. Laboratory results used to calculate the information mass fraction of C2-Chrysenes in IAEA-477 ( $\mu g k g^{-1}$ ).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	1.53	0.07	0.300	GC/MS	NIST 1941b
7	1.55	0.11	0.360	GC/HRMS	IAEA-459
13	6.47	1.97	0.011	GC/MS	IAEA-459
		Re	sults not used for t	the assignment value	
4	<100		0 170	GC/MS	NIST 1941h

# TABLE 50. C3-CHRYSENES RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-}1})$

4< LOQ</th>0.170GC/MSNIST 1941b\*Calculated as:  $2 \ge \frac{s}{\sqrt{n}}$  where s is the standard deviation and n is the number of measurements reported by participants.



FIG. 42. Laboratory results used to calculate the information mass fraction of C3-Chrysenes in IAEA-477 ( $\mu g k g^{-1}$ ).

TABLE 51. BENZO[B]FLU	ORANTHENE RESULTS	S REPORTED BY PAR	TICIPANTS (µg kg <sup>-1</sup> )
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LAB Code	Mean	Uncertainty* Detection limit		Instrumentation	(C)RM use for QC
1	3.09	0.11	0.700	GC/MS	NIST 1941b
3	5.45	0.26	0.500	HPLC-FLD	QPH094MS
6	11.10	1.46	0.500	GC/MS	NIST 1941b
7	3.42	0.27	0.010	GC/HRMS	IAEA-459



FIG. 43. Laboratory results used to calculate the information mass fraction of Benzo(b) fluoranthene in IAEA-477 (µg kg<sup>-1</sup>).
LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	1.59	0.05	0.700	GC/MS	NIST 1941b
6	4.00	0.27	0.500	GC/MS	NIST 1941b
7	0.90	0.16	0.010	GC-HRMS	IAEA-459

### TABLE 52. BENZO[J]FLUORANTHENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g kg^{-1}$ )

\*Calculated as: 2 x  $\frac{s}{\sqrt{n}}$  where s is the standard deviation and n is the number of measurements reported by participants.



FIG. 44. Laboratory results used to calculate the information mass fraction of Benzo(j) fluoranthene in IAEA-477 (µg kg<sup>-1</sup>).

# TABLE 53. BENZO[A]FLUORANTHENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-}1}$ )

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	0.61	0.04	0.700	GC/MS	NIST 1941b
13	0.80	0.19	0.024	GC/MS	IAEA-459
	~ S				

\*Calculated as: 2 x  $\frac{s}{\sqrt{n}}$  where s is the standard deviation and n is the number of measurements reported by participants.

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for OC
1	0.45	0.02	0 700	CC/MS	
T	0.45	0.02	0.700	GC/IVIS	NIST 1941D
2	0.45	0.04	0.290	GC/MSMS	IAEA-408
3	0.33	0.03	0.500	HPLC-FLD	QPH094MS
6	0.67	0.09	0.500	GC/MS	NIST 1941b
7	0.25	0.02	0.020	GC/HRMS	IAEA-459
9	0.43	0.04	0.009	GC/MSMS	IAEA-459
10	0.50	0.00	0.160	GC/MS	NIST 1941b
11	0.37	0.07	0.400	GC/MS	IAEA-383
13	0.40	0.05	0.131	GC/MS	IAEA-459
		Re	sults not used for	the assignment value	
4	<loq< td=""><td></td><td>0.650</td><td>GC/MS</td><td>NIST 1941b</td></loq<>		0.650	GC/MS	NIST 1941b
12	<2.69		1.620	GC/MS	IAEA-383
	c				

## TABLE 54. DIBENZ(A,H)ANTHRACENE RESULTS REPORTED BY PARTICIPANTS (µg kg<sup>-1</sup>)

\*Calculated as: 2 x  $\frac{s}{\sqrt{n}}$  where s is the standard deviation and n is the number of measurements reported by participants.



FIG. 45. Laboratory results used to calculate the information mass fraction of Dibenz(a,h) anthracene in IAEA-477 (µg kg<sup>-1</sup>).

LAB Code	Mean	Uncertainty*	Detection limit	Instrumentation	(C)RM use for QC
1	13.91	0.65	0.300	GC/MS	NIST 1941b
2	38.20	0.57	0.300	GC/MSMS	IAEA-408
4	8.82	0.26	1.100	GC/MS	NIST 1941b
6	74.83	7.33	0.500	GC/MS	NIST 1941b
7	16.62	0.58	0.010	GC/HRMS	IAEA-459
10	70.97	2.57	0.170	GC/MS	NIST 1941b
11	31.57	3.63	0.400	GC/MS	IAEA-383
12	39.07	3.44	1.630	GC/MS	IAEA-383
13	26.81	2.52	0.010	GC/MS	IAEA-459

# TABLE 55. PERYLENE RESULTS REPORTED BY PARTICIPANTS ( $\mu g \; kg^{\text{-}1})$

\*Calculated as: 2 x  $\frac{s}{\sqrt{n}}$  where s is the standard deviation and n is the number of measurements reported by participants.



FIG. 46. Laboratory results used to calculate the information mass fraction of Perylene in IAEA-477 (µg kg<sup>-1</sup>).

Alibrants source Calibrants used Surrogates source Surrogates used	Calibrants used Surrogates source Surrogates used	Surrogates source Surrogates used	Surrogates used		CRMs source	Matrix CRMs
						Used
NIST 2260a CAMBRIDGE d8-Naphthalene	2260a CAMBRIDGE d8-Naphthalene	CAMBRIDGE d8-Naphthalene	d8-Naphthalene		NIST	SRM
Isotope d8-Acenaphtylene	Isotope d8-Acenaphtylene	Isotope d8-Acenaphtylene	d8-Acenaphtylene			1941b
Laboratories – 98% d10-Acenaphtene	Laboratories – 98% d10-Acenaphtene	Laboratories – 98% d10-Acenaphtene	d10-Acenaphtene			
d10-Fluorene	d10-Fluorene	d10-Fluorene	d10-Fluorene			
d10-Phenanthrene	d10-Phenanthrene	d10-Phenanthrene	d10-Phenanthrene			
d10-Anthracene	d10-Anthracene	d10-Anthracene	d10-Anthracene			
d8-Dibenzothioph	d8-Dibenzothioph	d8-Dibenzothioph	d8-Dibenzothioph	ene		
d10-Fluoranthen	d10-Fluoranthen	d10-Fluoranthen	d10-Fluoranthen	Ð		
d10-Pyrene	d10-Pyrene	d10-Pyrene	d10-Pyrene			
d12-Benz[ <i>a</i> ]ar	d12-Benz[a]ar	d12-Benz[a]ar	d12-Benz[ <i>a</i> ]ar	ithracene		
d12-Chrysene	d12-Chrysene	d12-Chrysene	d12-Chrysene			
d12-Benzo[ <i>b</i> ]f	d12-Benzo[ <i>b</i> ]f	d12-Benzo[ <i>b</i> ]f	d12-Benzo[ <i>b</i> ]f	luoranthene		
d12-Benzo[k]f	d12-Benzo[k]f	d12-Benzo[k]f	d12-Benzo[ <i>k</i> ]f	luoranthene		
d12-Benzo[e	d12-Benzo[e	d12-Benzo[e	d12-Benzo[ <i>e</i>	Jpyrene		
d12-Benzo[ <i>a</i> ]p	d12-Benzo[ <i>a</i> ]p	d12-Benzo[ <i>a</i> ]p	d12-Benzo[ <i>a</i> ]p	yrene		
d12-Indeno[1,2	d12-Indeno[1,2	d12-Indeno[1,2	d12-Indeno[ <i>1,2</i>	<i>,,3-c,d</i> ]pyrene		
d14-Dibenz[ <i>a</i> , <i>h</i> ]	d14-Dibenz[a,h]	d14-Dibenz[ <i>a</i> , <i>h</i> ]	d14-Dibenz[a,h	Janthracene		
d12-Benzo[ <i>g,h</i> ,	d12-Benzo[ <i>g</i> , <i>h</i> ,	d12-Benzo[ <i>g,h</i> ,	d12-Benzo[ <i>g,h,</i>	Jperylene		

Table 56. Calibrants, surrogates and CRMs used by Participants

**APPENDIX III: TRACEABILITY TABLE** 

	Matrix CRMs Used	IAEA-408	Proficiency test	
	CRMs source	IAEA	QUASIMEME	
	Surrogates used	d8-Naphthalene d10-Anthracene d10-Fluoranthene d12-Benzo[g),h,i]perylene d12-Benzo[g,h,i]perylene		
	Surrogates source	Chiron AS, DR.EHRENSTRORFE R GMBH		
. (cont.)	Calibrants used	Acenaphthene Acenapthylene Anthracene Benz[a]anthracene Benzo[a]pyrene Benzo[b]fluoranthene Benzo[j]fluoranthene Benzo[g,h,j]perylene Chrysene Benzo[g,h]anthracene Benzo[g,h]anthracene Eluoranthene Fluoranthene Fluoranthene Perylene Naphthalene Perylene Perylene Naphthalene Perylene Perylene Indeno[1,2,3-c,d] pyrene Naphthalene Perylene Perylene Perylene 2.6-Dimethylnaphthalene 2.6-Dimethylnaphthalene 2.6-Dimethylphenanthrene 3.6-Dimethylphenanthrene 1,2,8-	PM-831A-1 (16 PAH compounds in acetonitrile-	methanol 9:1)
	Calibrants source	Chiron AS, DR.EHRENSTRORF ER GMBH	ULTRASCIENTIFIC	
Table 56	Lab Code	2	£	

	Matrix CRMs Used	1941b, 1944
	CRMs source	NST
	Surrogates used	d8-Naphthalene d10- Acenaphthene d10-Phenanthrene d10-Phenanthrene d12-Perylene GC internal std: d14-p-terphenyl
(cont.)	Surrogates source	AccuStandard, Chiron AS
	Calibrants used	Naphthalene 1-Methylnaphthalene 2-Methylnaphthalene 2-Methylnaphthalene 2.6-Dimethylnaphthalene Acenaphthene 1-n-Propylnaphthalene Acenaphthene 1-n-Propylfluorene 1-methylfluorene 9-n-Butylfluorene 9-n-Butylfluorene 9-n-Butylfluorene 9-n-Butylfluorene 1,2- Dimethyldibenzothiophene 1,2- Methylphenanthrene 4-Methylphenanthrene 2-Methylphenanthrene 1,2,5-/1,2,7- Dimethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene 1,2,6,9- Tetramethylphenanthrene Retene (1-Methyl-7- isopropylphenanthrene
	Calibrants source	AccuStandard, Chiron AS
Table 56	Lab Code	4

	Matrix CRMs	Used												
	CRMs source													
	Surrogates used													
	Surrogates source													
	Calibrants used		Benz[ <i>a</i> ]anthracene	Chrysene	6-Methylchrysene	6-Ethylchrysene	Benzo[ <i>b</i> ]Fluoranthene	Benzo[k]Fluoranthene	Benzo[ <i>e</i> ]pyrene	Benzo[ <i>a</i> ]Pyrene	Perylene	Indeno[ <i>1,2,3-c,d</i> ]pyrene	Dibenz[ <i>a,h</i> ]anthracene	Benzo[ <i>g,h,i</i> ]perylene
. (cont.)	Calibrants source													
Table 56.	Lab Code			4	(cont)									

Lab	Calibrants source	Calibrants used	Surrogates source	Surrogates used	CRMs source	Matrix
Code						CRMs Used
ъ	SUPELCO	Polynuclear Aromatic Hydrocarbons Mix	RESTEK, DR.	Revised SV Internal Standard MIX Catalog No. 31885, Lot No. A0125855	IAEA	IAEA-459
		2000μg/mL each	EHRENSTORFER	d8-Naphthalene		
		component in benzene:	GMBH	d10- Acenaphthene		
		dichloromethane (50:50)		d10- Phenanthrene		
		Part Number: CRM47543		d12-Chrysene		
		Naphthalene		d12-Perylene		
		1-Methylnaphthalene				
		2-Methylnaphthalene		GC internal std:		
		Acenaphthylene		d10-Fluoranthene		
		Acenaphthene				
		Fluorene				
		Phenanthrene				
		Anthracene				
		Fluoranthene				
		Pyrene				
		Benz[ <i>a</i> ]anthracene				
		Chrysene				
		Benzo[ <i>b</i> ]fluoranthene				
		Benzo[k]fluoranthene				
		Benzo[ <i>a</i> ]pyrene				
		Indeno[ <i>12,3-c,d</i> ]pyrene				
		Dibenz[ <i>a</i> , <i>h</i> ]anthracene				
		Benzo[ <i>g,h,i</i> ]perylene				

Table 56. (cont.)

	Matrix CRMs Used	IMR Laboratory material (LRM)
	CRMs source	(LRM)
	Surrogates used	d8-Naphthalene d10-Biphenyl d8-Acenaphthylene d10-Anthracene d10-Pyrene d12-Perylene d12-Indeno[1,2,3-c,d]pyrene d12-Indeno[1,2,3-c,d]pyrene d10-Phenanthrene
(cont.)	Surrogates source	Chiron AS
	Calibrants used	Naphthalene 2-Methylnaphthalene 1-Methylnaphthalene Biphenyl 2.6-Dimethylnaphthalene 1.3-Dimethylnaphthalene 2.3-Dimethylnaphthalene 2.3-Dimethylnaphthalene Acenaphthylene Acenaphthylene 1.4-Dimethylnaphthalene 2.3.5-Trimethylnaphthalene 1.2.3-Trimethylnaphthalene 1.2.3-Trimethylnaphthalene 1.2.5.6- Trimethylnaphthalene 1.2.5.6- Trimethylnaphthalene Bluorene 9-Ethylfluorene Dibenzothiophene Phenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2-Methylphenanthrene 2.5-Dimethylphenanthrene
	Calibrants source	Chiron AS
Table 56	Lab Code	٥

	Matrix	CRMs Used																					
	<b>CRMs source</b>																						
	Surrogates used																						
	Surrogates source																						
	Calibrants used		1.2-Dimethylphenanthrene	2.6.9-	Trimethylphenanthrene	1.2.6-	Trimethylphenanthrene	1.2.7-	Trimethylphenanthrene	1.2.6.9-	Tetramethylphenanthrene	Fluoranthene	Pyrene	Benz[ <i>a</i> ]anthracene	Chrysene	Benzo[ <i>b</i> ]fluoranthene	Benzo[k]fluoranthene	Benzo[ <i>a</i> ]pyrene	Benzo[ <i>e</i> ]pyrene	Perylene	Indeno[ <i>1,2,3-c,d</i> ]pyrene	Dibenz[ <i>a,h</i> ]anthracene	Benzo[ <i>g,h,i</i> ]perylene
· (coller)	<b>Calibrants source</b>			Chiron AS																			
ו מחוב הכ	Lab	Code		9	(cont.)																		

Table 56. (cont.)

	Matrix CRMs Used	IAEA-459
	CRMs source	IAEA
	Surrogates used	Native individual PAHs
	Surrogates source	
	Calibrants used	US EPA 16 PAH Cocktail ( <sup>13</sup> C, 99%): Acenaphthene Acenaphthylene Anthracene Benz[ <i>a</i> ]anthracene Benzo[ <i>b</i> ]fluoranthene Benzo[ <i>a</i> ]pyrene Benzo[ <i>a</i> ]pyrene Benzo[ <i>a</i> , <i>h</i> ]perylene Benzo[ <i>a</i> , <i>h</i> ]anthracene Chrysene Chrysene Dibenzo[ <i>a</i> , <i>h</i> ]anthracene Fluoranthene Fluorene Indeno[1,2,3-c, <i>d</i> ] pyrene Naphthalene Pyrene Pyrene
· (collec)	Calibrants source	CIL
ו מחוב הר	Lab Code	7

Table 56. (cont.)

Matrix CRMs	IAEA-459
CRMs source	IAEA
Surrogates used	PAH Mix31 d10-Acenaphthene d12-Chrysene d8-Naphthalene d12-Perylene d10-Phenanthrene
Surrogates source	DR. EHRENSTORFER GMBH
Calibrants used	PAH Mix9 Naphthalene Acenaphthylene Fluorene Fluoranthrene Anthracene Fluoranthene Fluoranthene Pyrene Benz[a]anthracene Chrysene Benzo[b]fluoranthene Benzo[a]pyrene Benzo[a,h]anthracene Indeno[1,2,3-c,d] pyrene Benzo[g,h,i]perylene
Calibrants source	DR. EHRENSTORFER GMBH
Lab Code	σ

	Matrix CRMs Used	1941b
. (cont.)	CRMs source	NIST
	Surrogates used	d10-Acenaphthene d12-Benzo[g, <i>h</i> , <i>i</i> ]Perylene d12-Chrysene d8-Naphthalene d12-Perylene d10-Pyrene d10-Pyrene
	Surrogates source	Chemservice
	Calibrants used	1,2-Dimethylnaphthalene 1-Methylnaphthalene 2,3,5-Trimethylnaphthalene 2,6-Dimethylnaphthalene 2,6-Dimethylphenanthrene 3,6-Dimethylphenanthrene 3,6-Dimethylphenanthrene Acenaphthylene Anthracene Benz[ <i>a</i> ]anthracene Benz0[ <i>a</i> ]jpyrene Benz0[ <i>a</i> ]jpyrene Benz0[ <i>a</i> ]jperylene Benz0[ <i>a</i> ]jprene Benz0[ <i>a</i> ]jprene Perylene Perylene Prenanthrene Prene
	Calibrants source	Chemservice
Table 56.	Lab Code	10

del	Calibrante cource	Calibrants used	Surrogates cource	Surrogates used	CPMc cource	Matriv
Lau			our ugares source	our ogates used	CUIVIS SUULCE	
Code						CRMs Used
11	ULTRASCIENTIFIC	PAH MIX (22 analytes)	ULTRASCIENTIFIC	8270 Base/Neutral Surrogate Standard Mixture	NIST	1944
					IAEA	IAEA-383
				Semivolatiles internal std mix EPA 8270 (6 analytes)		
12	ULTRASCIENTIFIC	PAH Mixture ITPM-023	ULTRASCIENTIFIC	7 Deuterated PAH mix ISM-750	IAEA	IAEA-383
		Naphthalene		d10- Acenaphthene		
		Acenaphthylene		d10- Phenanthrene		
		Acenaphthene		d10-Fluorantene		
		Fluorene		d12-Benzo[ <i>a</i> ]antracene		
		Phenanthrene		d12-Dibenzo[a,h]antracene		
		Anthracene				
		Fluoranthene		GC internal std:		
		Pyrene		3 Deuterated PAH mix ISM-740A		
		Benz[ <i>a</i> ]anthracene				
		Chrysene				
		Benzo[ <i>b</i> ]fluoranthene				
_		Benzo[k]fluoranthene				
		Benzo[ <i>e</i> ]pyrene				
		Benzo[ <i>a</i> ]pyrene				
		Perylene				
		Indeno[ <i>1,2,3-c,d</i> ]pyrene				
		Benzo[ <i>g,h,y</i> ]perylene				
		Dibenz[ <i>a,h</i> ]anthracene				
		Dibenz[ <i>a,l</i> ]pyrene				
		Dibenz[ <i>a,e</i> ]pyrene				
		Dibenz[ <i>a,i</i> ]pyrene				
		Dibenz[ <i>a,h</i> ]pyrene				
		Benzolfluoranthene				

	Matrix CRMs Used	IAEA-459
	CRMs source	IAEA
. (cont.)	Surrogates used	d8-Naphthalene d10- Acenaphtylene d8-Acenaphtylene d10- Phenanthrene d12-Chrysene d12-Perylene
	Surrogates source	ULTRASCIENTIFIC
	Calibrants used	Naphthalene 1-Methylnaphthalene 2-Methylnaphthalene 2.6-Dimethylnaphthalene 2.3.5-Trimethylnaphthalene Biphenyl Acenaphthene Acenaphthene Fluorene 1-Metylfluorene Phenanthrene Phenanthrene Phenanthrene 2-Methylphenanthrene Phenanthrene Phenanthrene Phenanthrene 2-Methylphenanthrene Phenanthrene Phenanthrene Phenanthrene Phenanthrene Benzo[a]fluoranthene Benzo[a]phrene Ben
	Calibrants source	ULTRASCIENTIFIC
Table 56.	Lab Code	13

All available IAEA reference materials may be found in the Reference Material Online Catalogue, <u>http://nucleus.iaea.org/rpst/ReferenceProducts/ReferenceMaterials</u>.

### REFERENCES

- [1] INTERNATIONAL ORGANIZATION FOR STANDARDIZATION, ISO Guide 35:2017, Reference Materials Guidance for characterization and assessment of homogeneity and stability, ISO, Geneva (2017).
- [2] JOINT COMMITTEE FOR GUIDES IN METROLOGY (JCGM), Evaluation of measurement data – Guide to the expression of uncertainty in measurement, JCGM 100: 2008 (GUM 1995 with minor corrections), (2008). http://www.bipm.org/utils/common/documents/jcgm/JCGM 100 2008 E.pdf
- [3] INTERNATIONAL ORGANIZATION FOR STANDARDIZATION, ISO 17034: 2016, General requirements for the competence of reference material producers, ISO, Geneva (2016).
- [4] INTERNATIONAL ORGANIZATION FOR STANDARDIZATION, ISO/IEC 17025:2017. General requirements for the competence of testing and calibration laboratories, Geneva, (2017).
- [5] LINSINGER T., PAUWELS J., VAN DER VEEN A., SCHIMMEL H., LAMBERTY A., Homogeneity and stability of reference materials, Accredit. Qual. Assur. 6 1 (2001) 20–25.
- [6] INTERNATIONAL ATOMIC ENERGY AGENCY, (IAEA/AQ/52) (2017). Certification of mass fractions of polycyclic aromatic hydrocarbons, organochlorines and polybrominated diphenyl ethers in IAEA-459 marine sediment sample, IAEA Analytical Quality in Nuclear Applications Series No. 52, IAEA, Vienna
- [7] SRM 1944. Certificate of Analysis. New York/New Jersey Waterway Sediment (NIST, Gaithersburg, MD, 2011);
- [8] SRM 1941b, Organics in Marine Sediment, (NIST, Gaithersburg, MD, 2015
- [9] International vocabulary of metrology Basic and general concepts and associated terms (VIM). JCGM 200:2008
- [10] VESPER, A., EMONS, H., GNEZDA, M., JAIN, C., MILLER, G., REJ, R., SCHUMANN, G., TATE, J., THIENPONT, L., VAKS, E., Characterization and qualification of commutable reference materials for laboratory medicine; Approved Guideline, CLSI document EP30-A, Clinical and Laboratory Standards Institute, Wayne, PA (2010).

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