



# **Recent Advances in Handheld XRF for Site Remediation**

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# IEC



**International Electrotechnical Commission**  
**Creator of international standards for**  
**electrotechnical equipment**

**TC45 – Nuclear Instrumentation**

**PT 62495 – Portable X-ray Fluorescence**  
**Analysis Equipment Utilizing a Miniature**  
**X-ray Tube**

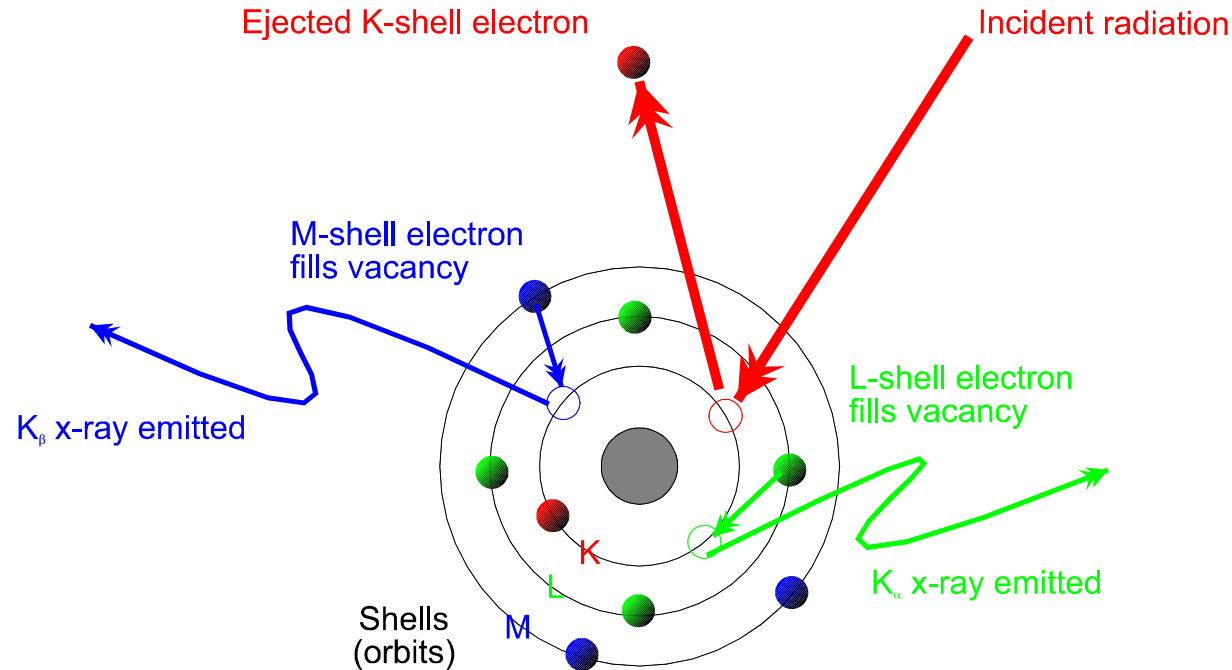
# Outline



## Outline

- **XRF Theory**
- **EPA 6200**
- **Hardware Advances**
  - **Miniature X-ray tubes**
  - **Silicon Detectors**
  - **GPS**
- **Software Advances**
  - **Mapping**
  - **User Empirical Calibrations**
  - **Type Standardization**
- **Limits of Detection**

# Characteristic x-ray production



K shell x-rays analyzed up to about 30 kV AN ~30;  
ex. Ti (K<sub>α</sub> @ 4.5 kV), Ag (22.2 kV), Sb (26.3 kV)

L shell x-rays analyzed for elements above AN ~30  
ex. Hf (K<sub>α</sub> @ 55.8, La @ 7.9); Pb (K<sub>α</sub> 75.0, La 10.5)

# XRF Rules



1. **Energy of X-ray tells you what element is present**
2. **Number of X-rays tells you how much is present**
3. **Measurement precision increases with increased number of counts**
4. **Energy decreases with atomic number (Z)**
5. **Absorption increases with decreasing energy**

# EPA 6200



## EPA 6200

**Field work 1997**

**Published 1998**

**Updated 2007**

## Covers

**Radioactive Sources**

**Proportional counters**

**Early solid state  
detectors**

<b>EPA 6200</b>	
<b>LOD</b>	
<b>mg/Kg</b>	
<b>Arsenic (As)</b>	<b>40</b>
<b>Cadmium (Cd)</b>	<b>100</b>
<b>Chromium (Cr)</b>	<b>150</b>
<b>Lead (Pb)</b>	<b>20</b>
<b>Mercury (Hg)</b>	<b>30</b>
<b>Silver (Ag)</b>	<b>70</b>
<b>Strontium (Sr)</b>	<b>10</b>

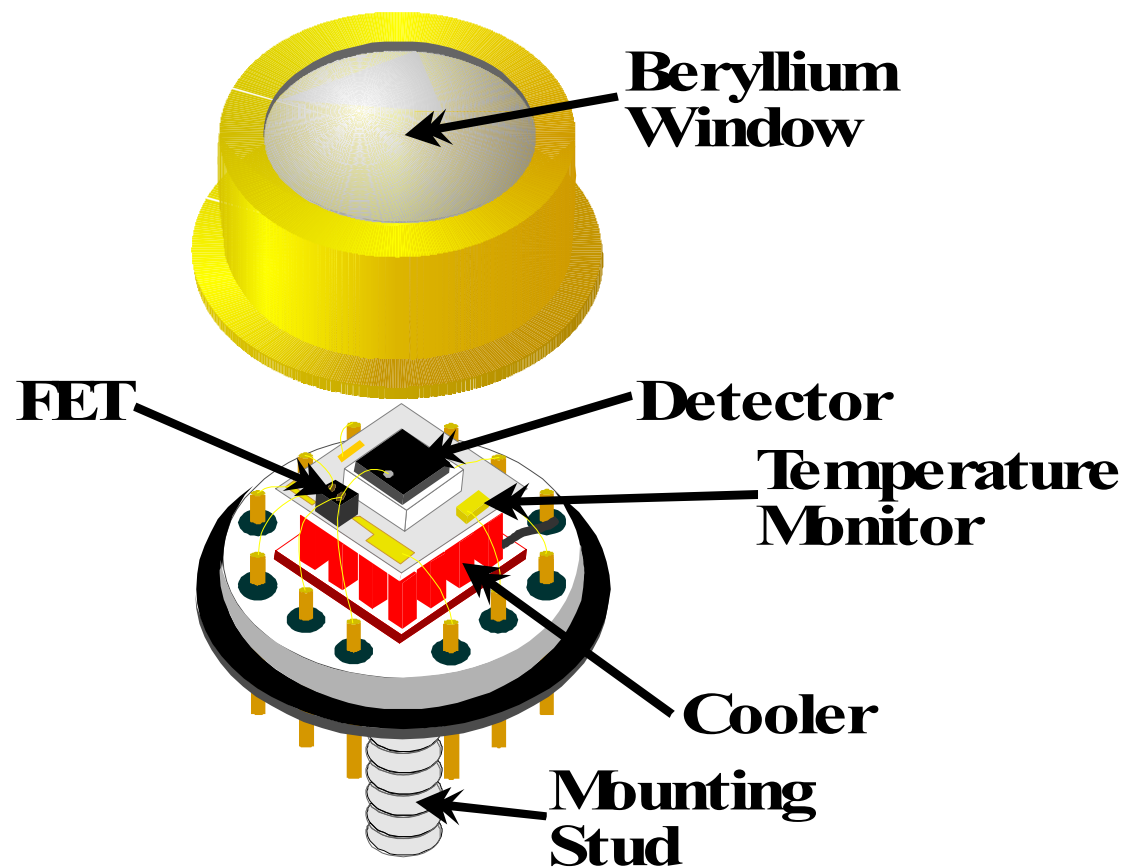
# Miniature X-ray Tubes



**Up to 50 kV**  
**Completely Adjustable**  
**No Radioactive Material**  
**Weight ~10-15 grams**



# SiPIN Detector

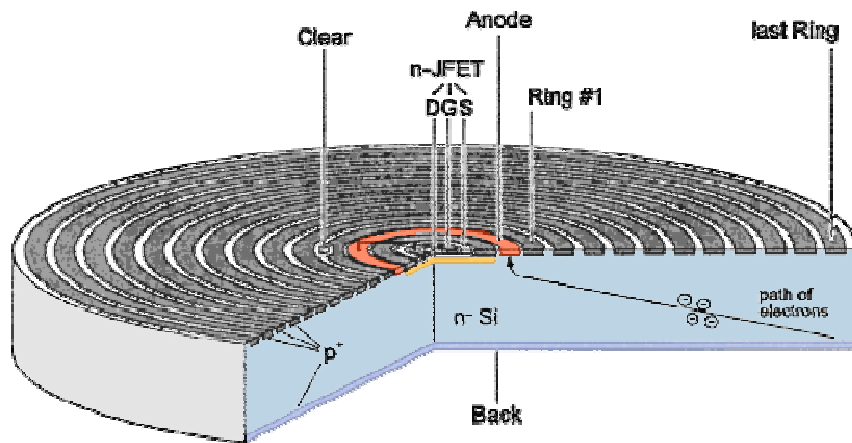


**Detector capacitance increases with increasing area**

**Detector resolution decreases with increasing area**



# Silicon Drift Detector



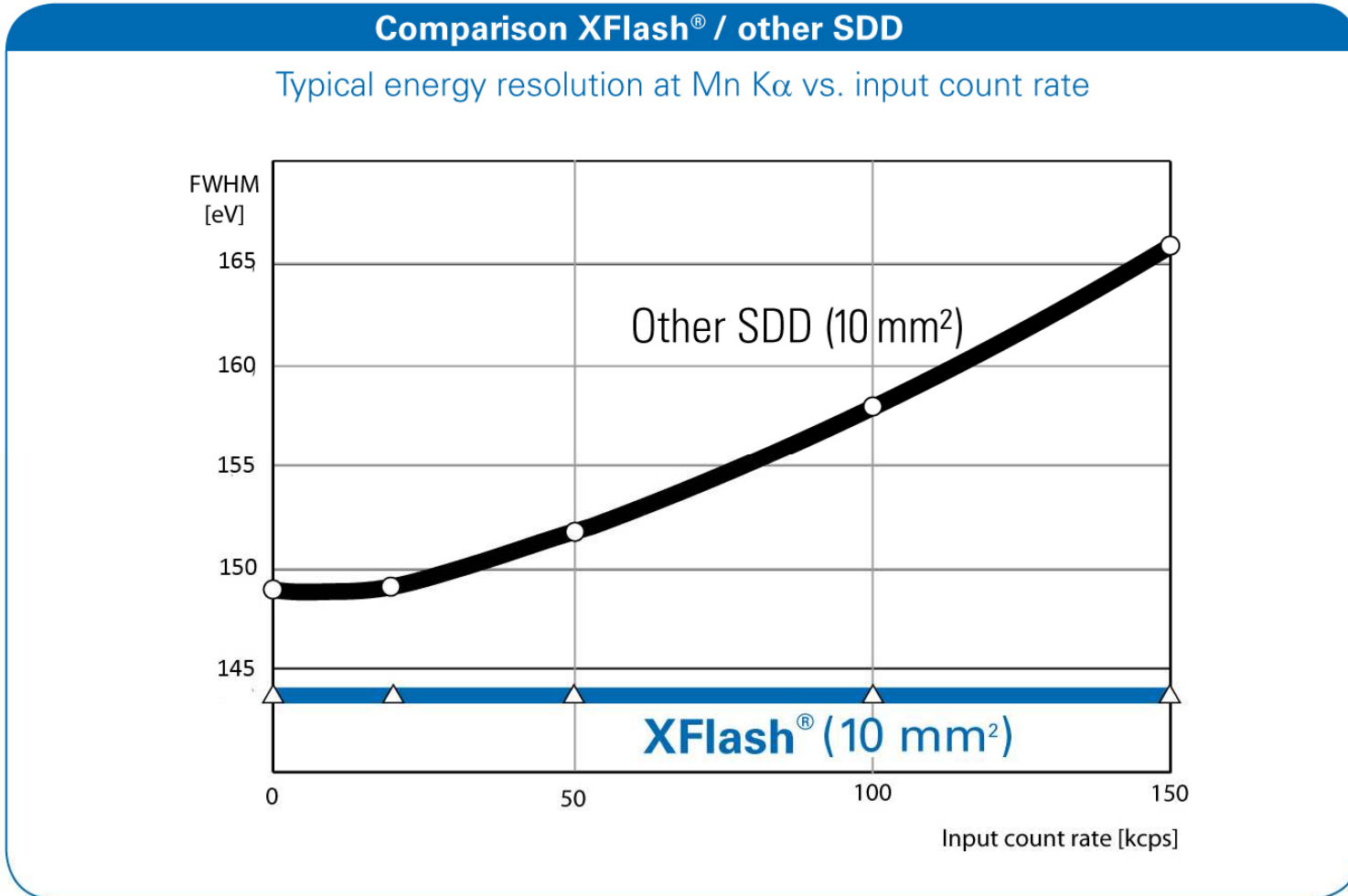
## SDD

- ❖ Integrated drift structure results in electron capture at center electrode
- ❖ Small contact means low capacitance
- ❖ Thin entrance window

## SDD Offers

- ❖ Improved Resolution
- ❖ High Count Rate
- ❖ Low Z sensitivity

# SDD Resolution



# Results of SDD



**High Count rate means better precision in a given counting time**

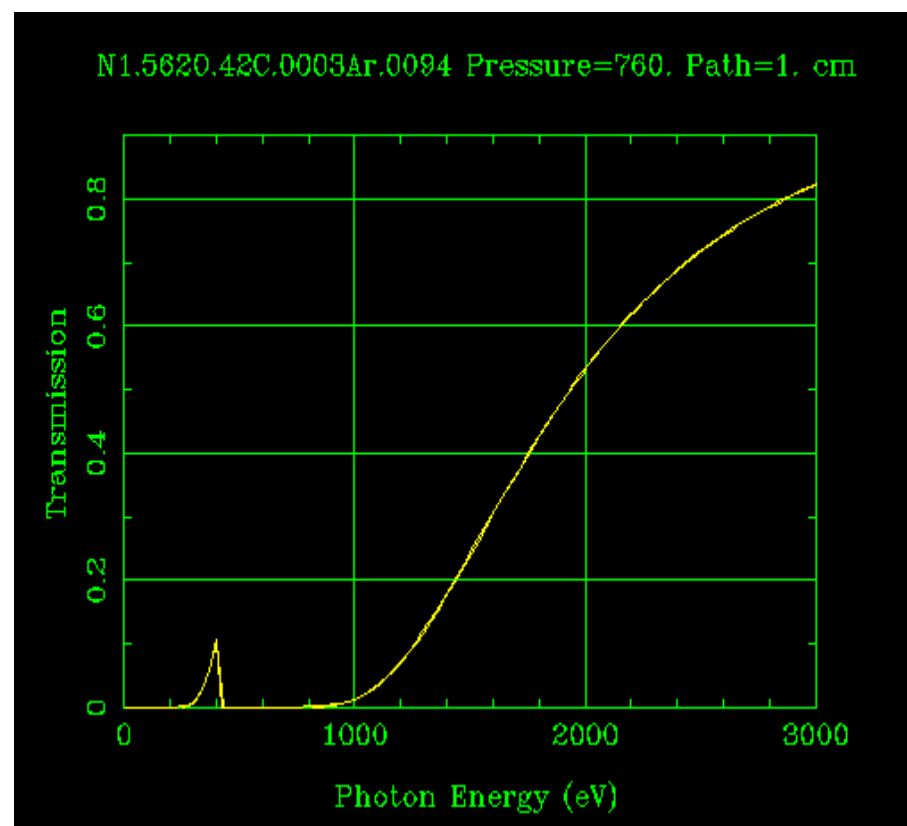
**Better resolution means better signal to noise thus further improving the precisions**

**Resolution is especially important at low Z where neighboring peaks are  $\sim 250$  eV apart**



# Transmission of 1 cm Air Path

	Energy	Trans.
Mg	1.3	11.9%
Al	1.5	24.0%
Si	1.7	36.6%
P	2.0	53.2%
S	2.3	65.5%



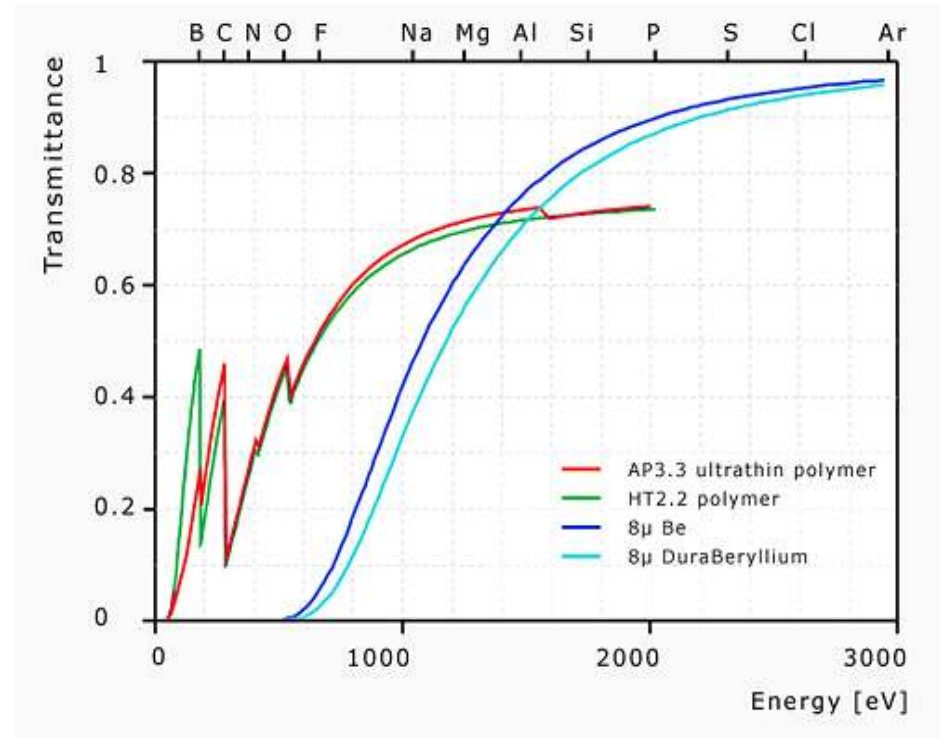
# Careful Design for Low Z Detection



**Thin Detector window**

**Thin Be window**

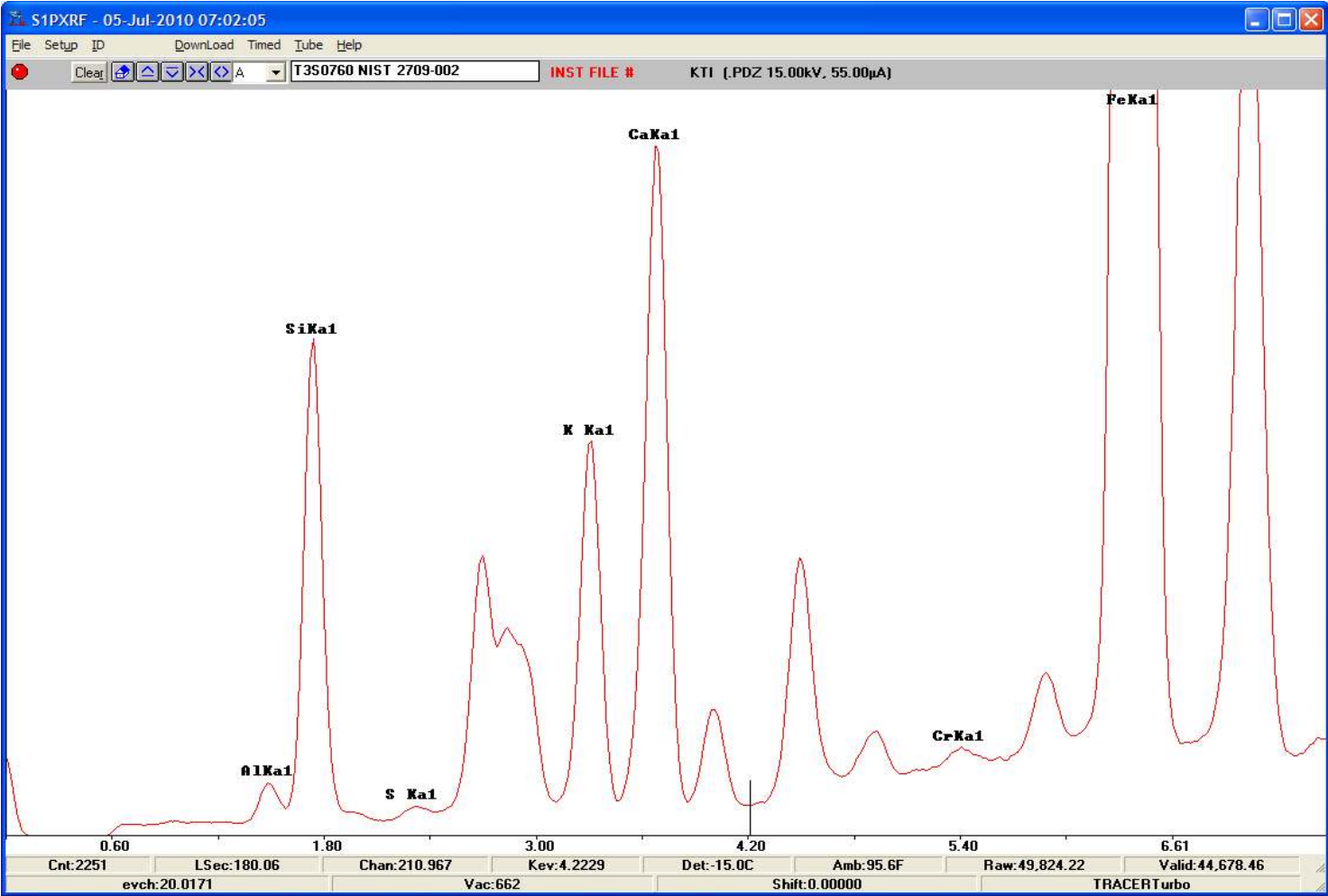
**Minimize distance  
between sample  
and detector**



# Low Z Spectrum



**Al @ 7.5%; Si @ 29.7%; K @ 2.0%**



# Low Z Detection



## SiPIN Detectors

- **Low Z limit  $\sim$ TI (22)**
- **Some Detection at Ca (20)**

## SDD

- **Low Z limit  $\sim$ Si (14)**
- **Some Detection as Mg (12)**

		<b>LOD</b>
<b>AN</b>	<b>Element</b>	<b>mg/Kg</b>
<b>12</b>	<b>Mg</b>	<b>6600</b>
<b>13</b>	<b>Al</b>	<b>1050</b>
<b>14</b>	<b>Si</b>	<b>790</b>
<b>16</b>	<b>S</b>	<b>90</b>
<b>20</b>	<b>Ca</b>	<b>50</b>
<b>30</b>	<b>Zn</b>	<b>4</b>

# Low Z detection



## Why does it matter?

- **Allows selection of proper calibration**  
Ex. if not Si matrix then substantial error will occur, the software can select a S based matrix or others
- **Provides a more complete and accurate analysis**



# Typical SDD Analyzers



## Typical Specifications:

**Weight: ~2 Kg**

**Batteries: 4-8 hours**

**Voltage: 40-50 kV**

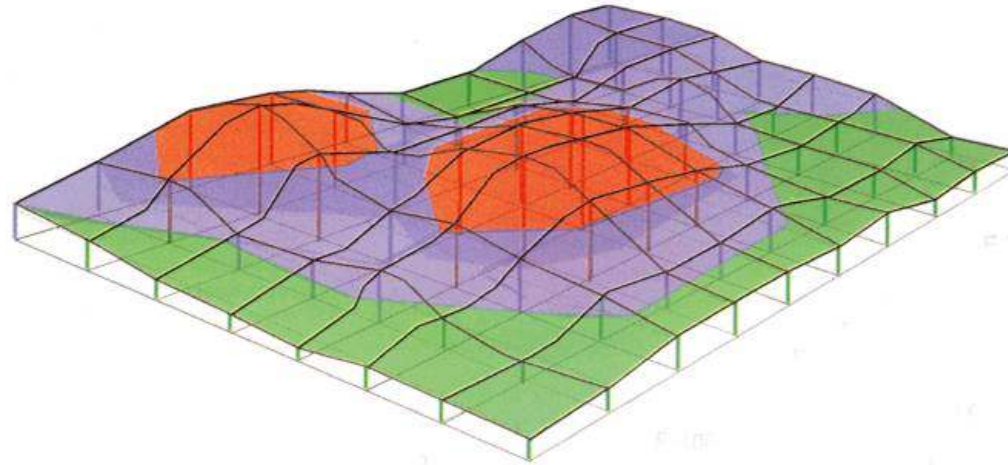
**Input Count Rates: 200 kcps**

**Heavy Metal LOD: 2-10 mg/kg**

**Factory Supplied Calibration**



# GPS/GIS



**GPS is easily integrated to handheld XRF analyzer**

**Differential GPS can achieve spatial resolution of ~10 cm**

**Data can be exported with the assay data and all GPS coordinates to a mapping program**

# Sample Preparation



**XRF is a surface measurement (depth of a few mm depending on Z).**

**Precision and accuracy will depend on the level of sample preparation**

**Sample preparation required depends on you data quality objectives**

**Quantitative results can be achieved in situ**

**Almost lab grade quantitative can be achieved with on site sample preparation**

# Sample Preparation



## In situ measurements

- **Remove any possible surface debris, stones, vegetation, and excess moisture from measurement area.**
- **Select a test area where soil particle-size is small and which looks homogeneous.**
- **Position the nose of the analyzer against the ground and start the measurement.**
- **After each measurement, clean the nose of the instrument with a soft cloth or tissue.**
- **Send 5 – 10% of samples to laboratory**

# Sample Preparation



## On Site Sample Preparation (any part improves results)

- **Dry sample**
  - Absorbent paper
  - Dry overnight in shallow pan
  - Heat in oven until constant weight (care must be taken to prevent volatile compounds escape)
- **Grind/Sieve**
  - Sieve out large debris
  - Grind to 60 mesh (better 120 mesh)
  - Sieve
- **Prepare homogeneous sample in XRF cups.**

# XRF Sample Cups and Holder



**Sample Bags**



**Analyzer in Stand**

# Site Specific Calibration



**Empirical – based on standards **which are representative** of samples to be measured**

**Advantage –**

**Can exactly match the site conditions**

**For homogeneous samples give quantitative results**

**Disadvantage –**

**Requires understanding of XRF**

**Number of standards required**

**Time required to create calibration**

# Site Specific Calibration



**Fundamental Parameters – based on measurement of a few standards, **assumptions** about the nature of the sample and theoretical understanding of XRF**

**Advantage -**

**Ready “out of the box”**

**Disadvantage –**

**Does not represent site conditions**

**Probably gives “semi-quantitative” results**



# Site Specific Calibration



## Factors which affect the calibration

**Matrix**

**Particle size**

**Density**

**Moisture content**

## Typical FP calibration assumptions

**SiO<sub>2</sub>**

**Small particle size (120 mesh)**

**Very low moisture content**



# Type Standardization "Calibration Correction"

**Most differences between measured assay  
and actual are from difference between  
calibration assumption and reality**

**Most of these differences are linearly  
related to the concentration**

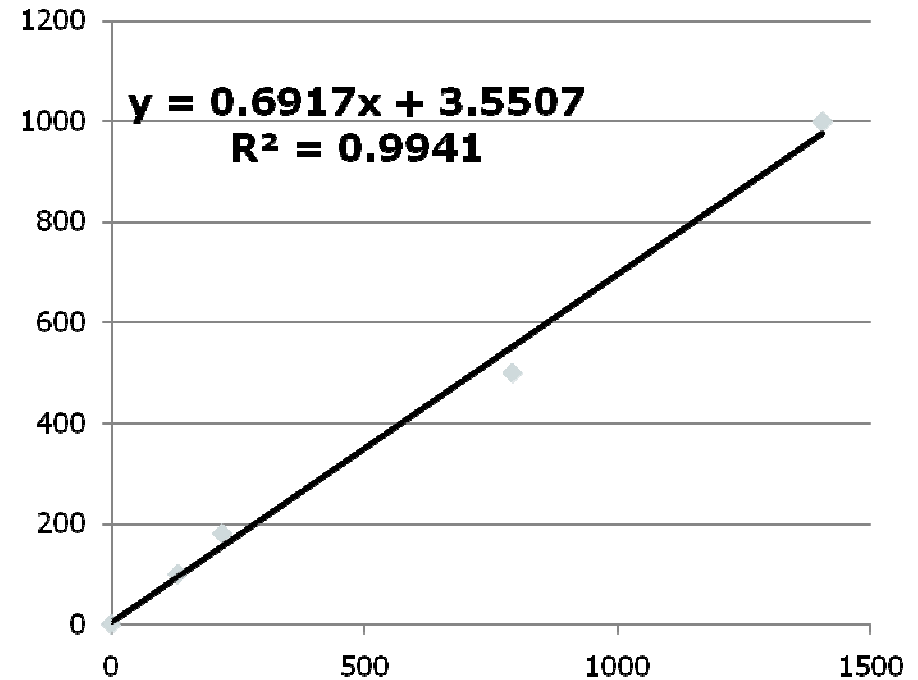
$$\text{Actual} = \text{Slope} * \text{Measured} + \text{Offset}$$

**Type standardization allows correction of  
measured data**

# Correction Calculation



Element 1	Ni	
SLOPE	0.6917	
OFFSET	3.5507	
CORR	99.7%	
Reference values	S1 Meas. Results	Adjusted S1 Result
0	0	3.55
100	131.1	94.24
180	219.45	155.35
500	792.17	551.53
1000	1404.84	975.33



# What to Do for Site Specific Assay



**Always try Type Standardization first**

- **Measure 5-10 well characterized samples**
- **Plot Actual vs. Measured**
- **If the result is a straight line plot with a correlation coefficient  $> 0.9$  use Type Standardization**

**If not or if the **element of interest** is not in the basic calibration then Site Specific calibration will be required**

# Limit of Detection



## How LOD is calculated

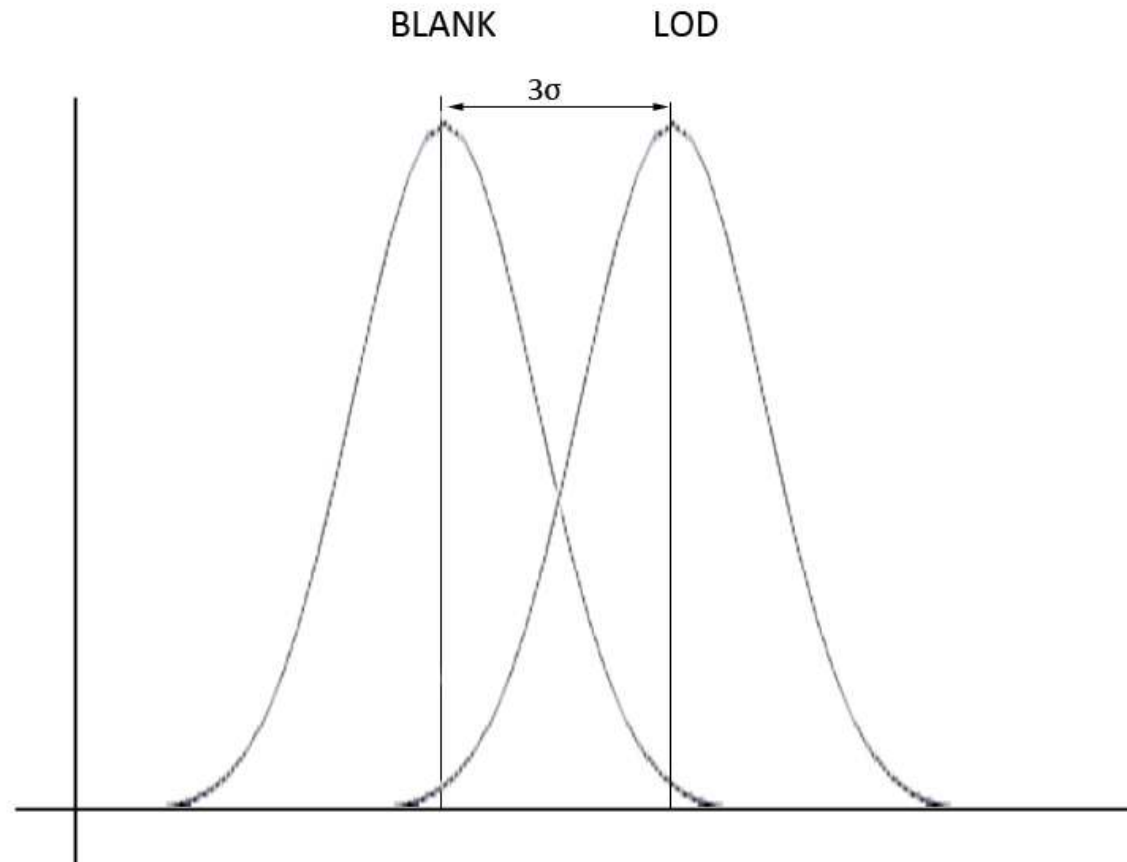
$$\text{LOD} = 3 * \sqrt{\text{BKG}} * \text{SENS}$$

**Where:**

**BKG = average counts in the region of interest in counts/second**

**SENS = slope of the calibration curve in ppm/cps**

# Result



# LOD Issues



**LOD= Concentration**

**If sample at exactly LOD is measured a result will be calculated only about 50% of the time**

**This does not indicate a concentration which can be measured**

**Good for comparing one instrument (or calibration to another)**

# Limit of Quantification



## Limit of Quantification

$$\text{LOQ} = 3 * \text{LOD}$$

**Good indicator of concentration which can be measured accurately**



# LOD and LOQ



	<b>EPA 6200</b>	<b>Current</b>	<b>Current</b>
	<b>LOD</b>	<b>LOD</b>	<b>LOQ</b>
	<b>mg/Kg</b>	<b>mg/Kg</b>	<b>mg/Kg</b>
<b>Arsenic (As)</b>	<b>40</b>	<b>4</b>	<b>12</b>
<b>Cadmium (Cd)</b>	<b>100</b>	<b>21</b>	<b>63</b>
<b>Chromium (Cr)</b>	<b>150</b>	<b>40</b>	<b>120</b>
<b>Lead (Pb)</b>	<b>20</b>	<b>12</b>	<b>36</b>
<b>Mercury (Hg)</b>	<b>30</b>	<b>5</b>	<b>15</b>
<b>Silver (Ag)</b>	<b>70</b>	<b>8</b>	<b>24</b>
<b>Strontium (Sr)</b>	<b>10</b>	<b>4</b>	<b>12</b>

**Data is specifically for Bruker S1 TURBO<sup>SD</sup>**  
**Representative of state of the art**

# Conclusions



## In situ Handheld XRF

- **Offers rapid measurement (1-2 minutes)**
- **Can easily transfer data along with GPS coordinates**
- **With proper calibration and sample preparation will give quantitative results**
- **Has LOQ for most heavy metals on the order of 10 to 25 mg/kg**



**Thank You For Your Attention**